

# Appendix A

Inventory and Characteristics of Spent Nuclear Fuel, High-Level Radioactive Waste, and Other Materials

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# APPENDIX A. INVENTORY AND CHARACTERISTICS OF SPENT NUCLEAR FUEL, HIGH-LEVEL RADIOACTIVE WASTE, AND OTHER MATERIALS

## A.1 Introduction

This appendix describes the inventory and characteristics of the spent nuclear fuel and high-level radioactive waste that the U.S. Department of Energy (DOE) anticipates it would place in a monitored geologic repository at Yucca Mountain. It includes information about other highly radioactive material that DOE could dispose of in the proposed repository. It also provides information on the background and sources of the material, present storage conditions, the final disposal forms, and the amounts and characteristics of the material. The data provided in this appendix are the best available estimates of projected inventories.

The Proposed Action inventory evaluated in this environmental impact statement (EIS) consists of 70,000 metric tons of heavy metal (MTHM), comprised of 63,000 MTHM of commercial spent nuclear fuel and 7,000 MTHM of DOE materials. The DOE materials consist of 2,333 MTHM of spent nuclear fuel and 4,667 MTHM (8,315 canisters) of solidified high-level radioactive waste. The inventory includes surplus weapons-usable plutonium, which would be in the forms of spent mixed-oxide fuel and immobilized plutonium.

The Nuclear Waste Policy Act, as amended (also called the NWPA), prohibits the U.S. Nuclear Regulatory Commission from approving the emplacement of more than 70,000 MTHM in the first repository until a second repository is in operation [Section 114(d)]. However, in addition to the Proposed Action, this EIS evaluates the cumulative impacts for two additional inventories (referred to as Inventory Modules 1 and 2):

- The Module 1 inventory consists of the Proposed Action inventory plus the remainder of the total projected inventory of commercial spent nuclear fuel (for maximum projections, see Section A.2.1.5.1), high-level radioactive waste, and DOE spent nuclear fuel. Emplacement of Inventory Module 1 wastes in the repository would raise the total amount emplaced above 70,000 MTHM. As mentioned above, emplacement of more than 70,000 MTHM of spent nuclear fuel and high-level radioactive waste would require legislative action by Congress unless a second licensed repository was in operation.
- Inventory Module 2 includes the Module 1 inventory plus the inventories of the candidate materials, commercial Greater-Than-Class-C low-level radioactive waste and DOE Special-Performance-Assessment-Required waste. There are several reasons to evaluate the potential for disposing of these candidate materials in a monitored geologic repository in the near future. Because both materials exceed Class C low-level radioactive limits for specific radionuclide concentrations as defined in 10 CFR Part 61, they are generally unsuitable for near-surface disposal. Also, the Nuclear Regulatory Commission specifies in 10 CFR 61.55(a)(2)(iv) the disposal of Greater-Than-Class-C waste in a repository unless the Commission approved disposal elsewhere. Further, during the scoping process for this EIS, several commenters requested that DOE evaluate the disposal of other radioactive waste types that might require isolation in a repository. Disposal of Greater-Than-Class-C and Special-Performance-Assessment-Required wastes at the proposed Yucca Mountain Repository could require a determination by the Nuclear Regulatory Commission that these wastes require permanent isolation. The present 70,000-MTHM limit on waste at the Yucca Mountain Repository could have to be addressed either by legislation or by opening a second licensed repository.

The Yucca Mountain Science and Engineering Report evaluates the 70,000-MTHM Proposed Action inventory as the base case for analysis (DIRS 153849-DOE 2001, all) and considers a repository layout for a best estimate "full inventory" case (DIRS 153849-DOE 2001, p. 2-83), which would accommodate approximately 97,000 MTHM.

#### A.1.1 INVENTORY DATA SUMMARY

There are six general inventory categories, as follows:

- Commercial spent nuclear fuel
- DOE spent nuclear fuel
- High-level radioactive waste
- Surplus weapons-usable plutonium
- Commercial Greater-Than-Class-C waste
- DOE Special-Performance-Assessment-Required waste

This section summarizes the detailed inventory data in Section A.2. The data provide a basis for the impact analysis in this EIS. Data are provided for the candidate materials included in the initial 70,000 MTHM for the Proposed Action and other inventory that is not currently proposed but might be considered for repository disposal in the foreseeable future.

This summary provides general descriptive and historic information about each waste type, including the following:

- Primary purpose and use of the data
- General comparison of the data between waste types
- Potential for change in inventory data

Table A-1 lists the inventory data that DOE used in the EIS analyses and their descriptions throughout the document.

#### A.1.1.1 Sources

Figure A-1 shows the locations of generators or sources of spent nuclear fuel and high-level radioactive waste. Spent nuclear fuel is fuel that has been withdrawn from a nuclear reactor following irradiation. The Proposed Action includes the disposal of 63,000 MTHM of commercial spent nuclear fuel in the repository. More than 99 percent of the commercial spent nuclear fuel would come from commercial nuclear reactor sites in 33 states (DIRS 104382-DOE 1995, all). In addition, DOE manages an inventory of spent nuclear fuel. The Proposed Action includes 2,333 MTHM of spent nuclear fuel from four DOE locations: the Savannah River Site in South Carolina, the Hanford Site in Washington, the Idaho National Engineering and Environmental Laboratory, and Fort St. Vrain in Colorado.

High-level radioactive waste is the highly radioactive material resulting from the reprocessing or treatment of spent nuclear fuel. The Proposed Action includes disposing of 4,667 MTHM of high-level radioactive waste in the repository. High-level radioactive waste is stored at the Savannah River Site, the Hanford Site, the Idaho National Engineering and Environmental Laboratory, and the West Valley Demonstration Project in New York.

The President has declared an amount of plutonium to be surplus to national security needs (DIRS 118979-DOE 1999, p. 1-3). This surplus weapons-usable plutonium includes purified plutonium, nuclear weapons components, and plutonium residues. This inventory is included in the Proposed Action, and the Department would dispose of it as either spent mixed oxide fuel from a commercial nuclear reactor

**Table A-1.** Use of Appendix A radioactivity inventory data in EIS chapters and appendixes (page 1 of 2).

Item <sup>a</sup>	Appendix A	EIS section
Number of commercial nuclear sites	Table A-3	1.1, 2.2, 2.2.2, 2.4.1, 6.1, Ch. 7 introduction, 7.2, 7.2.1, 7.3, J.1.3.1.1
Number of DOE sites	A.1.1.1	1.1, 2.2, 2.2.2, 2.4.1, 6.1, Ch. 7 introduction, 7.2, 7.2.1, 7.3
Mapped location of sites	Figure A-1	Figure 1-1, several Chapter 6, 7, App. J and K figures
Commercial SNF material	A.2.1.5.3	1.2.2.1
Commercial SNF dimensions	Table A-18	1.2.1 and Figure 1-2
Commercial SNF cladding material	A.2.1.5.3	1.2.2.1, K.2.1.4.1
Percentage of commercial SNF with stainless-steel cladding	A.2.1.5.3	1.2.2.1, 5.5.1, K.2.1.4.1
MOX SNF part of commercial SNF Proposed Action	A.2.4.5.1.1	1.2.2.1, 1.2.4
Number of sites with existing or planned ISFSIs	Table A-4	1.2.1
Amount of commercial SNF projected for each site	Tables A-7 and A-8	6.1.1, K.2.1.6
DOE SNF storage locations	Table A-20	1.2.2.2, K.2.1.6
HLW generators	A.2.3.2	1.2.3
HLW vitrification status	A.2.3.3	1.2.3
Weapons-usable Pu declared surplus	A.2.4.1	1.2.4
Two forms: MOX and immobilized Pu	A.2.4.1	1.2.4
Proposed Action inventory	A.1	1.2.2.1, 2.1, 5.1, 8.1.2.1, K.2.2
Total projected inventory commercial SNF	Figure A-2	7.2, 7.3, 8.1.2.1
Total projected inventory DOE SNF	Figure A-2	7.2, 7.3, 8.1.2.1, K.2.2
Total projected inventory HLW	Figure A-2	7.2, 7.3, 8.1.2.1, K.2.2
Total projected GTCC waste	Table A-54	7.3, 8.1.2.1, I.3.1.3
Total projected SPAR waste	Table A-59	7.3, 8.1.2.1, I.3.1.3
Kr-85 (gas) is contained in fuel gap of commercial SNF	A.2.1.5.2	4.1.2.3.2, H.2.1.4.1.2
Radionuclide inventory for commercial SNF	Tables A-9, A-10, and A-11	4.1.8.1, H.2.1.4, Table H-4, J.1.4.2.1, K.2.2
Cs-137, actinide, and total curies contained in a rail shipping cask for commercial SNF, HLW, DOE SNF, and naval fuel	Derived from Tables A-10, A-21, and A-28	Table J-12, Table J-15
Radiological inventory of GTCC and SPAR waste much less than commercial SNF or HLW	Derived from Tables A-9, A-21, A-28, A-57, and Section A.2.6.4	8.2.7, 8.2.8, 8.4.1.1, F.3
Average radionuclide inventory per package for SPAR and GTCC waste	Derived from Table A-57 and Section A.2.6.4	8.3.1.1, Table I-7
C-14 (gas) is contained in fuel gap of commercial SNF	Tables A-9, A-10, and A-11	5.5, 8.3.1.1, H.2.1.4.1.2, I.3.3, I.7
PWR burnup, initial enrichment, and average cooling time	A.2.1.5	G.2.3.2, J.1.4.2.1
BWR burnup, initial enrichment, and average cooling time	A.2.1.5	G.2.3.2
DOE SNF radionuclide inventory	Table A-21	

**Table A-1.** Use of Appendix A radioactivity inventory data in EIS chapters and appendixes (page 2 of 2).

Item <sup>a</sup>	Appendix A	EIS section
Assumed packaging method for GTCC and SPAR	A.2.5.4, A.2.6.4	I.3.1.3
Chemical makeup of waste inventory	Tables A-15, A-16, A-22, A-32, A-33, A-34, A-35, A-36, and A-37	Table I-8
MTHM per assembly for PWR and BWR	Table A-17	J.1.3.1.1
Most HLW stored in underground vaults	A.2.3.3	K.2.1.5.3

a. Abbreviations: SNF = spent nuclear fuel; MOX = mixed oxide; ISFSI = independent spent fuel storage installation; HLW = high-level radioactive waste; Pu = plutonium; GTCC = Greater-Than-Class-C; SPAR = Special-Performance-Assessment-Required; MTHM = metric tons of heavy metal; Kr = krypton; Cs = cesium; PWR = pressurized-water reactor; BWR = boiling-water reactor.

(that is, commercial spent nuclear fuel) or immobilized plutonium in a high-level radioactive waste canister (that is, as high-level radioactive waste), or a combination of these two inventory categories (DIRS 118979-DOE 1999, p. 1-3). Spent mixed-oxide fuel would come from one or more of the existing commercial reactor sites. DOE has selected the Savannah River site in South Carolina as the location for the immobilized plutonium disposition facilities.

For purposes of analysis, this EIS assumes that the high-level radioactive waste canisters, which would contain immobilized plutonium and borosilicate glass, would come from the Savannah River Site.

Greater-Than-Class-C waste is waste with concentrations of certain radionuclides that exceed the Class C limits stated in 10 CFR Part 61, thereby making it unsuitable for near-surface disposal. Greater-Than-Class-C waste is generated by a number of sources including commercial nuclear utilities, sealed radioactive sources, and wastes from "other generators." These other generators include carbon-14 users, industrial research and development applications, fuel fabricators, university reactors, and others. These wastes are currently stored at the commercial and DOE sites and exist in most states. They are included in Inventory Module 2 of the EIS but are not part of the Proposed Action.

Special-Performance-Assessment-Required wastes are also Greater-Than-Class-C wastes managed by DOE and are stored primarily at the Hanford Site, Idaho National Engineering and Environmental Laboratory, West Valley Demonstration Project, and Oak Ridge National Laboratory in Tennessee. These wastes are included in Inventory Module 2 of the EIS but are not part of the Proposed Action.

#### A.1.1.2 Present Storage and Generation Status

Commercial spent nuclear fuel is stored at reactor sites in either a spent fuel pool or in a dry storage configuration generally referred to as an independent spent fuel storage installation. Through 1999, approximately 40,000 MTHM of commercial spent nuclear fuel has been discharged from reactors (DIRS 153849-DOE 2001, p. 1-10). DOE spent nuclear fuel is also stored either underwater in basins or in a dry storage configuration.

As discussed in the next section, DOE would receive high-level radioactive waste at the repository in a solidified form in stainless-steel canisters. Until shipment to the repository, the canisters would be stored at the commercial and DOE sites. With the exception of the West Valley Demonstration Project, filled canisters are stored in below-grade facilities. The West Valley canisters would be stored in an above-ground shielded facility.





Figure A-1. Locations of commercial and DOE sites and Yucca Mountain.

#### A.1.1.3 Final Waste Form

Other than drying or potential repackaging, treating is not necessary for commercial spent nuclear fuel. Therefore, the final form would be spent nuclear fuel either as bare intact assemblies or in sealed canisters. Bare intact fuel assemblies are those with structural and cladding integrity such that they can be handled and shipped to the repository in an approved shipping container for repackaging in a waste package in the Waste Handling Building. Other assemblies would be shipped to the repository in canisters that were either intended or not intended for disposal. Canisters not intended for disposal would be opened and their contents repackaged in waste packages in the Waste Handling Building.

For most of the DOE spent nuclear fuel categories, the fuel would be shipped in disposable canisters (canisters that can be shipped and are suitable for direct insertion into waste packages without being opened) in casks licensed by the Nuclear Regulatory Commission. Uranium oxide fuels with intact zirconium alloy cladding are similar to commercial spent nuclear fuel and could be shipped either in DOE standard canisters or as bare intact assemblies. Uranium metal fuels from Hanford and aluminum-based fuels from the Savannah River Site could require additional treatment or conditioning before shipment to the repository. If treatment was required, these fuels would be packaged in DOE disposable canisters. Category 14 sodium-bonded fuels are also expected to require treatment before disposal.

High-level radioactive waste shipped to the repository would be in stainless-steel canisters. The waste would have undergone a solidification process that yielded a leach-resistant material, typically a glass form called borosilicate glass. In this process, the high-level radioactive waste is mixed with glass-forming materials, heated and converted to a durable glass waste form, and poured into stainless-steel canisters (DIRS 104406-Picha 1997, Attachment 4, p. 2). Ceramic and metal waste matrices would be sent to the repository from Argonne National Laboratory-West in Idaho. The ceramic and metal matrices would be different solidified mixtures that also would be in stainless-steel canisters. These wastes would be the result of the electrometallurgical treatment of sodium bonded fuels.

As briefly described in Section A.1.1.1, the surplus weapon-usable plutonium could be sent to the repository in two different waste forms—spent mixed-oxide fuel assemblies or an immobilized plutonium ceramic form in a high-level radioactive waste canister and surrounded by high-level radioactive waste. The spent mixed-oxide fuel assemblies would be very similar to conventional low-enriched uranium assemblies and DOE would treat them as such. The immobilized plutonium would be placed in small cans, inserted in the high-level radioactive waste canisters, and covered with molten borosilicate glass (can-in-canister technique). The canisters containing immobilized plutonium and high-level radioactive waste would be externally identical to the normal high-level radioactive waste canisters.

#### A.1.1.4 Waste Characteristics

#### A.1.1.4.1 Mass and Volume

As discussed in Section A.1, the Proposed Action includes 70,000 MTHM in the forms of commercial spent nuclear fuel, DOE spent nuclear fuel, high-level radioactive waste, and surplus weapons-usable plutonium. Figure A-2 shows percentages of MTHM included in the Proposed Action and the relative amounts of the totals of the individual waste types included in the Proposed Action. As stated above, the remaining portion of the wastes is included in Inventory Module 1. Because Greater-Than-Class-C and Special-Performance-Assessment-Required wastes are measured in terms of volume, Figure A-3 shows the relative volume of the wastes in Inventory Module 2 compared to the inventory in Module 1.

The No-Action Alternative (see Chapter 7 and Appendix K) used this information to estimate the mass and volume of the spent nuclear fuel and high-level radioactive waste at commercial and DOE sites in five regions of the contiguous United States.

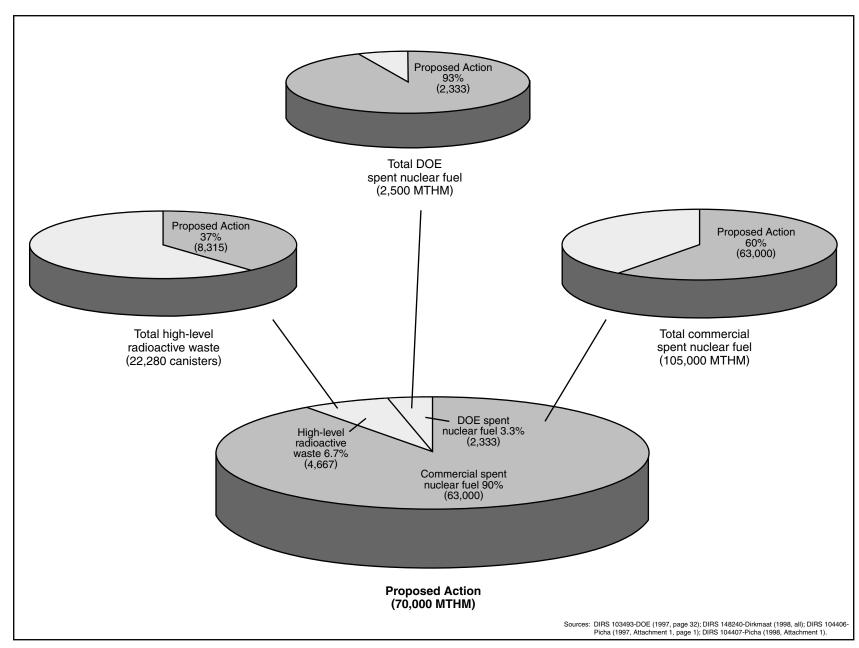
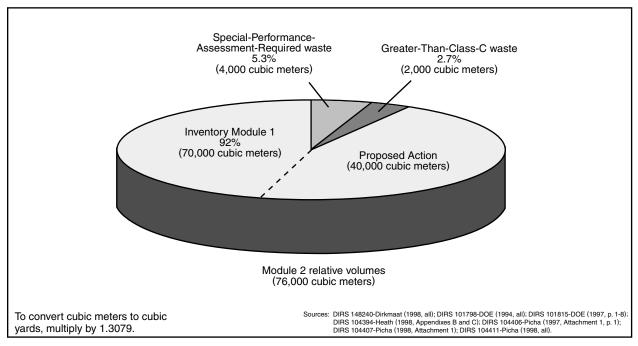


Figure A-2. Proposed Action spent nuclear fuel and high-level radioactive waste inventory.



**Figure A-3.** Inventory Module 2 volume.

The mass and volume data for commercial spent nuclear fuel are based on annual tracking of current inventories and projections of future generations. Because increases in spent nuclear fuel inventories due to plant life extensions have been factored into the Module 1 and 2 inventories, DOE anticipates few changes in the overall mass and volume projections for this waste type. The data projections for DOE spent nuclear fuel are fairly stable because most of the projected inventory already exists, as opposed to having a large amount projected for future generation. Mass and volume data for high-level radioactive waste estimates are not as reliable. Most high-level radioactive waste currently exists as a form other than solidified borosilicate glass. The solidification processes at the Savannah River Site and West Valley Demonstration Project began in the mid-1990s; therefore, their resulting masses and volumes are known. However, the processes at the Idaho National Engineering and Environmental Laboratory and the Hanford Site have not started. Therefore, there is some uncertainty about the mass and volume that would result from those processing operations. For this analysis, DOE assumed that the high-level radioactive waste from the Hanford Site and the Idaho National Engineering and Environmental Laboratory would represent approximately 63 and 6 percent of the total high-level radioactive waste inventory, respectively, in terms of the number of canisters.

#### A.1.1.4.2 Radionuclide Inventories

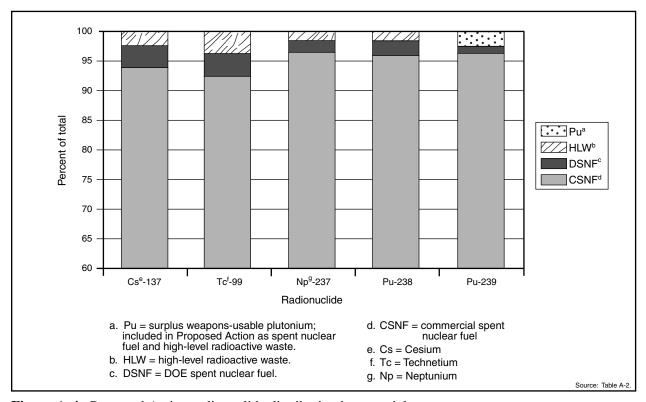
The primary purpose of presenting these data is to quantify the radionuclide inventory expected in the projected waste types. These data were used for accident scenario analyses associated with transportation, handling, and repository operations.

In a comparison of the relative amounts of radioactivity in a particular waste type, radionuclides of concern depend on the analysis being performed. For example, cesium-137 is the primary radionuclide of concern when reviewing preclosure impacts and shielding requirements. For postclosure impacts, the repository performance assessment identified technetium-99 and neptunium-237 as the nuclides that provide the greatest impacts. Plutonium-238 and -239 are shown in Chapter 7 to contribute the most to doses for the No-Action Alternative. Table A-2 presents the inventory of each of these radionuclides included in the Proposed Action. Figure A-4 shows that at least 92 percent of the total inventory of each of these radionuclides is in commercial spent nuclear fuel.

**Table A-2.** Selected radionuclide inventory for the Proposed Action (curies).<sup>a</sup>

	Commercial	DOE	High-level	Surplus	
Radionuclide <sup>b</sup>	spent nuclear fuel	spent nuclear fuel	radioactive waste	plutonium	Totals
Cesium-137	$4.5 \times 10^{9}$	$1.7 \times 10^{8}$	$1.7 \times 10^{8}$	NA <sup>c</sup>	$4.8 \times 10^{9}$
Technetium-99	$9.5 \times 10^{5}$	$2.9 \times 10^{4}$	$2.1 \times 10^4$	NA	$1.0 \times 10^{6}$
Neptunium-237	$3.0 \times 10^4$	$4.8 \times 10^{2}$	$4.5 \times 10^{2}$	NA	$3.1 \times 10^4$
Plutonium-238	$2.4 \times 10^{8}$	$5.6 \times 10^{6}$	$3.0 \times 10^{6}$	$7.6 \times 10^4$	$2.5 \times 10^{8}$
Plutonium-239	$2.4 \times 10^{7}$	$3.8 \times 10^{5}$	$4.4 \times 10^4$	$1.0 \times 10^{6}$	$2.5 \times 10^{7}$

- a. Source: Compiled from Tables A-11, A-21, A-28, A-29, A-30, A-31, A-50, and A-51.
- b. Half-lives are listed in Table A-11.
- c. NA = not applicable.



**Figure A-4.** Proposed Action radionuclide distribution by material type.

# A.1.1.4.3 Chemical Composition

The appendix presents data for the chemical composition of the primary waste types. For commercial spent nuclear fuel, the elemental composition of typical pressurized-water and boiling-water reactor fuel is provided on a per-assembly basis. Data are also provided on the number of stainless-steel clad assemblies in the current inventory.

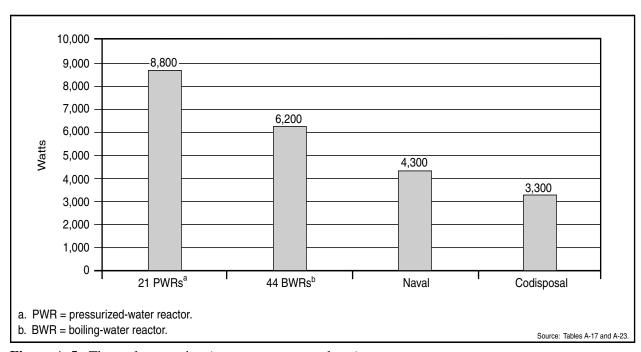
For DOE spent nuclear fuel and high-level radioactive waste, this appendix contains tables that describe the composition of the total inventory of the spent nuclear fuel (by representative category) or high-level radioactive waste (by site).

#### A.1.1.4.4 Thermal Output

Thermal generation data associated with each material type are provided in this appendix.

The data presented in the thermal output sections of this appendix for each waste type are presented as watts per assembly or MTHM for commercial spent nuclear fuel, and watts per canister for DOE spent nuclear fuel or high-level radioactive waste. Figure A-5 normalizes these data into a common, watts-per-waste-package comparison. The following waste packages are compared: one containing 21 average pressurized-water reactor assemblies, one containing 44 average boiling-water reactor assemblies, a codisposal waste package containing five high-level radioactive waste canisters and one DOE spent nuclear fuel canister, and a waste package containing one dual-purpose canister of naval spent nuclear fuel (also a DOE spent fuel).

Figure A-5 uses conservative assumptions to illustrate the bounding nature of the thermal data for commercial spent nuclear fuel. The commercial spent nuclear fuel data represent average assemblies that are assumed to have cooled for about 25 years. The naval spent nuclear fuel data are a best estimate of the thermal generation of a canister of naval spent nuclear fuel at a minimum cooling time of 5 years. The thermal data selected for the high-level radioactive waste are conservatively represented by the canisters from the Savannah River Site and are combined with the highest values of thermal output from all projected DOE spent nuclear fuel categories. As noted in Chapter 2, blending of hot and cold commercial spent nuclear fuel could be employed to meet waste package thermal load limits.



**Figure A-5.** Thermal generation (watts per waste package).

#### A.1.1.4.5 Canister Data

Commercial spent nuclear fuel, which would be shipped in canisters not suitable for disposal, would be removed from the canister and placed in a waste package. Typically, DOE spent nuclear fuel and high-level radioactive waste would be sent to the repository in disposable canisters. The design specifications for DOE spent nuclear fuel canisters are in DIRS 137713-DOE (1998, all). These canisters are generally of two diameters—nominally 46 and 61 centimeters (18 and 24 inches). They also would be designed for two different lengths, nominally 3 and 4.5 meters (10 and 15 feet), to enable codisposal with high-level radioactive waste canisters. Certain DOE spent nuclear fuel categories require specific disposal canister designs. Naval fuels would be sent to the repository in disposable canisters, which are described in DIRS 125735-Guida (1997, all) and DIRS 101941-USN (1996, pp. 3-1 to 3-11). N-Reactor fuels from the

Hanford Site would be sent to the repository in multicanister overpacks 64 centimeters (25.3 inches) in diameter, 420 centimeters (65 inches) long, which are described in DIRS 148489-DE&S Hanford (1997, all).

High-level radioactive waste would be sent to the repository in stainless-steel canisters, nominally 61 centimeters (24 inches) in diameter and either 3 or 4.5 meters (10 or 15 feet) in length, depending on the DOE site. The canister design specifications are contained in DIRS 101854-Marra, Harbour, and Plodinec (1995, all) and DIRS 103500-WVNS (n.d., WQR-2.2) for the operating vitrification processes at Savannah River Site and West Valley Demonstration Project, respectively. The other sites would use canister designs similar to those currently in use (DIRS 104406-Picha 1997, all).

These data were for analysis of the No-Action Alternative (see Chapter 7 and Appendix K) to determine the time required to breach the canisters after they are exposed to weather elements.

# A.2 Materials

This section describes the characteristics of the materials DOE has considered for disposal in the proposed Yucca Mountain Repository. All candidate materials would have to meet approved acceptance criteria.

#### A.2.1 COMMERCIAL SPENT NUCLEAR FUEL

#### A.2.1.1 Background

Spent nuclear fuel is fuel that has been withdrawn from a nuclear reactor following irradiation. Spent nuclear fuel from light-water reactors (pressurized-water and boiling-water reactors) would be the primary source of radioactivity and thermal load in the proposed monitored geologic repository. Spent nuclear fuels from civilian research reactors (General Atomics, Aerotest, etc.) account for less than 0.001 percent of the projected total in the Proposed Action (DIRS 104382-DOE 1995, all). The fuels addressed in this section are those discharged from commercial light-water reactors.

Section A.2.2 discusses the spent nuclear fuel from the Fort St. Vrain reactor in Colorado as part of DOE spent nuclear fuels, as are the fuels from Shippingport, Three Mile Island-2, and other fuels from commercial facilities that DOE has taken title to and is managing at its facilities.

#### A.2.1.2 Sources

The sources of commercial spent nuclear fuel are the commercial nuclear powerplants throughout the country. Table A-3 lists the individual reactors, reactor type, state, and actual or projected years of operation. The operating periods reflect six plants that have recently been granted extensions to their operating licenses. As noted in the table, additional extensions could be forthcoming, which could extend some of the operating periods. The operation period is also subject to change if a utility shuts down early. For conservatism, the estimated inventory of commercial spent nuclear fuel in Modules 1 and 2 was derived from the Energy Information Administration's "high case" projections. The high case assumes that all currently operating nuclear powerplants would renew their operating licenses for an additional 10 years.

#### A.2.1.3 Present Status

Nuclear power reactors store spent nuclear fuel in spent fuel pools under U.S. Nuclear Regulatory Commission licenses, and they can combine that option with above-grade dry storage in an independent

**Table A-3.** Commercial nuclear power reactors in the United States and their projected years of operation.<sup>a</sup>

Unit name	Reactor type <sup>b</sup>	State	Operations period <sup>c</sup>	Unit name	Reactor type <sup>b</sup>	State	Operations period <sup>c</sup>
Arkansas Nuclear One 1 <sup>d</sup>	PWR	AR	1974-2034	Millstone 2	PWR	CT	1975-2015
Arkansas Nuclear One 2	PWR	AR	1978-2018	Millstone 3	PWR	CT	1986-2025
Beaver Valley 1	PWR	PA	1976-2016	Monticello	BWR	MN	1971-2010
Beaver Valley 2	PWR	PA	1978-2018	Nine Mile Point 1	BWR	NY	1969-2009
Big Rock Point	BWR	MI	1963-1997	Nine Mile Point 2	BWR	NY	1987-2026
Braidwood 1	PWR	IL	1987-2026	North Anna 1	PWR	VA	1978-2018
Braidwood 2	PWR	IL	1988-2027	North Anna 2	PWR	VA	1980-2020
Browns Ferry 1	BWR	AL	1973-2013	Oconee 1 <sup>d</sup>	PWR	SC	1973-2033
Browns Ferry 2	BWR	AL	1974-2014	Oconee 2 <sup>d</sup>	PWR	SC	1973-2033
Browns Ferry 3	BWR	AL	1976-2016	Oconee 3 <sup>d</sup>	PWR	SC	1974-2034
Brunswick 1	BWR	NC	1976-2016	Oyster Creek	BWR	NJ	1969-2009
Brunswick 2	BWR	NC	1974-2014	Palisades	PWR	MI	1972-2007
Byron 1	PWR	IL	1985-2024	Palo Verde 1	PWR	AZ	1985-2024
Byron 2	PWR	IL	1987-2026	Palo Verde 2	PWR	AZ	1986-2025
Callaway	PWR	MO	1984-2024	Palo Verde 3	PWR	AZ	1987-2027
Calvert Cliffs 1 <sup>d</sup>	PWR	MD	1974-2034	Peach Bottom 2	BWR	PA	1973-2013
Calvert Cliffs 2 <sup>d</sup>	PWR	MD	1976-2036	Peach Bottom 3	BWR	PA	1974-2014
Catawba 1	PWR	SC	1985-2024	Perry 1	BWR	OH	1986-2026
Catawba 2	PWR	SC	1986-2026	Pilgrim 1	BWR	MA	1972-2012
Clinton	BWR	IL	1987-2026	Point Beach 1	PWR	WI	1970-2010
Comanche Peak 1	PWR	TX	1990-2030	Point Beach 2	PWR	WI	1973-2013
Comanche Peak 2	PWR	TX	1993-2033	Prairie Island 1	PWR	MN	1974-2013
Cooper Station	BWR	NE	1974-2014	Prairie Island 2	PWR	MN	1974-2014
Crystal River 3	PWR	FL	1977-2016	Quad Cities 1	BWR	IL	1972-2012
D. C. Cook 1	PWR	MI	1974-2014	Quad Cities 2	BWR	IL	1972-2012
D. C. Cook 2	PWR	MI	1977-2017	Rancho Seco	PWR	CA	1974-1989
Davis-Besse	PWR	OH	1977-2017	River Bend 1	BWR	LA	1985-2025
Diablo Canyon 1	PWR	CA	1984-2021	Salem 1	PWR	NJ	1976-2016
Diablo Canyon 2	PWR	CA	1985-2025	Salem 2	PWR	NJ	1981-2020
Dresden 1	BWR	IL	1959-1978	San Onofre 1	PWR	CA	1967-1992
Dresden 2	BWR	IL	1969-2006	San Onofre 2	PWR	CA	1982-2013
Dresden 3	BWR	IL	1971-2011	San Onofre 3	PWR	CA	1983-2013
Duane Arnold 1	BWR	IA	1974-2014	Seabrook 1	PWR	NH	1990-2026
Edwin I. Hatch 1	BWR	GA	1974-2014	Sequoyah 1	PWR	TN	1980-2020
Edwin I. Hatch 2	BWR	GA	1978-2018	Sequoyah 2	PWR	TN	1981-2021
Fermi 2	BWR	MI	1985-2025	Shearon Harris	PWR	NC	1987-2026
Fort Calhoun 1	PWR	NE	1973-2013	South Texas Project 1	PWR	TX	1988-2016
Ginna	PWR	NY	1969-2009	South Texas Project 2	PWR	TX	1989-2023
Grand Gulf 1	BWR	MS	1984-2022	St. Lucie 1	PWR	FL	1976-2016
Haddam Neck	PWR	CT	1968-1996	St. Lucie 2	PWR	FL	1983-2023
Hope Creek	BWR	NJ	1986-2026	Summer 1	PWR	SC	1982-2022
Humboldt Bay	BWR	CA	1962-1976	Surry 1	PWR	VA	1972-2012
H.B. Robinson 2	PWR	SC	1970-2010	Surry 2	PWR	VA	1973-2013
Indian Point 1	PWR	NY	1962-1974	Susquehanna 1	BWR	PA	1982-2022
Indian Point 2	PWR	NY	1973-2013	Susquehanna 2	BWR	PA	1984-2024
Indian Point 3	PWR	NY	1976-2015	Three Mile Island 1	PWR	PA	1974-2014
James A. FitzPatrick/ Nine Mile Point	BWR	NY	1974-2014	Trojan Turkey Point 3	PWR PWR	OR FL	1975-1992 1972-2012
Joseph M. Farley 1	PWR	AL	1977-2017	Turkey Point 4	PWR	FL	1973-2013
Joseph M. Farley 2	PWR	AL	1981-2021	Vermont Yankee	BWR	VT	1973-2012
Kewaunee	PWR	WI	1973-2013	Vogtle 1	PWR	GA	1987-2027
LaCrosse	BWR	WI	1967-1987	Vogtle 2	PWR	GA	1989-2029
LaSalle 1	BWR	IL	1970-2022	Columbia Generating	BWR	WA	1984-2023
LaSalle 2	BWR	IL	1970-2023	Station			
Limerick 1	BWR	PA	1985-2024	Waterford 3	PWR	LA	1985-2024
Limerick 2	BWR	PA	1989-2029	Watts Bar 1	PWR	TN	1996-2035
Maine Yankee	PWR	ME	1972-1996	Wolf Creek	PWR	KS	1985-2025
McGuire 1	PWR	NC	1981-2021	Yankee-Rowe	PWR	MA	1963-1991
McGuire 2	PWR	NC	1983-2023	Zion 1	PWR	IL	1973-1997
Millstone 1	BWR	CT	1970-2010	Zion 2	PWR	IL	1974-1996

a. Source: DIRS 103493-DOE (1997, Appendix C).

I

b. PWR = pressurized-water reactor; BWR = boiling-water reactor.

c. As defined by current shutdown or full operation through license period (as of 1997), except as noted in Footnote d.

d. These plants have recently been granted 20-year operating license extensions. Several additional plants have applied for operating license extensions, and others could do so in the future.

spent fuel storage installation. When a reactor is refueled, spent fuel is transferred to the spent fuel pool, where it typically remains until the available pool capacity is reached. When in-pool storage capacity has been fully used, utilities have turned to dry cask storage in an independent spent fuel storage installation to expand their onsite spent fuel storage capacities. In 1990, the Nuclear Regulatory Commission amended its regulations to authorize licensees to store spent nuclear fuel at reactor sites in approved storage casks (DIRS 101913-Raddatz and Waters 1996, all).

Commercial nuclear utilities currently use three Nuclear Regulatory Commission-approved general dry storage system design types—metal storage casks and metal canisters housed in either concrete casks or concrete vaults—for use in licensed independent spent fuel storage installations. Raddatz and Waters (DIRS 101913-1996, all) contains detailed information on models currently approved by the Commission. Table A-4 lists the numbers of existing and planned at-reactor independent spent fuel storage installations in the United States as of 2001.

**Table A-4.** Sites with existing or planned independent spent fuel storage installations.<sup>a</sup>

	1	•	
Insta	llations	Numb	er
Existing		18	
Planned		15	
	DIDC 155(04 D 11)	(2001	11)

a. Sources: DIRS 155604-Delligatti (2001, all).

# A.2.1.4 Final Spent Nuclear Fuel Form

The final form of commercial spent nuclear fuel to be disposed of in the proposed repository would be the spent reactor fuel assemblies. The repository would receive bare spent nuclear fuel assemblies, spent nuclear fuel packaged in canisters not intended for disposal, and spent nuclear fuel packaged in canisters intended for disposal.

#### A.2.1.5 Spent Nuclear Fuel Characteristics

There are 22 classes of nuclear fuel assemblies, with 127 individual fuel types in those classes. Seventeen of the classes are for pressurized-water reactor fuels and 5 are for boiling-water reactors (DIRS 102588-DOE 1992, Appendix 2A). For this EIS, the assemblies chosen for analysis represent an assembly type being used in the more recently built reactors. This results in physical characteristics that provide a realistic estimate for EIS analyses. Specifically chosen to represent the fuel types were the Westinghouse 17 × 17 LOPAR fuel assembly for the pressurized-water reactor and the General Electric BWR/4-6, 8 × 8 fuel assembly for the boiling-water reactor. Table A-5 lists the fissile content and performance parameters selected to define the radiological characteristics of these fuel assemblies. These parameters represent the average values for pressurized-water reactor and boiling-water reactor fuel to be received at the proposed repository.

**Table A-5.** Average spent nuclear fuel parameters.<sup>a</sup>

	Burnup	Initial enrichment (percent	Age	
Fuel type <sup>b</sup>	(MWd/MTHM) <sup>c</sup>	of U-235 by weight)	(years)	
Average PWR	41,200	3.75	23	
Average BWR	33,600	3.03	23	

- a. Source: DIRS 153849-DOE (2001, p. 3-13).
- b. PWR = pressurized-water reactor; BWR = boiling-water reactor.
- c. MWd/MTHM = megawatt-days per metric ton of heavy metal; to convert metric tons to tons, multiply by 1.1023.

In the Draft EIS, Appendix A, DOE used fuel characteristics similar to those in Table A-5 to estimate consequences from accidents during transportation and repository operations. Since the publication of the Draft EIS, there has been concern that the radionuclide inventories of these average fuel assemblies

could underestimate the potential dose consequences of an accidental release. In particular, using the average age of fuel likely to be sent to the repository does not fully take into account the effects of exponential radioactive decay and dose potential from accidental releases as the fuel aged.

As a result of these considerations, DOE undertook an effort to evaluate characteristics of commercial pressurized-water and boiling-water reactor spent nuclear fuel assemblies that span the entire range and distribution of the assemblies that would be shipped to the repository (DIRS 156919-Ikenberry 2001, all). The object of the effort was to characterize pressurized-water and boiling-water reactor assemblies that would represent a median hazard over the entire spectrum of commercial spent nuclear fuel. The result of this effort is in Table A-6, which lists the representative fuel used for accident analyses in this Final EIS. The effort included consideration of both mixed oxide (see Section A.2.4.5.1) as well as the bounding fuel types (highest burnup with lowest cooling time).

Table A-6. Representative commercial spent nuclear fuel characteristics for accident analyses.<sup>a</sup>

	Burnup	Initial enrichment	
Fuel type <sup>b</sup>	(MWd/MTHM) <sup>c</sup>	(percent of U-235 by weight)	Age (years)
Representative PWR	50,000	4.5	15
Representative BWR	40,000	3.5	14

- a. Source: DIRS 156919-Ikenberry (2001, all).
- b. PWR = pressurized-water reactor; BWR = boiling-water reactor.
- c. MWd/MTHM = megawatt-days per metric ton of heavy metal; to convert metric tons to tons, multiply by 1.1023.

#### A.2.1.5.1 Mass and Volume

As discussed in Section A.1, the Proposed Action includes 63,000 MTHM of commercial spent nuclear fuel. For the No-Action Alternative (continued storage) analysis, Table A-7 lists the distribution of this expected inventory by reactor site. The historic and projected spent nuclear fuel discharge and storage information in Table A-7 is consistent with the annual projections provided by the Energy Information Administration (DIRS 103493-DOE 1997, p. 32). The "1995 Actual" data presented in Table A-7 represents the amount of spent nuclear fuel stored at a particular site regardless of the reactor from which it was discharged. For analysis purposes, the table lists spent nuclear fuel currently stored at the General Electric Morris, Illinois, facility to be at Dresden, because these facilities are located near each other.

For analyses associated with the Proposed Action, the projected spent nuclear fuel from pressurized-water reactors comprises 65 percent of the 63,000 metric tons of heavy metal (DIRS 100265-CRWMS M&O 1997, p. A-2). The balance consists of spent nuclear fuel from boiling-water reactors. Using the nominal volume for the spent nuclear fuel assemblies described in Section A.2.1.5.5, the estimated volume of spent nuclear fuel in the Proposed Action, exclusive of packaging, is 29,000 cubic meters.

Section A.1 also discusses the additional inventory modules evaluated in this EIS. Inventory Modules 1 and 2 both include the maximum expected discharge inventory of commercial spent nuclear fuel. Table A-8 lists historic and projected amounts of spent nuclear fuel discharged from commercial reactors through 2046. The estimated unpackaged volume of spent nuclear fuel for these modules is approximately 47,000 cubic meters. For conservatism, these data were derived from the Energy Information Administration "high case" assumptions. The high case assumes that all currently operating nuclear units would renew their operating licenses for an additional 10 years (DIRS 103493-DOE 1997, p. 32).

#### A.2.1.5.2 Amount and Nature of Radioactivity

Spent nuclear fuel from commercial nuclear powerplants contains several hundred radionuclides when removed from the reactor. However, due to minor quantities, short half-lives, biological significance, and other factors, most of these are not important from a public health hazard standpoint. DOE has

**Table A-7.** Proposed Action spent nuclear fuel inventory (MTHM).<sup>a</sup>

	Fuel	1995	1996-		Equivalent		Fuel	1995	1996-		Equivalent
Site	type <sup>b</sup>	actual	2011 <sup>c</sup>	Total <sup>d</sup>	assemblies		type <sup>b</sup>	actual	2011 <sup>c</sup>	Total <sup>d</sup>	assemblies
Arkansas Nuclear One	PWR	643	466	1,109	2,526	Monticello	BWR	147	280	426	2,324
Beaver Valley	PWR	437	581	1,018	2,206	North Anna	PWR	570	613	1,184	2,571
Big Rock Point	BWR	44	14	58	439	Oconee	PWR	1,098	767	1,865	4,028
Braidwood	PWR	318	711	1,029	2,424	Oyster Creek	BWR	374	325	699	3,824
Browns Ferry	BWR	840	1,092	1,932	10,402	Palisades	PWR	338	247	585	1,473
Brunswick	Both	448	448	896	4,410	Palo Verde	PWR	556	1,118	1,674	4,082
Byron	PWR	404	664	1,068	2,515	Peach Bottom	BWR	908	645	1,554	8,413
Callaway	PWR	280	422	702	1,609	Perry	BWR	178	274	452	2,470
Calvert Cliffs	PWR	641	501	1,142	2,982	Pilgrim	BWR	326	201	527	2,853
Catawba	PWR	465	683	1,148	2,677	Point Beach	PWR	529	347	876	2,270
Clinton	BWR	174	303	477	2,588	Prairie Island	PWR	518	348	866	2,315
Comanche Peak	PWR	176	821	998	2,202	Quad Cities	BWR	813	464	1,277	6,953
Cooper	BWR	175	277	452	2,435	Rancho Seco	PWR	228	e	228	493
Crystal River	PWR	280	232	512	1,102	River Bend	BWR	176	356	531	2,889
D. C. Cook	PWR	777	656	1,433	3,253	Salem/Hope Creek	Both	793	866	1,659	7,154
Davis-Besse	PWR	243	262	505	1,076	San Onofre	PWR	722	701	1,423	3,582
Diablo Canyon	PWR	463	664	1,126	2,512	Seabrook	<b>PWR</b>	133	292	425	918
Dresden	BWR	1,557	590	2,146	11,602	Sequoyah	<b>PWR</b>	452	570	1,023	2,218
Duane Arnold	BWR	258	208	467	2,545	Shearon Harris	Both	498	252	750	2,499
Edwin I. Hatch	BWR	755	692	1,446	7,862	South Texas Project	<b>PWR</b>	290	722	1,012	1,871
Fermi	BWR	155	368	523	2,898	St. Lucie	<b>PWR</b>	601	419	1,020	2,701
Fort Calhoun	PWR	222	157	379	1,054	Summer	PWR	225	301	526	1,177
Ginna	PWR	282	180	463	1,234	Surry	PWR	660	534	1,194	2,604
Grand Gulf	BWR	349	506	856	4,771	Susquehanna	BWR	628	648	1,276	7,172
H. B. Robinson	<b>PWR</b>	145	239	384	903	Three Mile Island	<b>PWR</b>	311	236	548	1,180
Haddam Neck	<b>PWR</b>	355	65	420	1,017	Trojan	<b>PWR</b>	359		359	780
Humboldt Bay	BWR	29		29	390	Turkey Point	<b>PWR</b>	616	458	1,074	2,355
Indian Point	<b>PWR</b>	678	486	1,164	2,649	Vermont Yankee	BWR	387	222	609	3,299
James A. FitzPatrick/	BWR	882	930	1,812	9,830	Vogtle	PWR	335	745	1,080	2,364
Nine Mile Point						Columbia	BWR	243	338	581	3,223
Joseph M. Farley	PWR	644	530	1,174	2,555	Generating Station					
Kewaunee	PWR	282	169	451	1,172						
La Crosse	BWR	38		38	333	Waterford	PWR	253	247	500	1,217
La Salle	BWR	465	487	952	5,189	Watts Bar	PWR		251	251	544
Limerick	BWR	432	711	1,143	6,203	Wolf Creek	PWR	226	404	630	1,360
Maine Yankee	PWR	454	82	536	1,421	Yankee-Rowe	PWR	127		127	533
McGuire	PWR	714	725	1,439	3,257	Zion	PWR	841	211	1,052	2,302
Millstone	Both	959	749	1,709	6,447	Totals		31,926	31,074	63,000	218,700

a. Source: DIRS 155725-CRWMS M&O (1998, all).

determined that 51 radionuclides represent all of the health-significant species that can contribute to a radiological dose if released in an accident. The derivation of the list of radionuclides of interest in terms of impacts to the public is described in Appendix H, Section H.2.1.4.1. Tables A-9 and A-10 list these radionuclides and their inventories for average pressurized-water and boiling-water reactor spent nuclear fuel assemblies. The inventories are presented at the average decay years for each of the assemblies.

Table A-11 combines the average inventories (curies per MTHM) with the projected totals (63,000 MTHM and 105,000 MTHM) to provide a total projected radionuclide inventory for the Proposed Action and additional modules.

b. PWR = pressurized-water reactor; BWR = boiling-water reactor.

c. Projected.

d. To convert metric tons to tons, multiply by 1.1023.

e. -- = no spent nuclear fuel production.

**Table A-8.** Inventory Modules 1 and 2 spent nuclear fuel inventory (MTHM).<sup>a</sup>

	Fuel	1995			Equivalent		Fuel	1995	1996-		Equivalent
Site	type <sup>b</sup>	actual	1996-2046 <sup>c</sup>	Total <sup>d</sup>	assemblies	Site	type <sup>b</sup>	actual	2046 <sup>c</sup>	Total <sup>d</sup>	assemblies
Arkansas Nuclear One	PWR	643	1,007	1,650	3,757	Monticello	BWR	147	390	537	2,924
Beaver Valley	PWR	437	1.395	1,832	3,970	North Anna	PWR	570	1,384	1,955	4,246
Big Rock Point	BWR	44	14	58	439	Oconee	PWR	1,098	1,576	2,674	5,774
Braidwood	PWR	318	1,969	2,287	5,385	Oyster Creek	BWR	374	470	844	4,619
Browns Ferry	BWR	840	2,508	3,348	18,024	Palisades	PWR	338	395	733	1,845
Brunswick	Both	448	992	1,440	7,355	Palo Verde	PWR	556	3,017	3,573	8,712
Byron	<b>PWR</b>	404	1,777	2,181	5,139	Peach Bottom	BWR	908	1,404	2,312	12,523
Callaway	<b>PWR</b>	280	1,008	1,288	2,953	Perry	BWR	178	732	910	4,974
Calvert Cliffs	PWR	641	1,069	1,710	4,466	Point Beach	PWR	529	614	1,143	2,961
Catawba	PWR	465	1,752	2,217	5,168	Prairie Island	PWR	518	692	1,210	3,234
Clinton	BWR	174	910	1,084	5,876	Quad Cities	BWR	813	1,020	1,834	9,982
Comanche Peak	PWR	176	2,459	2,635	5,816	Pilgrim	BWR	326	444	770	4,170
Cook	PWR	777	1,379	2,155	4,892	Rancho Seco	PWR	228	e	228	493
Cooper	BWR	175	587	762	4,106	River Bend	BWR	176	956	1,132	6,153
Crystal River	PWR	280	525	805	1,734	Salem/Hope Creek	Both	793	2,452	3,245	11,584
Davis-Besse	PWR	243	582	825	1,757	San Onofre	PWR	722	1,321	2,043	5,144
Diablo Canyon	PWR	463	1,725	2,187	4,878	Seabrook	PWR	133	831	964	2,083
Dresden	BWR	1,557	984	2,541	13,740	Sequoyah	PWR	452	1,393	1,845	4,001
Duane Arnold	BWR	258	434	692	3,776	Shearon Harris	Both	498	707	1,205	3,535
Fermi	BWR	155	1,005	1,160	6,429	South Texas Project	PWR	290	2,029	2,319	4,286
Fort Calhoun	PWR	222	312	534	1,485	St. Lucie	PWR	601	1,010	1,611	4,265
Ginna	PWR	282	283	565	1,507	Summer	PWR	225	732	958	2,141
Grand Gulf	BWR	349	1,261	1,610	8,976	Surry	PWR	660	1,029	1,689	3,682
H. B. Robinson	PWR	145	364	509	1,197	Susquehanna	BWR	628	1,745	2,373	13,338
Haddam Neck	PWR	355	65	420	1,017	Three Mile Island	PWR	311	513	825	1,777
Hatch	BWR	755	1,517	2,272	12,347	Trojan	PWR	359		359	780
Humboldt Bay	BWR	29		29	390	Turkey Point	PWR	616	905	1,520	3,334
Indian Point	PWR	678	1,005	1,683	3,787	Vermont Yankee	BWR	387	434	822	4,451
James A. FitzPatrick/	BWR	882	2,018	2,900	15,732	Vogtle	PWR	335	2,122	2,458	5,378
Nine Mile Point						Columbia	BWR	243	924	1,167	6,476
Joseph M. Farley	PWR	644	1,225	1,869	4,070	Generating					
Kewaunee	PWR	282	330	612	1,591	Station					
La Crosse	BWR	38		38	333	Waterford	PWR	253	685	938	2,282
La Salle	BWR	465	1,398	1,863	10,152	Watts Bar	PWR		893	893	1,937
Limerick	BWR	432	1,958	2,390	12,967	Wolf Creek	PWR	226	1,052	1,278	2,759
Maine Yankee	PWR	454	82	536	1,421	Yankee-Rowe	PWR	127		127	533
McGuire	PWR	714	1,813	2,527	5,720	Zion	PWR	841	211	1,052	2,302
Millstone	Both	959	1,695	2,655	8,930	Totals		31,926	73,488	105,414	359,963

a. Source: DIRS 155725-CRWMS M&O (1998, all).

DOE used the fuel characteristics derived in Section A.2.1.5 and listed in Table A-6 to establish the fission product and radionuclide inventories of the pressurized-water and boiling-water reactor representative fuel assemblies used for accident analyses. For these analyses, DOE included a radionuclide contribution from activated corrosion products deposited on the surfaces of spent nuclear fuel assemblies during reactor operation. This material is called *crud*.

DOE used the fuel assembly surface concentration values in *Reexamination of Spent Fuel Shipment Risk Estimates* (DIRS 152476-Sprung et. al. 2000, all) to develop the radioactive inventory from crud. The crud contains eight radionuclides. However, because all of these radionuclides except cobalt-60 decay rapidly, after storage (aging) for 5 years or longer, cobalt-60 is the only significant radionuclide remaining. The surface concentration values at discharge from the reactor range from 2 to 140 microcuries per square centimeter for pressurized-water reactor fuel assemblies and from 11 to 595 microcuries per square centimeter for boiling-water reactor assemblies, based on measurements of fuel rods (DIRS 152476-Sprung et al. 2000, p. 7-48; DIRS 103696-Sandoval 1991, all). Due to the wide range in concentration values and the limited number of measurements, DOE elected to use the maximum (cobalt-60) crud concentration numbers (DIRS 152476-Sprung et al. 2000, p. 7-48).

b. PWR = pressurized-water reactor; BWR = boiling-water reactor.

Projected.

d. To convert metric tons to tons, multiply by 1.1023.

e. -- = no spent nuclear fuel production.

Table A-9. Radionuclide activity for average pressurized-water reactor fuel assemblies. a,b

	Curies per	<u> </u>	Curies per		Curies per
Radionuclide <sup>c</sup>	assembly	Isotope	assembly	Isotope	assembly
Hydrogen-3	$1.2 \times 10^{2}$	Antimony-125	$2.6 \times 10^{1}$	Uranium-236	$1.4 \times 10^{-1}$
Carbon-14	$6.6 \times 10^{-1}$	Tin-126	$4.5 \times 10^{-1}$	Uranium-238	$1.4 \times 10^{-1}$
Chlorine-36	$5.5 \times 10^{-3}$	Iodine-129	$1.8 \times 10^{-2}$	Neptunium-237	$2.4 \times 10^{-1}$
Iron-55	$3.4 \times 10^{0}$	Cesium-134	$4.4 \times 10^{1}$	Plutonium-238	$1.9 \times 10^{3}$
Cobalt-60	$2.2 \times 10^{2}$	Cesium-135	$2.7 \times 10^{-1}$	Plutonium-239	$1.8 \times 10^{2}$
Nickel-59	$1.3 \times 10^{0}$	Cesium-137	$3.4 \times 10^{4}$	Plutonium-240	$2.8 \times 10^{2}$
Nickel-63	$1.9 \times 10^{2}$	Promethium-147	$1.3 \times 10^{2}$	Plutonium-241	$2.4 \times 10^{4}$
Selenium-79	$2.3 \times 10^{-1}$	Samarium-151	$1.9 \times 10^{2}$	Plutonium-242	$1.0 \times 10^{0}$
Krypton-85	$1.1 \times 10^{3}$	Europium-154	$9.6 \times 10^{2}$	Americium-241	$1.6 \times 10^{3}$
Strontium-90	$2.3 \times 10^{4}$	Europium-155	$1.6 \times 10^{2}$	Americium-242/242m	$1.1 \times 10^{1}$
Zirconium-93	$1.2 \times 10^{0}$	Actinium-227	$7.3 \times 10^{-6}$	Americium-243	$1.4 \times 10^{1}$
Niobium-93m	$8.1 \times 10^{-1}$	Thorium-230	$1.4 \times 10^{-4}$	Curium-242	$9.1 \times 10^{0}$
Niobium-94	$6.0 \times 10^{-1}$	Protactinium-231	$1.6 \times 10^{-5}$	Curium-243	$9.7 \times 10^{0}$
Technetium-99	$7.3 \times 10^{0}$	Uranium-232	$2.1 \times 10^{-2}$	Curium-244	$9.0 \times 10^{2}$
Ruthenium-106	$3.3 \times 10^{-2}$	Uranium-233	$3.1 \times 10^{-5}$	Curium-245	$2.1 \times 10^{-1}$
Palladium-107	$6.6 \times 10^{-2}$	Uranium-234	$6.5 \times 10^{-1}$	Curium-246	$4.7 \times 10^{-2}$
Cadmium-113m	$1.1 \times 10^{1}$	Uranium-235	$8.0 \times 10^{-3}$		

a. Source: DIRS 150276-CRWMS M&O (2000, p. VIII-3).

Table A-10. Radionuclide activity for average boiling-water reactor fuel assemblies. a,b

	Curies per	•	Curies per		Curies per
Radionuclide <sup>c</sup>	assembly	Isotope	assembly	Isotope	assembly
Hydrogen-3	$4.2 \times 10^{1}$	Antimony-125	$1.1 \times 10^{1}$	Uranium-236	$4.5 \times 10^{-2}$
Carbon-14	$2.9 \times 10^{-1}$	Tin-126	$1.5 \times 10^{-1}$	Uranium-238	$5.7 \times 10^{-2}$
Chlorine-36	$2.1 \times 10^{-3}$	Iodine-129	$6.1 \times 10^{-3}$	Neptunium-237	$7.1 \times 10^{-2}$
Iron-55	$9.5 \times 10^{-1}$	Cesium-134	$1.6 \times 10^{1}$	Plutonium-238	$6.0 \times 10^{2}$
Cobalt-60	$6.5 \times 10^{1}$	Cesium-135	$9.9 \times 10^{-2}$	Plutonium-239	$6.0 \times 10^{1}$
Nickel-59	$3.4 \times 10^{-1}$	Cesium-137	$1.1 \times 10^{4}$	Plutonium-240	$9.3 \times 10^{1}$
Nickel-63	$4.5 \times 10^{1}$	Promethium-147	$4.9 \times 10^{1}$	Plutonium-241	$8.9 \times 10^{3}$
Selenium-79	$7.7 \times 10^{-2}$	Samarium-151	$6.6 \times 10^{1}$	Plutonium-242	$4.0 \times 10^{-1}$
Krypton-85	$3.7 \times 10^{2}$	Europium-154	$3.2 \times 10^{2}$	Americium-241	$6.1 \times 10^{2}$
Strontium-90	$7.5 \times 10^{3}$	Europium-155	$5.7 \times 10^{1}$	Americium-242/242m	$4.7 \times 10^{0}$
Zirconium-93	$4.6 \times 10^{-1}$	Actinium-227	$2.6 \times 10^{-6}$	Americium-243	$5.2 \times 10^{0}$
Niobium-93m	$3.1 \times 10^{-1}$	Thorium-230	$4.5 \times 10^{-5}$	Curium-242	$3.9 \times 10^{0}$
Niobium-94	$1.9 \times 10^{-2}$	Protactinium-231	$5.4 \times 10^{-6}$	Curium-243	$3.8 \times 10^{0}$
Technetium-99	$2.4 \times 10^{0}$	Uranium-232	$6.2 \times 10^{-3}$	Curium-244	$3.5 \times 10^{2}$
Ruthenium-106	$1.4 \times 10^{-2}$	Uranium-233	$9.1 \times 10^{-6}$	Curium-245	$7.9 \times 10^{-2}$
Palladium-107	$2.4 \times 10^{-2}$	Uranium-234	$2.1 \times 10^{-1}$	Curium-246	$1.7 \times 10^{-2}$
Cadmium-113m	$4.0 \times 10^{0}$	Uranium-235	$2.6 \times 10^{-3}$		

a. Source: DIRS 150276-CRWMS M&O (2000, p. VIII-5).

b. Burnup = 41,200 MWd/MTHM, enrichment = 3.75 percent, decay time = 23 years.

c. Half-lives are listed in Table A-11.

b. Burnup = 33,600 MWd/MTHM, enrichment = 3.03 percent, decay time = 23 years.

c. Half-lives are listed in Table A-11.

**Table A-11.** Total projected radionuclide inventories<sup>a,b</sup> (page 1 of 2).

		Pressurized-water reactor		Во	Boiling-water reactor				
			Total	curies	_	Total	curies	Grand to	als (curies)
	Half life	Curies per	Proposed	Additional	Curies per	Proposed	Additional	Proposed	Additional
Isotope	(yrs.) <sup>c</sup>	MTHM <sup>d</sup>	Action	modules	MTHM	Action	modules	Action	modules
Hydrogen-3	12.3	$2.5 \times 10^{2}$	$1.0 \times 10^{7}$	$1.7 \times 10^{7}$	$2.3 \times 10^{2}$	$5.1 \times 10^{6}$	$8.5 \times 10^{6}$	$1.6 \times 10^{7}$	$2.6 \times 10^{7}$
Carbon-14	$5.7 \times 10^3$	$1.4 \times 10^{0}$	$5.9 \times 10^{4}$	$9.8 \times 10^{4}$	$1.6 \times 10^{0}$	$3.6 \times 10^{4}$	$6.0 \times 10^{4}$	$9.5 \times 10^{4}$	$1.6 \times 10^{5}$
Chlorine-36	$3.0 \times 10^{5}$	$1.2 \times 10^{-2}$	$4.9 \times 10^{2}$	$8.2 \times 10^{2}$	$1.2 \times 10^{-2}$	$2.6 \times 10^{2}$	$4.3 \times 10^{2}$	$7.5 \times 10^{2}$	$1.2 \times 10^{3}$
Iron-55	2.7	$7.4 \times 10^{0}$	$3.0 \times 10^{5}$	$5.1 \times 10^{5}$	$5.3 \times 10^{0}$	$1.2 \times 10^{5}$	$1.9 \times 10^{5}$	$4.2 \times 10^{5}$	$7.0 \times 10^{5}$
Cobalt-60	5.3	$4.7 \times 10^{2}$	$1.9 \times 10^{7}$	$3.2 \times 10^{7}$	$3.6 \times 10^{2}$	$8.0 \times 10^{6}$	$1.3 \times 10^{7}$	$2.7 \times 10^{7}$	$4.5 \times 10^{7}$
Nickel-59	$7.6 \times 10^4$	$2.9 \times 10^{0}$	$1.2 \times 10^{5}$	$2.0 \times 10^{5}$	$1.9 \times 10^{0}$	$4.1 \times 10^{4}$	$6.9 \times 10^{4}$	$1.6 \times 10^{5}$	$2.7 \times 10^{5}$
Nickel-63	$1.0 \times 10^{2}$	$4.0 \times 10^{2}$	$1.7 \times 10^{7}$	$2.8 \times 10^{7}$	$2.5 \times 10^{2}$	$5.5 \times 10^{6}$	$9.2 \times 10^{6}$	$2.2 \times 10^{7}$	$3.7 \times 10^{7}$
Selenium-79	$6.5 \times 10^4$	$5.1 \times 10^{-1}$	$2.1 \times 10^{4}$	$3.5 \times 10^{4}$	$4.3 \times 10^{-1}$	$9.4 \times 10^{3}$	$1.6 \times 10^{4}$	$3.0 \times 10^{4}$	$5.0 \times 10^{4}$
Krypton-85	10.7	$2.5 \times 10^{3}$	$1.0 \times 10^{8}$	$1.7 \times 10^{8}$	$2.1 \times 10^{3}$	$4.6 \times 10^{7}$	$7.6 \times 10^{7}$	$1.5 \times 10^{8}$	$2.5 \times 10^{8}$
Strontium-90	29	$5.1 \times 10^4$	$2.1 \times 10^{9}$	$3.5 \times 10^{9}$	$4.2 \times 10^{4}$	$9.2 \times 10^{8}$	$1.5 \times 10^{9}$	$3.0 \times 10^{9}$	$5.0 \times 10^{9}$
Zirconium-93	$1.5 \times 10^6$	$2.6 \times 10^{0}$	$1.1 \times 10^{5}$	$1.8 \times 10^{5}$	$2.6 \times 10^{0}$	$5.7 \times 10^4$	$9.5 \times 10^{4}$	$1.6 \times 10^{5}$	$2.7 \times 10^{5}$
Niobium-93m	16	$1.8 \times 10^{0}$	$7.2 \times 10^4$	$1.2 \times 10^{5}$	$1.7 \times 10^{0}$	$3.9 \times 10^{4}$	$6.4 \times 10^{4}$	$1.1 \times 10^{5}$	$1.8 \times 10^{5}$
Niobium-94	$2.4 \times 10^4$	$1.3 \times 10^{0}$	$5.3 \times 10^4$	$8.9 \times 10^{4}$	$1.1 \times 10^{-1}$	$2.3 \times 10^{3}$	$3.9 \times 10^{3}$	$5.6 \times 10^4$	$9.3 \times 10^{4}$
Technetium-99	$2.1 \times 10^{5}$	$1.6 \times 10^{1}$	$6.5 \times 10^{5}$	$1.1 \times 10^{6}$	$1.4 \times 10^{1}$	$3.0 \times 10^{5}$	$5.0 \times 10^{5}$	$9.5 \times 10^{5}$	$1.6 \times 10^{6}$
Ruthenium-106	1.0	$7.2 \times 10^{-2}$	$3.0 \times 10^{3}$	$4.9 \times 10^{3}$	$7.9 \times 10^{-2}$	$1.8 \times 10^{3}$	$2.9 \times 10^{3}$	$4.7 \times 10^{3}$	$7.9 \times 10^{3}$
Palladium-107	$6.5 \times 10^6$	$1.4 \times 10^{-1}$	$5.9 \times 10^{3}$	$9.8 \times 10^{3}$	$1.3 \times 10^{-1}$	$2.9 \times 10^{3}$	$4.8 \times 10^{3}$	$8.8 \times 10^{3}$	$1.5 \times 10^{4}$
Cadmium-113m	14	$2.5 \times 10^{1}$	$1.0 \times 10^{6}$	$1.7 \times 10^{6}$	$2.2 \times 10^{1}$	$4.9 \times 10^{5}$	$8.1 \times 10^{5}$	$1.5 \times 10^{6}$	$2.5 \times 10^{6}$
Antimony-125	2.8	$5.6 \times 10^{1}$	$2.3 \times 10^{6}$	$3.9 \times 10^{6}$	$5.9 \times 10^{1}$	$1.3 \times 10^{6}$	$2.2 \times 10^{6}$	$3.6 \times 10^{6}$	$6.0 \times 10^{6}$
Tin-126	$1.0 \times 10^{6}$	$9.8 \times 10^{-1}$	$4.0 \times 10^{4}$	$6.7 \times 10^{4}$	$8.5 \times 10^{-1}$	$1.9 \times 10^{4}$	$3.1 \times 10^{4}$	$5.9 \times 10^4$	$9.8 \times 10^{4}$
Iodine-129	$1.7 \times 10^7$	$3.9 \times 10^{-2}$	$1.6 \times 10^{3}$	$2.7 \times 10^{3}$	$3.4 \times 10^{-2}$	$7.5 \times 10^{2}$	$1.2 \times 10^{3}$	$2.4 \times 10^{3}$	$3.9 \times 10^{3}$
Cesium-134	2.1	$9.5 \times 10^{1}$	$3.9 \times 10^{6}$	$6.5 \times 10^{6}$	$8.7 \times 10^{1}$	$1.9 \times 10^{6}$	$3.2 \times 10^{6}$	$5.8 \times 10^{6}$	$9.7 \times 10^{6}$
Cesium-135	$2.3 \times 10^6$	$5.8 \times 10^{-1}$	$2.4 \times 10^{4}$	$3.9 \times 10^{4}$	$5.5 \times 10^{-1}$	$1.2 \times 10^{4}$	$2.0 \times 10^{4}$	$3.6 \times 10^{4}$	$6.0 \times 10^{4}$
Cesium-137	30	$7.5 \times 10^4$	$3.1 \times 10^{9}$	$5.1 \times 10^{9}$	$6.4 \times 10^{4}$	$1.4 \times 10^{9}$	$2.3 \times 10^{9}$	$4.5 \times 10^{9}$	$7.4 \times 10^{9}$
Promethium-147	2.6	$2.8 \times 10^{2}$	$1.2 \times 10^{7}$	$1.9 \times 10^{7}$	$2.7 \times 10^{2}$	$6.0 \times 10^{6}$	$1.0 \times 10^{7}$	$1.8 \times 10^{7}$	$2.9 \times 10^{7}$
Samarium-151	90	$4.2 \times 10^{2}$	$1.7 \times 10^{7}$	$2.9 \times 10^{7}$	$3.7 \times 10^{2}$	$8.1 \times 10^{6}$	$1.4 \times 10^{7}$	$2.5 \times 10^{7}$	$4.2 \times 10^{7}$
Europium-154	8.6	$2.1 \times 10^{3}$	$8.5 \times 10^{7}$	$1.4 \times 10^{8}$	$1.8 \times 10^{3}$	$3.9 \times 10^{7}$	$6.5 \times 10^{7}$	$1.2 \times 10^{8}$	$2.1 \times 10^{8}$
Europium-155	4.8	$3.6 \times 10^{2}$	$1.5 \times 10^{7}$	$2.4 \times 10^{7}$	$3.2 \times 10^{2}$	$7.0 \times 10^{6}$	$1.2 \times 10^{7}$	$2.2 \times 10^{7}$	$3.6 \times 10^{7}$
Actinium-227	2.2	$1.6 \times 10^{-5}$	$6.5 \times 10^{-1}$	$1.1 \times 10^{0}$	$1.4 \times 10^{-5}$	$3.1 \times 10^{-1}$	$5.2 \times 10^{-1}$	$9.7 \times 10^{-1}$	$1.6 \times 10^{0}$
Thorium-230	$7.5 \times 10^4$	$3.0 \times 10^{-4}$	$1.2 \times 10^{1}$	$2.0 \times 10^{1}$	$2.5 \times 10^{-4}$	$5.5 \times 10^{0}$	$9.1 \times 10^{0}$	$1.8 \times 10^{1}$	$2.9 \times 10^{1}$
Protactinium-231	$3.3 \times 10^4$	$3.4 \times 10^{-5}$	$1.4 \times 10^{0}$	$2.3 \times 10^{0}$	$3.0 \times 10^{-5}$	$6.7 \times 10^{-1}$	$1.1 \times 10^{0}$	$2.1 \times 10^{0}$	$3.4 \times 10^{0}$
Uranium-232	69	$4.5 \times 10^{-2}$	$1.9 \times 10^{3}$	$3.1 \times 10^{3}$	$3.4 \times 10^{-2}$	$7.5 \times 10^{2}$	$1.3 \times 10^{3}$	$2.6 \times 10^{3}$	$4.3 \times 10^{3}$
Uranium-233	$1.6 \times 10^{5}$	$6.8 \times 10^{-5}$	$2.8 \times 10^{0}$	$4.7 \times 10^{0}$	$5.1 \times 10^{-5}$	$1.1 \times 10^{0}$	$1.9 \times 10^{0}$	$3.9 \times 10^{0}$	$6.5 \times 10^{0}$
Uranium-234	$2.5 \times 10^{5}$	$1.4 \times 10^{0}$	$5.8 \times 10^{4}$	$9.6 \times 10^{4}$	$1.2 \times 10^{0}$	$2.6 \times 10^{4}$	$4.3 \times 10^{4}$	$8.4 \times 10^{4}$	$1.4 \times 10^{5}$

**Table A-11.** Total projected radionuclide inventories<sup>a,b</sup> (page 1 of 2).

		Press	urized-water re	eactor	Вс	iling-water rea	ctor		
			Total	curies	-	Total curies		Grand totals (curies)	
	Half life	Curies per	Proposed	Additional	Curies per	Proposed	Additional	Proposed	Additional
Isotope	(yrs.) <sup>c</sup>	MTHM <sup>d</sup>	Action	modules	MTHM	Action	modules	Action	modules
Uranium-235	$7.0 \times 10^{8}$	$1.7 \times 10^{-2}$	$7.1 \times 10^{2}$	$1.2 \times 10^{3}$	$1.4 \times 10^{-2}$	$3.2 \times 10^{2}$	$5.3 \times 10^{2}$	$1.0 \times 10^{3}$	$1.7 \times 10^{3}$
Uranium-236	$2.3 \times 10^7$	$3.1 \times 10^{-1}$	$1.3 \times 10^{4}$	$2.1 \times 10^{4}$	$2.5 \times 10^{-1}$	$5.5 \times 10^{3}$	$9.1 \times 10^{3}$	$1.8 \times 10^{4}$	$3.0 \times 10^{4}$
Uranium-238	$4.5 \times 10^9$	$3.1 \times 10^{-1}$	$1.3 \times 10^{4}$	$2.1 \times 10^{4}$	$3.2 \times 10^{-1}$	$7.0 \times 10^{3}$	$1.2 \times 10^{4}$	$2.0 \times 10^{4}$	$3.3 \times 10^{4}$
Neptunium-237	$2.1 \times 10^6$	$5.2 \times 10^{-1}$	$2.1 \times 10^{4}$	$3.5 \times 10^{4}$	$4.0 \times 10^{-1}$	$8.7 \times 10^{3}$	$1.5 \times 10^{4}$	$3.0 \times 10^{4}$	$5.0 \times 10^{4}$
Plutonium-238	88	$4.1 \times 10^{3}$	$1.7 \times 10^{8}$	$2.8 \times 10^{8}$	$3.3 \times 10^{3}$	$7.4 \times 10^{7}$	$1.2 \times 10^{8}$	$2.4 \times 10^{8}$	$4.0 \times 10^{8}$
Plutonium-239	$2.4 \times 10^4$	$4.0 \times 10^{2}$	$1.6 \times 10^{7}$	$2.7 \times 10^{7}$	$3.3 \times 10^{2}$	$7.3 \times 10^{6}$	$1.2 \times 10^{7}$	$2.4 \times 10^{7}$	$4.0 \times 10^{7}$
Plutonium-240	$6.5 \times 10^3$	$6.0 \times 10^{2}$	$2.5 \times 10^{7}$	$4.1 \times 10^{7}$	$5.2 \times 10^{2}$	$1.1 \times 10^{7}$	$1.9 \times 10^{7}$	$3.6 \times 10^{7}$	$6.0 \times 10^{7}$
Plutonium-241	14	$5.2 \times 10^4$	$2.1 \times 10^{9}$	$3.5 \times 10^{9}$	$5.0 \times 10^{4}$	$1.1 \times 10^{9}$	$1.8 \times 10^{9}$	$3.2 \times 10^{9}$	$5.3 \times 10^{9}$
Plutonium-242	$3.8 \times 10^{5}$	$2.2 \times 10^{0}$	$9.2 \times 10^{4}$	$1.5 \times 10^{5}$	$2.2 \times 10^{0}$	$4.9 \times 10^{4}$	$8.2 \times 10^{4}$	$1.4 \times 10^{5}$	$2.3 \times 10^{5}$
Americium-241	$4.3 \times 10^{2}$	$3.6 \times 10^{3}$	$1.5 \times 10^{8}$	$2.4 \times 10^{8}$	$3.4 \times 10^{3}$	$7.4 \times 10^{7}$	$1.2 \times 10^{8}$	$2.2 \times 10^{8}$	$3.7 \times 10^{8}$
Americium-242/242m	$1.4 \times 10^{2}$	$2.4 \times 10^{1}$	$9.8 \times 10^{5}$	$1.6 \times 10^{6}$	$2.6 \times 10^{1}$	$5.7 \times 10^{5}$	$9.5 \times 10^{5}$	$1.6 \times 10^{6}$	$2.6 \times 10^{6}$
Americium-243	$7.4 \times 10^{3}$	$3.0 \times 10^{1}$	$1.2 \times 10^{6}$	$2.0 \times 10^{6}$	$2.9 \times 10^{1}$	$6.4 \times 10^{5}$	$1.1 \times 10^{6}$	$1.9 \times 10^{6}$	$3.1 \times 10^{6}$
Curium-242	0.45	$2.0 \times 10^{1}$	$8.1 \times 10^{5}$	$1.4 \times 10^{6}$	$2.1 \times 10^{1}$	$4.7 \times 10^{5}$	$7.9 \times 10^{5}$	$1.3 \times 10^{6}$	$2.1 \times 10^{6}$
Curium-243	29	$2.1 \times 10^{1}$	$8.6 \times 10^{5}$	$1.4 \times 10^{6}$	$2.1 \times 10^{1}$	$4.6 \times 10^{5}$	$7.7 \times 10^{5}$	$1.3 \times 10^{6}$	$2.2 \times 10^{6}$
Curium-244	18	$1.9 \times 10^{3}$	$8.0 \times 10^{7}$	$1.3 \times 10^{8}$	$1.9 \times 10^{3}$	$4.3 \times 10^{7}$	$7.1 \times 10^{7}$	$1.2 \times 10^{8}$	$2.0 \times 10^{8}$
Curium-245	$8.5 \times 10^{3}$	$4.6 \times 10^{-1}$	$1.9 \times 10^{4}$	$3.2 \times 10^{4}$	$4.4 \times 10^{-1}$	$9.7 \times 10^{3}$	$1.6 \times 10^{4}$	$2.9 \times 10^{4}$	$4.8 \times 10^{4}$
Curium-246	$4.8 \times 10^{3}$	$1.0 \times 10^{-1}$	$4.2 \times 10^{3}$	$7.0 \times 10^{3}$	$9.5 \times 10^{-2}$	$2.1 \times 10^{3}$	$3.5 \times 10^{3}$	$6.3 \times 10^{3}$	$1.0 \times 10^{4}$

a. Source: Compilation of Tables A-9 and A-10.

b. The radionuclides listed are those used in the most recent repository preclosure safety assessment (DIRS 150276-CRWMS M&O 2000, all) and include all those used in the postclosure impacts analysis (Chapter 5). The radionuclides listed have been revised from the list in the Draft EIS; DOE has determined that the revisions to the list (including both omissions and additions) resulted in no change to the preclosure accident impacts.

c. Half-life is defined as the time in which half of the atoms of a radioactive substance decay to another nuclear form.

d. MTHM = metric tons of heavy metal; 0.18 MTHM per boiling-water reactor assembly and 0.46 MTHM per pressurized-water reactor assembly.

Converting the surface concentration values to total assembly inventory requires estimates of the surface area of the assembly. Conservative estimated surface area values for pressurized-water and boiling-water reactor assemblies currently in operation are 450,000 square centimeters (1,200 square feet) for pressurized-water reactor assemblies and 170,000 square centimeters (460 square feet) for boiling-water reactor assemblies (DIRS 150276-CRWMS M&O 2000, p. VIII-4, 5).

The resulting cobalt-60 crud inventories at discharge from the reactor, therefore, are 450,000 square centimeters  $\times$  140 microcuries per square centimeter = 63 curies for pressurized-water reactor assemblies and 170,000 square centimeters  $\times$  595 microcuries per square centimeter = 100 curies for boiling-water reactor assemblies. Because these values would be at the time of discharge of the fuel from the reactor, the inventories must be corrected for radioactive decay. The half-life (time for half of the radionuclide to decay) of cobalt-60 is 5.27 years. Because the representative fuel assemblies (see Table A-6) are 14 years old for boiling-water reactor fuel and 15 years old for pressurized-water reactor fuel, the cobalt-60 inventories must be reduced by 2.66 (14/5.27) half-lives for boiling-water reactor fuel and 2.85 (15/5.27) half-lives for the pressurized-water reactor fuel. The resulting inventories are then  $100/(2)^{2.66} = 16$  curies per boiling-water reactor assembly and  $63/(2)^{2.85} = 9$  curies per pressurized-water reactor assembly. Because DOE used maximum values for both the surface concentration and surface area, these results are conservative.

Tables A-12 and A-13 list the radionuclide inventories for the representative pressurized-water and boiling-water reactor spent nuclear fuel assemblies, respectively. The list of radionuclides is modified from DIRS 150276-CRWMS M&O (2000, p. VIII-3), which DOE used for preclosure accident analyses. For accident evaluation, the location of the radionuclides on and in the fuel assemblies can be important, so these tables provide this information (DIRS 152476-Sprung et al. 2000, all). Some of the radionuclides are produced by neutron activation of stable elements in the structures of the fuel assembly; these are listed in the Location column. A few radionuclides reside in the gap between the fuel pellet and the cladding; these are also listed in the Location column. The majority of the radionuclides are in the fuel pellets, as listed in the tables, and a few are in both the fuel pellet and the fuel clad gap.

#### A.2.1.5.3 Chemical Composition

Commercial spent nuclear fuel consists of the uranium oxide fuel itself (including actinides, fission products, etc.), the cladding, and the assembly hardware.

Typical pressurized-water and boiling-water reactor fuels consist of uranium dioxide fuel pellets with a zirconium alloy cladding. Some assemblies, however, are clad in stainless-steel 304. These assemblies have been discharged from Haddam Neck, Yankee-Rowe, Indian Point, San Onofre, and LaCrosse and comprise 1.15 percent of the MTHM included in the Proposed Action. Table A-14 lists the number sites, storage locations, and fuel assemblies and MTHM discharged.

Tables A-15 and A-16 list the postirradiation elemental distributions for typical fuels. The data in these tables include the fuel, cladding material, and assembly hardware.

# A.2.1.5.4 Thermal Output

Heat generation rates are available as a function of spent fuel type, enrichment, burnup, and decay time in the Light-Water Reactor Radiological Database, which is an integral part of the *Characteristics of Potential Repository Wastes* (DIRS 102588-DOE 1992, p. 1.1-1). Table A-17 lists the thermal profiles for the average pressurized-water reactor and boiling-water reactor assemblies from the Light-Water Reactor Radiological Database. For the EIS analysis, the typical thermal profile, applied across the proposed inventory, yields a good approximation of the expected thermal load in the repository. Figure A-6 shows these profiles as a function of time.

**Table A-12.** Radionuclide activity for representative pressurized-water reactor fuel assemblies. a,b

	Curies per		<u>-</u>	Curies per	
Radionuclide <sup>c</sup>	assembly	Location	Radionuclide <sup>c</sup>	assembly	Location
Hydrogen-3	$2.0 \times 10^{2}$	Fuel clad gap	Samarium-151	$2.4 \times 10^{2}$	Fuel pellet
Carbon-14	$3.0 \times 10^{-1}$	Fuel clad gap	Europium-154	$1.5 \times 10^{3}$	Fuel pellet
Chlorine-36	$6.3 \times 10^{-3}$	Fuel clad gap	Europium-155	$2.2 \times 10^{2}$	Fuel pellet
Iron-55	$4.0 \times 10^{1}$	Structures	Actinium-227	$1.3 \times 10^{-5}$	Fuel pellet
Cobalt-60	$1.1 \times 10^{3}$	Structures	Thorium-230	$9.9 \times 10^{-5}$	Fuel pellet
Cobalt-60	$8.8 \times 10^{0}$	Surfaces (crud)	Protactinium-231	$3.3 \times 10^{-5}$	Fuel pellet
Nickel-59	$1.9 \times 10^{0}$	Structures	Uranium-232	$2.4 \times 10^{-2}$	Fuel pellet
Nickel-63	$2.5 \times 10^{2}$	Structures	Uranium-233	$3.2 \times 10^{-5}$	Fuel pellet
Selenium-79	$4.6 \times 10^{-2}$	Fuel pellet	Uranium-234	$6.7 \times 10^{-1}$	Fuel pellet
Krypton-85	$2.2 \times 10^{3}$	Fuel clad gap	Uranium-235	$8.8 \times 10^{-3}$	Fuel pellet
Strontium-90	$3.6 \times 10^{4}$	Fuel pellet, gap	Uranium-236	$1.9 \times 10^{-1}$	Fuel pellet
Yttrium-90 <sup>d</sup>	$3.6 \times 10^{4}$	Fuel pellet, gap	Uranium-238	$1.4 \times 10^{-1}$	Fuel pellet
Zirconium-93	$9.8 \times 10^{-1}$	Fuel pellet	Neptunium-237	$2.5 \times 10^{-1}$	Fuel pellet
Niobium-93m	$1.9 \times 10^{1}$	Fuel pellet	Plutonium-238	$2.6 \times 10^{3}$	Fuel pellet
Niobium-94	$8.1 \times 10^{-1}$	Fuel pellet	Plutonium-239	$1.8 \times 10^{2}$	Fuel pellet
Technetium-99	$9.1 \times 10^{0}$	Fuel pellet	Plutonium-240	$3.1 \times 10^{2}$	Fuel pellet
Ruthenium-106	$1.1 \times 10^{1}$	Fuel pellet	Plutonium-241	$3.9 \times 10^{4}$	Fuel pellet
Palladium-107	$7.8 \times 10^{-2}$	Fuel pellet	Plutonium-242	$1.5 \times 10^{0}$	Fuel pellet
Cadmium-113m	$1.2 \times 10^{1}$	Fuel pellet	Americium-241	$1.5 \times 10^{3}$	Fuel pellet
Tin-126	$3.7 \times 10^{-1}$	Fuel pellet	Americium-242m	$7.2 \times 10^{0}$	Fuel pellet
Antimony-125	$1.2 \times 10^{2}$	Fuel pellet	Americium-243	$2.0 \times 10^{1}$	Fuel pellet
Iodine-129	$2.2 \times 10^{-2}$	Fuel clad gap	Curium-242	$5.9 \times 10^{0}$	Fuel pellet
Cesium-134	$7.2 \times 10^{2}$	Fuel pellet, gap	Curium-243	$1.3 \times 10^{1}$	Fuel pellet
Cesium-135	$3.8 \times 10^{-1}$	Fuel pellet, gap	Curium-244	$1.8 \times 10^{3}$	Fuel pellet
Cesium-137	$5.2 \times 10^4$	Fuel pellet, gap	Curium-245	$2.9 \times 10^{-1}$	Fuel pellet
Barium-137m <sup>d</sup>	$5.2 \times 10^4$	Fuel pellet, gap	Curium-246	$9.1 \times 10^{-2}$	Fuel pellet
Promethium-147	$1.7 \times 10^{3}$	Fuel pellet			

a. Source: DIRS 156919-Ikenberry (2001, all).

#### A.2.1.5.5 Physical Parameters

Table A-18 lists reference characteristics of typical pressurized-water and boiling-water reactor fuel assemblies. These data are from the Integrated Data Base Report (DIRS 101815-DOE 1997, p. 1-8) and reflect characteristics of unirradiated assemblies.

For additional details, the Light-Water Reactor Assembly Database contains individual physical descriptions of the fuel assemblies and fuel pins. The Light-Water Reactor Nonfuel Assembly Hardware Database contains physical and radiological descriptions of nonfuel assembly hardware. These databases are integral parts of the *Characteristics of Potential Repository Wastes* (DIRS 102588-DOE 1992, Section 2.8).

#### A.2.2 DOE SPENT NUCLEAR FUEL

# A.2.2.1 Background

At present, DOE stores most of its spent nuclear fuel at three primary locations: the Hanford Site in Washington State, the Idaho National Engineering and Environmental Laboratory in Idaho, and the

b. Burnup = 50,000 MWd/MTHM, enrichment = 4.3 percent, decay time = 15 years.

c. Half-lives are listed in Table A-11.

d. Barium-137m and yttrium-90 are included and are assumed to be in equilibrium with cesium-137 and strontium-90, respectively.

**Table A-13.** Radionuclide activity for representative boiling-water reactor fuel assemblies. a,b,c

	Curies per			Curies per	
Radionuclide <sup>c</sup>	assembly	Location	Isotope	assembly	Location
Hydrogen-3	$6.6 \times 10^{1}$	Fuel clad gap	Samarium-151	$5.3 \times 10^{1}$	Fuel pellet
Carbon-14	$1.6 \times 10^{-1}$	Fuel clad gap	Europium-154	$3.9 \times 10^{2}$	Fuel pellet
Chlorine-36	$2.6 \times 10^{-3}$	Fuel clad gap	Europium-155	$7.5 \times 10^{1}$	Fuel pellet
Iron-55	$1.6 \times 10^{1}$	Structures	Actinium-227	0	Fuel pellet
Cobalt-60	$1.7 \times 10^{2}$	Structures	Thorium-230	$3.3 \times 10^{-5}$	Fuel pellet
Cobalt-60	$1.6 \times 10^{1}$	Surfaces (crud)	Protactinium-231	$1.2 \times 10^{-5}$	Fuel pellet
Nickel-59	$4.5 \times 10^{-1}$	Structures	Uranium-232	$4.6 \times 10^{-3}$	Fuel pellet
Nickel-63	$5.7 \times 10^{1}$	Structures	Uranium-233	0	Fuel pellet
Selenium-79	$1.4 \times 10^{-2}$	Fuel pellet	Uranium-234	$2.1 \times 10^{-1}$	Fuel pellet
Krypton-85	$7.0 \times 10^{2}$	Fuel clad gap	Uranium-235	$2.4 \times 10^{-3}$	Fuel pellet
Strontium-90	$1.1 \times 10^4$	Fuel pellet, gap	Uranium-236	$5.6 \times 10^{-2}$	Fuel pellet
Yttrium-90 <sup>d</sup>	$1.1 \times 10^4$	Fuel pellet, gap	Uranium-238	$5.7 \times 10^{-2}$	Fuel pellet
Zirconium-93	$3.0 \times 10^{-1}$	Fuel pellet	Neptunium-237	$6.0 \times 10^{-2}$	Fuel pellet
Niobium-93m	$5.0 \times 10^{-1}$	Fuel pellet	Plutonium-238	$5.7 \times 10^{-2}$	Fuel pellet
Niobium-94	$1.7 \times 10^{-2}$	Fuel pellet	Plutonium-239	$4.8 \times 10^{1}$	Fuel pellet
Technetium-99	$2.9 \times 10^{0}$	Fuel pellet	Plutonium-240	$1.0 \times 10^{3}$	Fuel pellet
Ruthenium-106	$4.9 \times 10^{0}$	Fuel pellet	Plutonium-241	$1.0 \times 10^{4}$	Fuel pellet
Palladium-107	$2.4 \times 10^{-2}$	Fuel pellet	Plutonium-242	$4.6 \times 10^{-1}$	Fuel pellet
Cadmium-113m	$3.5 \times 10^{0}$	Fuel pellet	Americium-241	$3.7 \times 10^{2}$	Fuel pellet
Tin-126	$1.1 \times 10^{-1}$	Fuel pellet	Americium-242m	$2.1 \times 10^{0}$	Fuel pellet
Antimony-125	$4.3 \times 10^{1}$	Fuel pellet	Americium-243	$4.8 \times 10^{0}$	Fuel pellet
Iodine-129	$6.7 \times 10^{-3}$	Fuel clad gap	Curium-242	$1.7 \times 10^{0}$	Fuel pellet
Cesium-134	$2.3 \times 10^{2}$	Fuel pellet, gap	Curium-243	$2.9 \times 10^{0}$	Fuel pellet
Cesium-135	$1.3 \times 10^{-1}$	Fuel pellet, gap	Curium-244	$3.5 \times 10^{2}$	Fuel pellet
Cesium-137	$1.6 \times 10^4$	Fuel pellet, gap	Curium-245	$3.6 \times 10^{-2}$	Fuel pellet
Barium-137m <sup>d</sup>	$1.6 \times 10^4$	Fuel pellet, gap	Curium-246	$1.8 \times 10^{-2}$	Fuel pellet
Promethium-147	$6.6 \times 10^{2}$	Fuel pellet			_

a. Source: DIRS 156919-Ikenberry (2001, all).

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**Table A-14.** Stainless-steel-clad spent nuclear fuel inventory.<sup>a</sup>

Discharging	Storage		
reactors	locations	Assemblies	$MTHM^b$
5	6	2,187	727

a. Source: DIRS 104353-Cole (1998, all).

Savannah River Site in South Carolina. Some DOE spent nuclear fuel is stored at the Fort St. Vrain dry storage facility in Colorado. DOE issued the *Record of Decision – Department of Energy Programmatic Spent Nuclear Fuel Management and Idaho National Engineering Laboratory Environmental Restoration and Waste Management Programs Final Environmental Impact Statement on June 1, 1995 (DIRS 103205-DOE 1995, all) and amended it in March 1996 (DIRS 147933-DOE 1996, all). The Record of Decision and its amendment specify three primary locations as storage sites for DOE spent nuclear fuel. With the exception of Fort St. Vrain, which will retain its spent nuclear fuel in dry storage, DOE will ship all its spent nuclear fuel from other sites to one of the three primary sites for storage and preparation for ultimate disposition.* 

b. Burnup = 40,000 MWd/MTHM, enrichment = 3.5 percent, decay time = 14 years.

c. Half-lives are listed in Table A-11.

d. Barium-137m and yttrium-90 are included and are assumed to be in equilibrium with cesium-137 and strontium-90, respectively.

b. MTHM = metric tons of heavy metal.

**Table A-15.** Elemental distribution of average pressurized-water reactor fuel.<sup>a</sup>

	Grams per	<u> </u>		Grams per	
Element	assembly <sup>b</sup>	Percent total <sup>c</sup>	Element	assembly <sup>b</sup>	Percent total <sup>c</sup>
Aluminum	47	0.01	Oxygen	62,000	9.35
Americium	600	0.09	Palladium	790	0.12
Barium	1,200	0.18	Phosphorus	85	0.01
Cadmium	77	0.01	Plutonium	4,600	0.69
Carbon	77	0.01	Praseodymium	610	0.09
Cerium	1,300	0.20	Rhodium	230	0.04
Cesium	1,100	0.17	Rubidium	200	0.03
Chromium	4,300	0.65	Ruthenium	1,200	0.18
Cobalt	38	0.01	Samarium	470	0.07
Europium	72	0.01	Silicon	170	0.03
Gadolinium	81	0.01	Silver	40	0.01
Iodine	130	0.02	Strontium	330	0.05
Iron	12,000	1.85	Technetium	420	0.06
Krypton	190	0.03	Tellurium	270	0.04
Lanthanum	670	0.10	Tin	1,900	0.29
Manganese	330	0.05	Titanium	51	0.01
Molybdenum	2,000	0.31	Uranium	440,000	65.78
Neodymium	2,200	0.33	Xenon	2,900	0.43
Neptunium	330	0.05	Yttrium	250	0.04
Nickel	5,000	0.75	Zirconium	120,000	17.77
Niobium	330	0.05			
Nitrogen	49	0.01	Totals	668,637	99.99

a. Source: DIRS 102588-DOE (1992, p. 1.1-1).

During the last four decades, DOE and its predecessor agencies have generated approximately 250 varieties of spent nuclear fuel from weapons production, nuclear propulsion, and research missions. A method described by (DIRS 104385-Fillmore 1998, all) allows grouping of these many varieties of spent nuclear fuel into 16 categories for the repository Total System Performance Assessment. The grouping method uses regulatory requirements to identify the parameters that would affect the performance of DOE spent nuclear fuel in the repository and meet analysis needs for the repository License Application. Three fuel parameters (fuel matrix, fuel compound, and cladding condition) would influence repository performance behavior. The methodology categorizes the characteristics of a select number of fuel types either bound or represent a particular characteristic of the whole category. Table A-19 lists these spent nuclear fuel categories, which continue to provide an accurate description of the DOE fuel characteristics for this Final EIS (DIRS 156369-Arenaz 2001, all).

Table A-19 includes sodium-bonded fuel (Category 14). DOE issued a Record of Decision for the Treatment and Management of Sodium-Bonded Spent Nuclear Fuel in September 2000. Electrometallurgical treatment, the preferred alternative, was chosen for EBR-II reactor fuel and other selected small lots. Fermi blanket fuel may be treated by the electrometallurgical process but the final decision has been deferred. Section A.2.3, which covers data associated with high-level radioactive waste, includes data on waste produced from the treatment of all Category 14 spent nuclear fuel (DIRS 104356-Dirkmaat 1997, p. 7). Therefore, this category is not considered as spent nuclear fuel in the EIS.

b. To convert grams to ounces, multiply by 0.035274.

c. Table only includes elements that constitute at least 0.01 percent of the total; therefore, the total of the percentage column is slightly less than 100 percent.

Table A-16. Elemental distribution of average boiling-water reactor fuel.<sup>a</sup>

Element	Grams per assembly <sup>b</sup>	Percent total <sup>c</sup>	Element	Grams per assembly <sup>b</sup>	Percent total <sup>c</sup>
Aluminum	31	0.01	Nitrogen	25	0.01
Americium	220	0.07	Oxygen	25,000	7.82
Barium	390	0.12	Palladium	270	0.09
Cadmium	27	0.01	Plutonium	1,500	0.48
Carbon	36	0.01	Praseodymium	200	0.06
Cerium	430	0.14	Rhodium	79	0.03
Cesium	390	0.12	Rubidium	64	0.02
Chromium	1,900	0.60	Ruthenium	410	0.13
Cobalt	26	0.01	Samarium	160	0.05
Europium	24	0.01	Silicon	80	0.03
Gadolinium	310	0.10	Strontium	110	0.03
Iodine	43	0.01	Technetium	140	0.04
Iron	5,100	1.63	Tellurium	91	0.03
Krypton	62	0.02	Tin	1,600	0.50
Lanthanum	220	0.07	Titanium	83	0.03
Manganese	160	0.05	Uranium	170,000	55.35
Molybdenum	630	0.20	Xenon	950	0.30
Neodymium	730	0.23	Yttrium	81	0.03
Neptunium	97	0.03	Zirconium	96,000	30.52
Nickel	3,000	0.94			
Niobium	29	0.01	Totals	310,698	99.94

a. Source: DIRS 102588-DOE (1992, p. 1.1-1).

Table A-17. Average assembly thermal profiles.<sup>a</sup>

Years after	Pressurized-	water reactor	Boiling-w	ater reactor
discharge	W/MTHM <sup>b</sup>	W/assembly <sup>c</sup>	W/MTHM	W/assembly <sup>d</sup>
1	10,500	4,800	8,400	1,500
3	3,700	1,700	3,000	550
5	2,200	1,000	1,800	340
10	1,500	670	1,200	220
25	990	450	820	150
30	920	420	770	140
50	670	310	570	100
100	370	170	320	58
300	160	73	140	26
500	120	53	100	19
1,000	66	31	58	11
2,000	35	16	30	5
5,000	22	10	19	3
10,000	16	8	13	3

a. Source: DIRS 102588-DOE (1992, p. 1.1-1).

b. To convert grams to ounces, multiply by 0.035274.

c. Table only includes elements that contribute at least 0.01 percent of the total; therefore, the total of the percentage column is slightly less than 100 percent.

b. W/MTHM = watts per metric ton of heavy metal; to convert metric tons to tons, multiply by 1.1023.

c. W/assembly = watts per assembly; assumes 0.46 MTHM per assembly.

d. Assumes 0.18 MTHM per assembly.

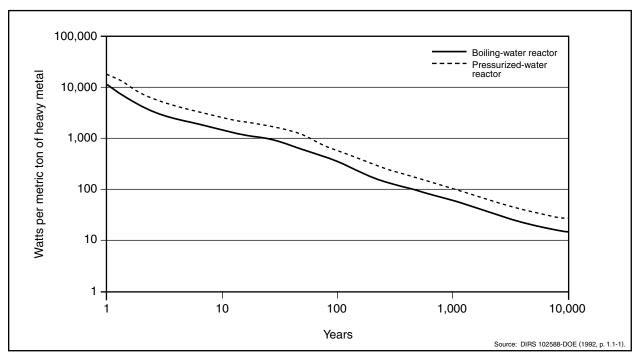


Figure A-6. Average thermal profiles over time.

Table A-18. Reference characteristics for average commercial spent nuclear fuel assemblies.<sup>a</sup>

Characteristics <sup>b</sup>	Boiling-water reactor	Pressurized-water reactor
Overall assembly length (meters)	4.5	4.1
Cross section (centimeters)	$14 \times 14$	$21 \times 21$
Fuel rod length (meters)	4.1	3.9
Active fuel height (meters)	3.8	3.7
Fuel rod outer diameter (centimeters)	1.3	0.95
Fuel rod array	$8 \times 8$	$17 \times 17$
Fuel rods per assembly	63	264
Assembly total weight (kilograms)	320	660
Uranium per assembly (kilograms)	180	460
Uranium oxide per assembly (kilograms)	210	520
Zirconium alloy per assembly (kilograms)	100°	$110^{d}$
Hardware per assembly (kilograms)	8.6 <sup>e</sup>	26 <sup>f</sup>
Nominal volume per assembly (cubic meters)	$0.086^{g}$	$0.19^{g}$

a. Source: DIRS 101815-DOE (1997, p. 1-8).

#### A.2.2.2 Sources

The DOE National Spent Fuel Program maintains a spent nuclear fuel data base (DIRS 153072-Wheatley 2000, all). Table A-19 provides a brief description of each of the fuel categories and a typical fuel. Section A.2.2.5.3 provides more detail on the chemical makeup of each category.

b. To convert meters to feet, multiply by 3.2808; to convert centimeters to inches, multiply by 0.3937; to convert kilograms to pounds, multiply by 2.2046; to convert cubic meters to cubic feet, multiply by 35.314.

c. Includes zirconium alloy fuel rod spacers and fuel channels.

d. Includes zirconium alloy control rod guide thimbles.

e. Includes stainless-steel tie plates, Inconel springs, and plenum springs.

f. Includes stainless-steel nozzles and Inconel-718 grids.

g. Based on overall outside dimension; includes spacing between the stacked fuel rods of the assembly.

Table A-19. DOE spent nuclear fuel categories. a,b,c

	DOE SNF category	Typically from	Description of fuel
1.	Uranium metal	N-Reactor	Uranium metal fuel compounds with aluminum or zirconium alloy cladding
2.	Uranium-zirconium	HWCTR	Uranium alloy fuel compounds with zirconium alloy cladding
3.	Uranium- molybdenum	Fermi	Uranium-molybdenum alloy fuel compounds with zirconium alloy cladding
4.	Uranium oxide, intact	Commercial PWR	Uranium oxide fuel compounds with zirconium alloy or stainless-steel cladding in fair to good condition
5.	Uranium oxide, failed/ declad/aluminum clad	TMI core debris	Uranium oxide fuel compounds: (1) without cladding; (2) clad with zirconium alloy, Hastelloy, nickel-chromium, or stainless steel in poor or unknown condition; or (3) nondegraded aluminum clad
6.	Uranium-aluminide	ATR	Uranium-aluminum alloy fuel compounds with aluminum cladding
7.	Uranium-silicide	FRR MTR	Uranium silicide fuel compounds with aluminum cladding
8.	Thorium/uranium carbide, high-integrity	Fort St. Vrain	Thorium/uranium carbide fuel compounds with graphite cladding in good condition
9.	Thorium/uranium carbide, low-integrity	Peach Bottom	Thorium/uranium carbide fuel compounds with graphite cladding in unknown condition
10.	Plutonium/uranium carbide, nongraphite	FFTF carbide	Uranium carbide or plutonium-uranium carbide fuel compounds with or without stainless-steel cladding
11.	Mixed oxide	FFTF oxide	Plutonium/uranium oxide fuel compounds in zirconium alloy, stainless-steel, or unknown cladding
12.	Uranium/thorium oxide	Shippingport LWBR	Uranium/thorium oxide fuel compounds with zirconium alloy or stainless-steel cladding
13.	Uranium-zirconium hydride	TRIGA	Uranium-zirconium hydride fuel compounds with or without Incalloy, stainless-steel, or aluminum cladding
14.	Sodium-bonded	EBR-II driver and blanket, Fermi-I blanket	Uranium and uranium-plutonium metallic alloy with predominantly stainless-steel cladding
15.	Naval fuel	Surface ship/ submarine	Uranium-based with zirconium alloy cladding
16.	Miscellaneous	Not specified	Various fuel compounds with or without zirconium alloy, aluminum, Hastelloy, tantalum, niobium, stainless-steel or unknown cladding

a. Source: DIRS 104385-Fillmore (1998, all).

#### A.2.2.3 Present Storage and Generation Status

Table A-20 lists storage locations and inventory information on DOE spent nuclear fuels. During the preparation of the *Department of Energy Programmatic Spent Nuclear Fuel Management and Idaho National Engineering Laboratory Environmental Restoration and Waste Management Programs Final Environmental Impact Statement* (DIRS 101802-DOE 1995, all), DOE evaluated and categorized

b. Abbreviations: SNF = spent nuclear fuel; HWCTR = heavy-water cooled test reactor; PWR = pressurized-water reactor; TMI = Three Mile Island; ATR = Advanced Test Reactor; FRR MTR = foreign research reactor – material test reactor; FFTF = Fast Flux Test Facility; LWBR = light-water breeder reactor; TRIGA = Training Research Isotopes—General Atomic; EBR-II = Experimental Breeder Reactor II.

c. For ongoing repository performance analyses, the 16 DOE fuel categories have been reduced to 11 categories (DIRS 118968-DOE 2000, all) since the publication of the Draft EIS. The reduction reflects a better understanding of the behavior of DOE fuels under repository conditions and allows the combining of some of the 16 DOE fuel categories. The reduced DOE fuel categories will help streamline future repository analyses of DOE fuels.

**Table A-20.** National Spent Nuclear Fuel Database projection of DOE spent nuclear fuel locations and inventories to 2035. a,b

	Fuel category and name	Storage Site	No. of units <sup>c</sup>	Mass (kilograms) <sup>d</sup>	Volume (cubic meters) <sup>e</sup>	Fissile mass (kilograms)	Equivalent uranium mass (kilograms)	МТНМ
1.	Uranium metal <sup>f</sup>	INEEL Hanford SRS <i>Totals</i>	85 100,000 350 100,435	4,500 2,160,000 120,000 2,284,500	0.7 200 18 218.7	13 25,000 110 25,123	1,700 2,100,000 17,000 2,118,700	1.7 2100 17 2119
2.	Uranium-zirconium	INEEL	69	120	0.7	34	40	0.04
3.	Uranium-molybdenum	INEEL	29,000	4,600	0.3	970	3,800	3.8
4.	Uranium oxide, intact	INEEL Hanford <i>Totals</i>	14,000 87 14,087	150,000 44,000 <i>194,000</i>	41 11 52	2,200 240 2,440	80,000 18,000 <i>98,000</i>	80 18 99
5.	Uranium oxide, failed/declad/aluminum clad	INEEL Hanford SRS <i>Totals</i>	2,000 13 7,600 9,613	340,000 270 58,000 398,270	140 4.2 96 240.2	2,200 4 2,600 4,804	83,000 160 3,200 86,360	84 0.2 3.2 87
6.	Uranium-aluminide	SRS	18,000	130,000	150	6,000	8,800	8.7
7.	Uranium-silicide	SRS	7,400	47,000	53	1,200	12,000	12
8.	Thorium/uranium carbide, high-integrity	FSV INEEL <i>Totals</i>	1,500 1,600 3,100	190,000 130,000 <i>320,000</i>	130 82 212	640 350 990	820 440 1,260	15 9.9 25
9.	Thorium/uranium carbide, low-							
	integrity	INEEL	810	55,000	17	180	210	1.7
10.	Plutonium/uranium carbide, nongraphite	INEEL Hanford <i>Totals</i>	130 2 132	140 330 <i>470</i>	0 0.1 <i>0.1</i>	10 11 21	73 64 <i>137</i>	0.08 0.07 0.2
11.	Mixed oxide	INEEL Hanford <i>Totals</i>	2,000 620 2,620	6,100 110,000 <i>116,100</i>	2.4 33 <i>35.1</i>	240 2,400 2,640	2,000 8,000 10,000	2.1 10 <i>12</i>
12.	Uranium/thorium oxide	INEEL	260	120,000	18	810	810	50
13.	Uranium-zirconium hydride	INEEL Hanford <i>Totals</i>	9,800 190 9,990	33,000 660 <i>33,660</i>	8.1 33 8. <i>3</i>	460 7 467	2,000 36 2,036	2 0.04 2
15.	Naval fuel <sup>g</sup>	INEEL	300	4,400,000	888	64,000	65,000	65
16.	Miscellaneous	INEEL Hanford SRS	1,500 73 8,800	33,000 1,700 9,200	11 0.2 8.2	360 30 550	5,500 130 2,900	7.7 0.2 2.9
Gran	d totals	Totals	10,373 <b>210,000</b>	43,900 <b>8,150,000</b>	19.4 <b>1,900</b>	940 <b>110,000</b>	8,530 <b>2,420,000</b>	11 <b>2,500</b>

Source: DIRS 148240-Dirkmaat (1998, all).

all the materials listed in the table as spent nuclear fuel, in accordance with the definition in the Nuclear Waste Policy Act, as amended.

# A.2.2.4 Final Spent Nuclear Fuel Form

For all spent nuclear fuel categories except 14, the expected final spent nuclear fuel form does not differ from the current or planned storage form. Before its disposal in the repository, candidate material would be in compliance with approved acceptance criteria.

b. Abbreviations: SNF = spent nuclear fuel; INEEL = Idaho National Engineering and Environmental Laboratory; SRS = Savannah River Site; FSV = Fort St. Vrain.

c. Unit is defined as an assembly, bundle of elements, can of material, etc., depending on the particular spent nuclear fuel category.

d. To convert kilograms to pounds, multiply by 2.2046; to convert metric tons to tons, multiply by 1.1023.

e. To convert cubic meters to cubic yards, multiply by 1.3079.

f. N-Reactor fuel is stored in aluminum or stainless-steel cans at the K-East and K-West Basins. The mass listed in this table does not include the storage cans.

g. Information supplied by the Navy (DIRS 104356-Dirkmaat 1997, Attachment, p. 2).

DOE has prepared an EIS at the Savannah River Site (DIRS 156897-DOE 2000, all) to evaluate potential treatment alternatives for spent nuclear fuel and its ultimate disposal in the repository. The products of any proposed treatment of the Savannah River Site aluminum-based fuels are adequately represented by the properties of the present aluminum-based fuel (Categories 6, 7, and part of 5) for this Yucca Mountain EIS. They are bounded by the same total radionuclide inventory, heat generation rates, dissolution rates, and number of canisters. No additional data about the products will be required to ensure that they are represented in the EIS inventory.

# A.2.2.5 Spent Nuclear Fuel Characteristics

#### A.2.2.5.1 Mass and Volume

Table A-20 lists total volume, mass, and MTHM for each DOE spent nuclear fuel category from the National Spent Nuclear Fuel Database (DIRS 153072-Wheatley 2000, all).

# A.2.2.5.2 Amount and Nature of Radioactivity

ORIGEN2 (Oak Ridge Isotope Generation), an accepted computer code for calculating spent nuclear fuel radionuclide inventories, was used to generate activity data for radionuclides in the DOE spent nuclear fuel inventory. The inventory came from the 1997 version of the National Spent Nuclear Fuel Database (DIRS 153072-Wheatley 2000, all).

Table A-21 lists the activities expressed in terms of curies per handling unit for the radionuclides of interest (uranium, fission products and actinides). The table lists activity estimates decayed to 2030 for all categories except 15. A handling unit for DOE is a spent nuclear fuel canister. The canister quantities (except the naval fuel) are estimated based on the fuel's current as-stored condition at each of the DOE facilities. The planned storage, transportation, and disposal unit for naval spent nuclear fuel is a canister. Each naval spent nuclear fuel canister would contain several spent fuel assemblies. The actual canister quantities for repository disposition could be different depending on final package configuration and whether the fuels were treated as discussed in Section A.1.1.3.

The activity for naval spent nuclear fuel (Category 15) is provided for a representative naval canister. DIRS 104356-Dirkmaat (1997, Attachment A, Table 3) provided these activities for 5 years after shutdown, which would be the minimum cooling time before naval fuel would reach the repository. The power history assumed operations at power for a full core life. The assumptions about the power history and minimum cooling time conservatively bound the activity for naval fuel that would be emplaced in a monitored geologic repository. In addition, ORIGEN-S was used to calculate the activity associated with activation products in the cladding, which are listed in Table A-21. For completeness, the data also include the activity that would be present in the activated corrosion products deposited on the fuel.

#### A.2.2.5.3 Chemical Composition

This section discusses the chemical compositions of each of the 16 categories of DOE spent nuclear fuel (DIRS 148240-Dirkmaat 1998, all).

• Category 1: Uranium metal. The fuel in this category consists primarily of uranium metal. N-reactor fuel represents the category because its mass is so large that the performance of the rest of the fuel in the category, even if greatly different from N-Reactor fuel, would not change the overall category performance. The fuel is composed of uranium metal about 1.25 percent enriched in uranium-235, and is clad with a zirconium alloy. Approximately 50 percent of the fuel elements are believed to have failed cladding. This fuel typically has low burnup. Another contributor to this category is the Single Pass Reactor fuel at the Hanford Site.

**Table A-21.** Radionuclide activity by DOE spent nuclear fuel category<sup>a</sup> (page 1 of 2).

				, ,	1			$\mathcal{C}$	C	/					
	Category														
Storage	1	2	3	4	5	6	7	8	9	10	11	12	13	15	16
siteb							Numb	er of handli	ng units						
Hanford	440	0	0	34	1	0	0	0	0	2	324	0	3	0	5
INEEL	6	8	70	195	406	0	0	503 <sup>d</sup>	60	3	43	71	97	300	39
SRS	9	0	0	0	425	750	225	0	0	0	0	0	0	0	2
Totals	455	8	70	229	832	750	225	503	60	5	367	71	100	300	46
Radio- nuclide <sup>g</sup>							Curie	s per handli	ing unit						
Ac-227	2.2×10 <sup>-5</sup>	4.8×10 <sup>-9</sup>	6.9×10 <sup>-6</sup>	1.7×10 <sup>-4</sup>	$1.4 \times 10^{-5}$	3.4×10 <sup>-7</sup>	2.3×10 <sup>-7</sup>	0	2.8×10 <sup>-3</sup>	8.9×10 <sup>-9</sup>	1.5×10 <sup>-9</sup>	4.3×10 <sup>-1</sup>	5.6×10 <sup>-8</sup>	9.8×10 <sup>-5</sup>	6.8×10 <sup>-7</sup>
Am-241	$1.1 \times 10^{3}$	3.9×10 <sup>-1</sup>	4.6×10 <sup>-5</sup>	$1.6 \times 10^{3}$	7.3	3.3	$3.6 \times 10^{1}$	3.7	2.7	$2.4 \times 10^{2}$	$4.3 \times 10^{2}$	8.3×10 <sup>-1</sup>	2.0×10 <sup>-1</sup>	5.0×10 <sup>1</sup>	$1.2 \times 10^{2}$
Am-242m		1.2×10 <sup>-3</sup>	0	2.6	1.4×10 <sup>-2</sup>	2.3×10 <sup>-3</sup>	1.3×10 <sup>-2</sup>	1.0×10 <sup>-3</sup>	1.4×10 <sup>-3</sup>	4.1×10 <sup>-1</sup>	7.5×10 <sup>-1</sup>	8.7×10 <sup>-3</sup>	2.3×10 <sup>-3</sup>	4.6×10 <sup>-1</sup>	1.5×10 <sup>-1</sup>
Am-243	$2.8 \times 10^{-1}$	$3.8 \times 10^{-3}$	$7.3 \times 10^{-13}$		2.2×10 <sup>-2</sup>	$2.5 \times 10^{-3}$	$3.6 \times 10^{-2}$	$2.7 \times 10^{-2}$	$1.3 \times 10^{-3}$	$6.7 \times 10^{-3}$	1.8×10 <sup>-1</sup>	$1.7 \times 10^{-3}$	$2.5 \times 10^{-4}$	$6.7 \times 10^{-1}$	4.9×10 <sup>-1</sup>
C-14	1.5	8.2×10 <sup>-6</sup>	$2.2 \times 10^{-3}$	$1.0 \times 10^{-1}$	$1.1 \times 10^{-3}$	9.9×10 <sup>-7</sup>	$1.8 \times 10^{-5}$	2.2×10 <sup>-1</sup>	3.7×10 <sup>-2</sup>	1.5×10 <sup>-5</sup>	9.9×10 <sup>-4</sup>	6.7×10 <sup>-1</sup>	8.5×10 <sup>-2</sup>	1.6×10 <sup>1</sup>	1.7×10 <sup>-3</sup>
Cf-252	e													1.2×10 <sup>-6</sup>	
Cl-36	0	0	$5.6 \times 10^{-6}$	3.5×10 <sup>-4</sup>	1.7×10 <sup>-5</sup>	0	0	2.7×10 <sup>-3</sup>	1.1×10 <sup>-3</sup>	0	1.1×10 <sup>-5</sup>	1.5×10 <sup>-2</sup>	2.6×10 <sup>-3</sup>	6.9×10 <sup>-1</sup>	4.2×10 <sup>-6</sup>
Cm-242	$< 7.4 \times 10^{1}$	$< 7.4 \times 10^{1}$	0	$< 7.4 \times 10^{1}$	$< 7.3 \times 10^{1}$	$< 7.4 \times 10^{1}$	1.4	$< 7.4 \times 10^{1}$							
Cm-243														7.9×10 <sup>-1</sup>	
Cm-244	8.5	1.6×10 <sup>-1</sup>	6.8×10 <sup>-14</sup>	$3.5 \times 10^{2}$	9.3×10 <sup>-1</sup>	2.1×10 <sup>-2</sup>	3.0×10 <sup>-1</sup>	8.3×10 <sup>-1</sup>	3.5×10 <sup>-2</sup>	2.8×10 <sup>-1</sup>	7.6	1.6×10 <sup>-1</sup>	$6.8 \times 10^{-3}$	6.3×10 <sup>1</sup>	1.9×10 <sup>1</sup>
Cm-245	$3.6 \times 10^{-3}$	$8.0 \times 10^{-6}$	1.9×10 <sup>-19</sup>		$3.8 \times 10^{-4}$	1.8×10 <sup>-6</sup>	2.0×10 <sup>-5</sup>	1.4×10 <sup>-4</sup>	4.0×10 <sup>-6</sup>	1.4×10 <sup>-5</sup>	$3.1 \times 10^{-3}$	3.3×10 <sup>-5</sup>	$1.4 \times 10^{-7}$	$7.2 \times 10^{-3}$	$7.1 \times 10^{-3}$
Cm-246	5.3×10 <sup>-4</sup>	$5.5 \times 10^{-7}$	6.1×10 <sup>-23</sup>	2.4×10 <sup>-2</sup>	$6.4 \times 10^{-5}$	$8.6 \times 10^{-8}$	1.5×10 <sup>-6</sup>	6.9×10 <sup>-5</sup>	1.3×10 <sup>-7</sup>	9.7×10 <sup>-7</sup>	5.3×10 <sup>-4</sup>	2.2×10 <sup>-6</sup>	3.9×10 <sup>-9</sup>	$1.4 \times 10^{-3}$	1.2×10 <sup>-3</sup>
Cm-247														9.4×10 <sup>-9</sup>	
Cm-248														$2.6 \times 10^{-8}$	
Co-60	1.4×10 <sup>-1</sup>	0	1.1×10 <sup>-2</sup>	$1.8 \times 10^{1}$	1.6×10 <sup>-12</sup>				2.5×10 <sup>-2</sup>	1.8	1.4	4.3	1.8×10 <sup>-1</sup>	$3.7 \times 10^{3(f)}$	
Cs-134	$2.7 \times 10^{-1}$	$4.6 \times 10^{-2}$	1.9×10 <sup>-8</sup>	9.6×10 <sup>-2</sup>	$8.3 \times 10^{-3}$	$1.7 \times 10^{-1}$	$3.7 \times 10^{-1}$	$7.6 \times 10^{-3}$	$3.6 \times 10^{-7}$	$3.4 \times 10^{-2}$	$7.5 \times 10^{-3}$	$6.0 \times 10^{-3}$	$3.3 \times 10^{-4}$	$8.4 \times 10^4$	5.7×10 <sup>-1</sup>
Cs-135	1.8×10 <sup>-1</sup>	$7.7 \times 10^{-3}$	$4.5 \times 10^{-3}$	1.8×10 <sup>-1</sup>	$2.9 \times 10^{-2}$	$2.8 \times 10^{-2}$	1.9×10 <sup>-2</sup>	1.7×10 <sup>-2</sup>	2.6×10 <sup>-2</sup>	1.4×10 <sup>-2</sup>	$3.2 \times 10^{-3}$	2.0×10 <sup>-1</sup>	$3.2 \times 10^{-2}$	4.6	1.4×10 <sup>-1</sup>
Cs-137	$2.0 \times 10^4$	$7.4 \times 10^3$	0	$2.9 \times 10^4$	$3.6 \times 10^{3}$	$3.8 \times 10^{3}$	$8.1 \times 10^3$	$2.4 \times 10^{3}$	$1.9 \times 10^{3}$	$1.5 \times 10^4$	$4.0 \times 10^{3}$	$2.5 \times 10^{3}$	$3.1 \times 10^3$	$4.5 \times 10^{5}$	$8.7 \times 10^4$
H-3	$2.3 \times 10^{1}$	4.4	8.6×10 <sup>-2</sup>	$3.6 \times 10^{1}$	1.3	5.9×10 <sup>-1</sup>	$1.3 \times 10^{1}$	2.0	1.5	7.3	2.8	$2.3 \times 10^{1}$	9.6×10 <sup>-1</sup>	$1.4 \times 10^{3}$	1.3×10 <sup>1</sup>
I-129	$1.6 \times 10^{-2}$	$1.6 \times 10^{-3}$	1.2×10 <sup>-4</sup>	1.8×10 <sup>-2</sup>	$7.5 \times 10^{-4}$	$1.8 \times 10^{-3}$	$3.8 \times 10^{-3}$	$2.1 \times 10^{-3}$	7.3×10 <sup>-4</sup>	$2.9 \times 10^{-3}$	$3.6 \times 10^{-4}$	1.1×10 <sup>-2</sup>	$7.2 \times 10^{-4}$	1.2×10 <sup>-1</sup>	2.3×10 <sup>-2</sup>
Kr-85	$3.6 \times 10^{2}$	$9.3 \times 10^{1}$	7.7×10 <sup>-1</sup>	$3.1 \times 10^{2}$	$2.7 \times 10^{1}$	$1.3 \times 10^{2}$	$2.6 \times 10^{2}$	$6.0 \times 10^{1}$	7.2	$4.8 \times 10^{1}$	$2.4 \times 10^{1}$	$6.2 \times 10^{2}$	$1.7 \times 10^{1}$	$3.6 \times 10^4$	$4.2 \times 10^{2}$
Nb-93m	8.0×10 <sup>-1</sup>	$8.7 \times 10^{-3}$	$4.6 \times 10^{-3}$	6.7×10 <sup>-1</sup>	1.1×10 <sup>-2</sup>	1.6×10 <sup>-2</sup>	3.1×10 <sup>-2</sup>	$9.2 \times 10^{-3}$	$4.6 \times 10^{-2}$	1.5×10 <sup>-2</sup>	1.3×10 <sup>-2</sup>	3.1×10 <sup>-1</sup>	$7.1 \times 10^{-3}$	3.6	1.7×10 <sup>-1</sup>
Nb-94	5.7×10 <sup>-6</sup>	$1.6 \times 10^{-6}$	8.4×10 <sup>-4</sup>	7.3×10 <sup>-3</sup>	4.2×10 <sup>-5</sup>	3.1×10 <sup>-6</sup>	$7.4 \times 10^{-6}$	1.3×10 <sup>-4</sup>	4.9×10 <sup>-4</sup>	$2.9 \times 10^{-6}$	1.9×10 <sup>-5</sup>	1.6×10 <sup>-2</sup>	$4.6 \times 10^{-3}$	$1.8 \times 10^{2}$	3.5×10 <sup>-5</sup>
Ni-59	8.2×10 <sup>-2</sup>	0	6.9×10 <sup>-3</sup>	9.4×10 <sup>-2</sup>	2.3×10 <sup>-4</sup>	0	0	1.7×10 <sup>-2</sup>	1.5×10 <sup>-3</sup>	0	$2.1 \times 10^{-3}$	5.1×10 <sup>-2</sup>	5.0×10 <sup>-1</sup>	$6.3 \times 10^{1}$	8.2×10 <sup>-4</sup>
Ni-63	7.7	0	1.4×10 <sup>-1</sup>	$3.0 \times 10^{2}$	2.5×10 <sup>-2</sup>	2.3×10 <sup>-22</sup>		4.1×10 <sup>-1</sup>	1.5×10 <sup>-1</sup>	5.0	8.7	6.2	$6.2 \times 10^{1}$	$7.8 \times 10^{3}$	1.0×10 <sup>-1</sup>
Np-237	1.7×10 <sup>-1</sup>	$2.0 \times 10^{-2}$	3.3×10 <sup>-4</sup>	1.8×10 <sup>-1</sup>	$3.1 \times 10^{-3}$	1.2×10 <sup>-2</sup>	1.8×10 <sup>-2</sup>	1.6×10 <sup>-2</sup>	$7.4 \times 10^{-3}$	3.7×10 <sup>-2</sup>	$6.5 \times 10^{-3}$	7.1×10 <sup>-4</sup>	$1.9 \times 10^{-3}$	1.6	2.4×10 <sup>-1</sup>
Pa-231	5.8×10 <sup>-5</sup>	2.3×10 <sup>-7</sup>	2.0×10 <sup>-5</sup>	3.0×10 <sup>-4</sup>	2.6×10 <sup>-5</sup>	4.2×10 <sup>-6</sup>	2.8×10 <sup>-6</sup>	1.9×10 <sup>-2</sup>	4.8×10 <sup>-3</sup>	4.1×10 <sup>-7</sup>	1.2×10 <sup>-7</sup>	1.1	9.0×10 <sup>-7</sup>	5.2×10 <sup>-4</sup>	1.0×10 <sup>-5</sup>
Pb-210	3.2×10 <sup>-10</sup>	$8.6 \times 10^{-13}$	1.4×10 <sup>-10</sup>	$9.0 \times 10^{-8}$	5.2×10 <sup>-9</sup>	2.1×10 <sup>-11</sup>	1.2×10 <sup>-11</sup>	4.6×10 <sup>-6</sup>	2.6×10 <sup>-7</sup>	1.5×10 <sup>-12</sup>	3.1×10 <sup>-10</sup>	7.8×10 <sup>-5</sup>	1.4×10 <sup>-12</sup>	8.9×10 <sup>-7</sup>	$7.510^{-10}$

**Table A-21.** Radionuclide activity by DOE spent nuclear fuel category<sup>a</sup> (page 2 of 2).

								Category <sup>b</sup>							
Radio-	1	2	3	4	5	6	7	8	9	10	11	12	13	15	16
nuclideg							Curies	per handli	ng unit						
Pd-107	$3.3 \times 10^{-2}$	1.1×10 <sup>-3</sup>	1.3×10 <sup>-4</sup>	4.8×10 <sup>-2</sup>	8.3×10 <sup>-4</sup>	9.3×10 <sup>-4</sup>	3.5×10 <sup>-3</sup>	8.7×10 <sup>-4</sup>	4.8×10 <sup>-4</sup>	$2.0 \times 10^{-3}$	1.0×10 <sup>-3</sup>	$2.4 \times 10^{-3}$	$6.0 \times 10^{-4}$	6.0×10 <sup>-2</sup>	1.8×10 <sup>-2</sup>
Pu-238	$2.5 \times 10^{2}$	4.3×10 <sup>1</sup>	1.7×10 <sup>-2</sup>	$1.2 \times 10^{3}$	5.8	1.7×10 <sup>1</sup>	2.8×10 <sup>1</sup>	8.1×10 <sup>1</sup>	1.8×10 <sup>1</sup>	$1.1 \times 10^{2}$	7.9×10 <sup>1</sup>	2.8	2.1	1.2×10 <sup>4</sup>	$5.3 \times 10^{2}$
Pu-239	$5.1 \times 10^{2}$	1.1	2.0	$1.5 \times 10^{2}$	$1.3 \times 10^{1}$	2.4	$2.2 \times 10^{1}$	2.3×10 <sup>-1</sup>	$4.1 \times 10^{-1}$	$1.9 \times 10^{2}$	$3.2 \times 10^{2}$	$1.8 \times 10^{-1}$	4.5	$1.2 \times 10^{1}$	$5.2 \times 10^{1}$
Pu-240	$3.0 \times 10^{2}$	$6.1 \times 10^{-1}$	$6.1 \times 10^{-3}$	$2.4 \times 10^{2}$	4.4	1.2	$1.6 \times 10^{1}$	$3.8 \times 10^{-1}$	$3.2 \times 10^{-1}$	$1.6 \times 10^{2}$	$2.8 \times 10^{2}$	$1.0 \times 10 - 1$	1.8	$1.4 \times 10^{1}$	$3.7 \times 10^{1}$
Pu-241	$3.8 \times 10^{3}$	$2.1 \times 10^{2}$	$6.0 \times 10^{-4}$	$1.4 \times 10^4$	$2.9 \times 10^{2}$	$6.3 \times 10^{1}$	$7.0 \times 10^{2}$	0	$3.0 \times 10^{1}$	$1.7 \times 10^{3}$	$2.6 \times 10^{3}$	$2.4 \times 10^{1}$	$1.3 \times 10^{2}$	$4.0 \times 10^{3}$	$3.5 \times 10^{3}$
Pu-242	1.6×10 <sup>-1</sup>	$9.2 \times 10^{-4}$	$3.8 \times 10^{-11}$	9.1×10 <sup>-1</sup>	$3.0 \times 10^{-3}$	$9.9 \times 10^{-4}$	$1.6 \times 10^{-2}$	0	$4.2 \times 10^{-4}$	$1.6 \times 10^{-3}$	$2.0 \times 10^{-2}$	2.3×10 <sup>-4</sup>	$2.5 \times 10^{-4}$	$8.0 \times 10^{-2}$	$7.0 \times 10^{-2}$
Ra-226	$4.6 \times 10^{-6}$	2.2×10 <sup>-12</sup>	6.5×10 <sup>-10</sup>	2.6×10 <sup>-7</sup>	$2.0 \times 10^{-8}$	$3.8 \times 10^{-10}$	2.3×10 <sup>-10</sup>	$4.9 \times 10^{-6}$	9.3×10 <sup>-7</sup>	2.3×10 <sup>-9</sup>	5.3×10 <sup>-9</sup>	$4.5 \times 10^{-5}$	2.3×10 <sup>-12</sup>	$5.4 \times 10^{-6}$	4.1×10 <sup>-9</sup>
Ra-228	$3.7 \times 10^{-10}$	1.2×10 <sup>-13</sup>	$4.0 \times 10^{-9}$	1.3×10 <sup>-4</sup>	1.1×10 <sup>-5</sup>	$7.3 \times 10^{-13}$	1.1×10 <sup>-12</sup>	$6.5 \times 10^{-3}$	$2.4 \times 10^{-3}$	6.9×10 <sup>-13</sup>	2.0×10 <sup>-11</sup>	$7.1 \times 10^{-2}$	3.5×10 <sup>-9</sup>	1.8×10 <sup>-7</sup>	1.5×10 <sup>-11</sup>
Rh-102														$2.8 \times 10^{-2}$	
Ru-106	$3.1 \times 10^{-5}$	6.3×10 <sup>-7</sup>	$3.1 \times 10^{-15}$	$3.9 \times 10^{-7}$	1.2×10 <sup>-6</sup>	1.3×10 <sup>-5</sup>	4.2×10 <sup>-5</sup>	3.2×10 <sup>-9</sup>	$3.0 \times 10^{-15}$	$2.6 \times 10^{-6}$	3.1×10 <sup>-8</sup>	$2.2 \times 10^{-10}$	1.5×10 <sup>-9</sup>	$6.0 \times 10^{3}$	$5.7 \times 10^{-5}$
Se-79	$2.6 \times 10^{-1}$	3.0×10 <sup>-2</sup>	$1.7 \times 10^{-3}$	1.9×10 <sup>-1</sup>	1.6×10 <sup>-2</sup>	$5.0 \times 10^{-2}$	1.0×10 <sup>-1</sup>	2.9×10 <sup>-2</sup>	$1.4 \times 10^{-2}$	5.2×10 <sup>-2</sup>	$3.6 \times 10^{-3}$	2.5×10 <sup>-1</sup>	1.3×10 <sup>-2</sup>	$3.4 \times 10^{-1}$	$4.7 \times 10^{-1}$
Sm-151	$3.3 \times 10^{2}$	$2.7 \times 10^{1}$	6.9	$5.3 \times 10^{2}$	$2.5 \times 10^{1}$	$4.2 \times 10^{1}$	$3.4 \times 10^{1}$	$4.5 \times 10^{1}$	$2.6 \times 10^{1}$	$1.8 \times 10^{2}$	$2.4 \times 10^{2}$	$9.1 \times 10^{1}$	$2.4 \times 10^{1}$	$1.4 \times 10^{3}$	$3.8 \times 10^{2}$
Sn-126	3.5×10 <sup>-1</sup>	2.6×10 <sup>-2</sup>	$3.8 \times 10^{-3}$	2.4×10 <sup>-1</sup>	1.2×10 <sup>-2</sup>	1.7×10 <sup>-2</sup>	4.1×10 <sup>-2</sup>	$1.4 \times 10^{-2}$	1.2×10 <sup>-2</sup>	$4.7 \times 10^{-2}$	$4.8 \times 10^{-3}$	$2.8 \times 10^{-1}$	$1.2 \times 10^{-2}$	1.2	$3.3 \times 10^{-1}$
Sr-90	$1.6 \times 10^4$	$7.1 \times 10^3$	0	$2.1 \times 10^4$	$3.2 \times 10^{3}$	$3.7 \times 10^{3}$	$7.6 \times 10^3$	$2.3 \times 10^{3}$	$1.8 \times 10^{3}$	$1.3 \times 10^4$	$1.6 \times 10^{3}$	$2.6 \times 10^{3}$	$2.9 \times 10^{3}$	$4.4 \times 10^{5}$	$8.3 \times 10^4$
Tc-99	7.7	9.9×10 <sup>-1</sup>	4.5×10 <sup>-2</sup>	6.6	4.2×10 <sup>-1</sup>	1.0	2.2	7.4×10 <sup>-1</sup>	$4.1 \times 10^{-1}$	1.8	1.3×10 <sup>-1</sup>	2.3	4.3×10 <sup>-1</sup>	$7.0 \times 10^{1}$	$1.4 \times 10^{1}$
Th-229	$3.9 \times 10^{-8}$	1.1×10 <sup>-10</sup>	2.4×10 <sup>-9</sup>	$4.0 \times 10^{-4}$	$3.2 \times 10^{-5}$	2.2×10 <sup>-9</sup>	1.2×10 <sup>-9</sup>	2.8×10 <sup>-2</sup>	$6.8 \times 10^{-3}$	2.5×10 <sup>-10</sup>	1.7×10 <sup>-9</sup>	1.8×10 <sup>-1</sup>	1.2×10 <sup>-9</sup>	9.4×10 <sup>-6</sup>	8.7×10 <sup>-9</sup>
Th-230	$4.4 \times 10^{-6}$	8.6×10 <sup>-9</sup>	$1.2 \times 10^{-7}$	$3.7 \times 10^{-5}$	2.9×10 <sup>-6</sup>	$1.8 \times 10^{-7}$	1.2×10 <sup>-7</sup>	$1.9 \times 10^{-3}$	1.3×10 <sup>-4</sup>	5.1×10 <sup>-7</sup>	1.2×10 <sup>-6</sup>	$6.9 \times 10^{-3}$	3.9×10 <sup>-9</sup>	$1.8 \times 10^{-3}$	1.2×10 <sup>-6</sup>
Th-232	$5.1 \times 10^{-10}$	2.0×10 <sup>-12</sup>	4.3×10 <sup>-9</sup>	$1.4 \times 10^{-4}$	1.2×10 <sup>-5</sup>	1.9×10 <sup>-11</sup>	3.0×10 <sup>-11</sup>	$5.1 \times 10^{-3}$	2.5×10 <sup>-3</sup>	4.4×10 <sup>-12</sup>	5.5×10 <sup>-11</sup>	$8.4 \times 10^{-2}$	$1.0 \times 10^{-8}$	2.3×10 <sup>-7</sup>	$9.8 \times 10^{-11}$
U-232	9.9×10 <sup>-5</sup>	3.5×10 <sup>-5</sup>	1.9×10 <sup>-6</sup>	0	2.2×10 <sup>-5</sup>	1.7×10 <sup>-4</sup>	1.4×10 <sup>-4</sup>	2.3	2.4×10 <sup>-1</sup>	0	0	$7.1 \times 10^{2}$	$2.4 \times 10^{-5}$	5.6×10 <sup>-1</sup>	$3.5 \times 10^{-4}$
U-233	2.5×10 <sup>-5</sup>	9.1×10 <sup>-7</sup>	$9.9 \times 10^{-7}$	1.6×10 <sup>-1</sup>	1.2×10 <sup>-2</sup>	2.6×10 <sup>-6</sup>	1.8×10 <sup>-6</sup>	6.9	2.6	1.7×10 <sup>-6</sup>	9.3×10 <sup>-7</sup>	$1.2 \times 10^{2}$	5.6×10 <sup>-6</sup>	$3.1 \times 10^{-3}$	1.6×10 <sup>-5</sup>
U-234	2.0	8.6×10 <sup>-4</sup>	5.0×10 <sup>-4</sup>	1.7×10 <sup>-1</sup>	1.1×10 <sup>-2</sup>	2.2×10 <sup>-3</sup>	1.8×10 <sup>-3</sup>	5.6×10 <sup>-1</sup>	4.4×10 <sup>-1</sup>	$4.9 \times 10^{-3}$	$8.0 \times 10^{-3}$	5.9	2.1×10 <sup>-4</sup>	$1.5 \times 10^{1}$	1.8×10 <sup>-2</sup>
U-235	$8.4 \times 10^{-2}$	8.2×10 <sup>-3</sup>	3.2×10 <sup>-2</sup>	1.7×10 <sup>-2</sup>	1.2×10 <sup>-2</sup>	1.8×10 <sup>-2</sup>	1.3×10 <sup>-2</sup>	$2.2 \times 10^{-3}$	$6.8 \times 10^{-3}$	1.5×10 <sup>-2</sup>	2.2×10 <sup>-4</sup>	$4.0 \times 10^{-4}$	$9.9 \times 10^{-3}$	2.9×10 <sup>-1</sup>	1.2×10 <sup>-1</sup>
U-236	3.3×10 <sup>-1</sup>	3.4×10 <sup>-2</sup>	1.7	1.4×10 <sup>-1</sup>	1.2×10 <sup>-2</sup>	3.7×10 <sup>-2</sup>	5.9×10 <sup>-2</sup>	2.1×10 <sup>-2</sup>	1.7×10 <sup>-2</sup>	$6.0 \times 10^{-2}$	4.1×10 <sup>-3</sup>	$8.1 \times 10^{-4}$	1.3×10 <sup>-2</sup>	2.5	$4.4 \times 10^{-1}$
U-238	1.6	1.5×10 <sup>-4</sup>	1.4×10 <sup>-2</sup>	1.3×10 <sup>-1</sup>	$3.4 \times 10^{-2}$	$8.9 \times 10^{-4}$	1.6×10 <sup>-2</sup>	$5.4 \times 10^{-5}$	7.1×10 <sup>-5</sup>	2.7×10 <sup>-4</sup>	2.7×10 <sup>-3</sup>	1.3×10 <sup>-5</sup>	$5.8 \times 10^{-3}$	1.2×10 <sup>-3</sup>	$2.4 \times 10^{-2}$
Zr-93	1.0	1.5×10 <sup>-1</sup>	$6.7 \times 10^{-3}$	9.1×10 <sup>-1</sup>	5.0×10 <sup>-2</sup>	1.0×10 <sup>-1</sup>	2.1×10 <sup>-1</sup>	1.1	$6.4 \times 10^{-2}$	2.7×10 <sup>-1</sup>	1.7×10 <sup>-2</sup>	5.7×10 <sup>-1</sup>	7.8×10 <sup>-2</sup>	$1.1 \times 10^{1}$	1.9

a. Source: DIRS 148240-Dirkmaat (1998, all); DIRS 155857-McKenzie (2001, Attachment B, p. 9). Values are rounded to two significant figures.

 $b. \quad INEEL = Idaho \ National \ Engineering \ and \ Environmental \ Laboratory; \ SRS = Savannah \ River \ Site.$ 

c. Categories 1-13 and 16 decayed to 2030. Category 15 cooled for 5 years.

d. Includes 334 canisters from Fort St. Vrain.

e. -- = not found in appreciable quantities.

f. Amount of cobalt-60 as crud is 5.8 curies.

g. Half-lives are listed in Table A-11, with the exception of Cf-252 = 2.65 years, Cm-242 =  $1.6 \times 10^7$  years, Cm-248 =  $3.4 \times 10^5$  years, Rh-102 = 2.9 years, Th-229 =  $7.9 \times 10^3$  years, and Th-232 =  $1.4 \times 10^{10}$  years.

- Category 2: Uranium-zirconium. The fuel in this category consists primarily of a uranium-(91 percent) zirconium alloy. The Heavy Water Components Test Reactor fuel is the representative fuel because it is the largest part of the inventory. This fuel is approximately 85-percent enriched in uranium-235 and is clad with a zirconium alloy.
- Category 3: Uranium molybdenum. The fuel in this category consists of uranium- (10 percent)-molybdenum alloy and 25-percent enriched in uranium-235, and is clad with a zirconium alloy. Fermi driver core 1 and 2 are the only fuels in the category. The fuel is currently in an aluminum container. The proposed disposition would include the aluminum container.
- Category 4: Uranium oxide, intact. The fuel in this category consists of uranium oxide that has been formed into pellets or plates and clad with a corrosion-resistant material. Commercial fuel is the representative fuel for this category because it is a large part of the inventory. The fuel is made of uranium oxide, some of which is highly enriched in uranium-235 and some of which is low enriched in uranium-235. The fuel elements are clad with a zirconium alloy.
- Category 5: Uranium oxide, failed/declad/aluminum clad. The fuel in this category is chemically similar to the fuels in Category 4, except accident or destructive examination has disrupted it. The failed fuel from Three Mile Island Reactor 2 represents this category because it comprises 96 percent of the total MTHM of the category. The Three Mile Island Reactor 2 fuel is melted uranium oxide. The accident greatly disrupted the cladding. Other fuel in this category is declad or has a large amount of cladding damage. Approximately 4 percent consists of intact aluminum clad fuel included in this category because the aluminum cladding is less corrosion resistant than Category 4 cladding material.
- Category 6: Uranium-aluminide. This category consists of fuel with a uranium-aluminum compound dispersed in a continuous aluminum metal phase. The fuel is clad with an aluminum alloy. The uranium-235 enrichment varies from 10 to 93 percent.
- Category 7: Uranium-silicide. The fuel in this category is a uranium-silicide compound dispersed in a continuous aluminum metal phase. The fuel is clad with an aluminum alloy. The uranium-235 enrichment varies from 8 to 93 percent, but most are less than 20 percent.
- Category 8: Thorium/uranium carbide, high-integrity. This category consists of fuels with thorium carbide or uranium carbide formed into particles with a high-integrity coating. Fort St. Vrain Reactor fuel represents the category because it makes up 95 percent of the mass of the category. This fuel is uranium carbide and thorium carbide formed into particles and coated with layers of pyrolytic carbon and silicon carbide. The particles are bonded in a carbonaceous matrix material and emplaced in a graphite block. The fuel was made with uranium enriched to 93 percent in uranium-235. The thorium was used to generate fissile uranium-233 during irradiation. Some fuel does not have a silicon carbide coating, but its effect on the category is very small. Less than 1 percent of the fuel particles are breached.
- Category 9: Thorium/uranium carbide, low-integrity. This category consists of fuels with uranium carbide or thorium carbide made into particles with a coating of an earlier design than that described for Category 8. Peach Bottom Unit 1, Core 1 is the only fuel in this category. This fuel is chemically similar to Category 8 fuel except 60 percent of the particle coating is breached. Peach Bottom Unit 1, Core 2 is included in Category 8 because its fuel particles are basically intact and are more rugged than the Peach Bottom Unit 1, Core 1 particles.

- Category 10: Plutonium/uranium carbide, nongraphite. This category consists of fuel that contains uranium carbide. Much of it also contains plutonium carbide. Fast Flux Test Facility carbide assemblies represent this category because they make up 70 percent of the category and contain both uranium and plutonium. The Fast Flux Test Facility carbide fuel was constructed from uncoated uranium and plutonium carbide spheres that were loaded directly into the fuel pins, or pressed into pellets that were loaded into the pins. The pins are clad with stainless steel.
- Category 11: Mixed oxide. This category consists of fuels constructed of both uranium oxide and plutonium oxide. The Fast Flux Test Facility mixed-oxide test assembly is the representative fuel because it comprises more than 80 percent of the category. The fuels are a combination of uranium oxide and plutonium oxide pressed into pellets and clad with stainless steel or a zirconium alloy. The uranium-235 enrichment is low, but the fissile contribution of the plutonium raises the effective enrichment to 15 percent.
- Category 12: Uranium/thorium oxide. This category consists of fuels constructed of uranium oxide and thorium oxide. Shippingport light-water breeder reactor fuel is the representative fuel because it comprises more than 75 percent of the inventory. The Shippingport light-water breeder reactor fuel is made of uranium-233, and the irradiation of the thorium produces more uranium-233. The mixture is pressed into pellets and clad with a zirconium alloy.
- Category 13: Uranium-zirconium hydride. This category consists of fuels made of uranium-zirconium hydride. Training Research Isotopes-General Atomic fuels comprise more than 90 percent of the mass of this category. The fuel is made of uranium-zirconium hydride formed into rods and clad primarily with stainless steel or aluminum. The uranium is enriched as high as 90 percent in uranium-235, but most is less than 20 percent enriched.
- Category 14: Sodium-bonded. For purposes of analysis in this EIS, it is assumed that all Category 14 fuels would be treated during the proposed electrometallurgical treatment that would result in high-level radioactive waste. The chemical composition of the resulting high-level radioactive waste is described in Section A.2.3.5.3. Category 14 is included here for completeness.
- Category 15: Naval fuel. Naval nuclear fuel is highly robust and designed to operate in a high-temperature, high-pressure environment for many years. This fuel is highly enriched (93 to 97 percent) in uranium-235. In addition, to ensure that the design will be capable of withstanding battle shock loads, the naval fuel material is surrounded by large amounts of zirconium alloy (DIRS 124679-Beckett 1998, Attachment 2).

DOE plans to emplace approximately 300 canisters of naval spent nuclear fuel in the Yucca Mountain Repository. There are several different designs for naval nuclear fuel, but all designs employ similar materials and mechanical arrangements. The total weight of typical fuel assemblies in a canister would be 11,000 to 13,000 kilograms (24,000 to 29,000 pounds). Of this total, less than 500 kilograms (1,100 pounds) would be uranium. Approximately 1,000 to 2,000 kilograms (2,200 to 4,400 pounds) of the total weight of these fuel assemblies is from hafnium in the poison devices (primarily control rods) permanently affixed to the fuel assemblies (DIRS 124679-Beckett 1998, Attachment 2).

There would be approximately 9,000 to 12,000 kilograms (20,000 to 26,500 pounds) of zirconium alloy in the fuel structure in the typical canister. The typical chemical composition of zirconium alloy is approximately 98 percent zirconium, 1.5 percent tin, 0.2 percent iron, and 0.1 percent chromium (DIRS 124679-Beckett 1998, Attachment 2).

The small remainder of the fuel mass in a typical canister of naval spent nuclear fuel [less than 500 kilograms (1,100 pounds)] would consist of small amounts of such metals and nonmetals as fission products and oxides (DIRS 124679-Beckett 1998, Attachment 2).

• Category 16: Miscellaneous. This category consists of the fuels that do not fit into the previous 15 categories. The largest amount of this fuel, as measured in MTHM, is uranium metal or alloy. The other two primary contributors are uranium alloy and uranium-thorium alloy. These three fuel types make up more than 80 percent of the MTHM in the category. It is conservative to treat the total category as uranium metal. Other chemical compounds included in this category include uranium oxide, uranium nitride, uranium alloys, plutonium oxide, plutonium nitride, plutonium alloys, and thorium oxide.

Table A-22 lists the primary materials of construction and chemical composition for each category.

# A.2.2.5.4 Thermal Output

Table A-23 lists the maximum heat generation per handling unit for each spent nuclear fuel category (DIRS 104354-Dirkmaat 1997, Attachment, pp. 74 to 77; DIRS 104377-Dirkmaat 1998, all). The category 15 (naval fuel) thermal data used the best estimate radionuclide content from DIRS 104354-Dirkmaat (1997, Attachment, pp. 74 to 77) at a minimum cooling time of 5 years.

# A.2.2.5.5 Quantity of Spent Nuclear Fuel Per Canister

Table A-24 lists the projected number of canisters required for each site and category. The amount of fuel per canister would vary widely among categories and would depend on a variety of parameters. The average mass of naval spent nuclear fuel in a short naval canister would be approximately 13 metric tons (14 tons) with an associated volume of 2.7 cubic meters (95 cubic feet). Naval spent nuclear fuel in a long naval canister would have an average mass of approximately 18 metric tons (20 tons) and a volume of 3.5 cubic meters (124 cubic feet) (DIRS 104354-Dirkmaat 1997, Attachment, p. 108).

### A.2.2.5.6 Spent Nuclear Fuel Canister Parameters

The Idaho National Engineering and Environmental Laboratory would use a combination of 46- and 61-centimeter (18- and 24-inch)-diameter stainless-steel canisters for spent nuclear fuel disposition. The Savannah River Site would use 18-inch canisters, and Hanford would use 64-centimeter (25.3-inch) multicanister overpacks and 18-inch canisters. Table A-24 lists the specific number of canisters per site. Detailed canister design specifications for the standard 18- and 24-inch canisters are contained in DIRS 137713-DOE (1998, all). Specifications for the Hanford multicanister overpacks are in DIRS 103499-Parsons (1999, all).

There are two conceptual canister designs for naval fuel: one with a length of 539 centimeters (212 inches) and one with a length of 475 centimeters (187 inches). Both canisters would have a maximum diameter of 169 centimeters (67 inches) (DIRS 104354-Dirkmaat 1997, Attachment, pp. 86 to 88). Table A-25 summarizes the preliminary design information.

For both designs, the shield plug, shear ring, and outer seal plate would be welded to the canister shell after the fuel baskets were loaded in the canister. The shield plug, shear ring, and welds, along with the canister shell and bottom plug, would form the containment boundary for the disposable container. The shell, inner cover, and outer cover material for the two canisters would be low-carbon austenitic stainless steel or stabilized austenitic stainless steel. Shield plug material for either canister would be stainless steel (DIRS 104354-Dirkmaat 1997, Attachment, pp. 86 to 88).

								Category							
Fuel	1	2	3	4	5	6	7	8	9	10	11	12	13	15	16
Components															
Uranium	2,120,000	40	3,800	98,000	87,000	8,800	12,000	1,300	210	140	9,900	810	2,000	65,000	8,500
Aluminum	1,700	(c)				18,000	4,200								
Molybdenum			380										9		
Zirconium	140	440		7,500									23,000		
Thorium								27,000	1,500			48,000			2,200
Plutonium										16	2,400				8
Silicon	260						880								
Silicon carbide								53,000							
Carbon	1,200			30				220,000	53,000				1,700		
Cladding and structu	ıre														
Aluminum	100		640		18,000	64,000	52,000						11,000		500
Stainless steel				11,000	3,000	- 1,	,		8,000	320	2,400	31,000	17,000		20,000
Zirconium alloy	160,000	70	280	64,000	58,000				-,		500	12,000	100	3,600,000	100
Inconel				1,000	1,700										
Container															
Stainless steel	2,640,000	5,600	50,000	165,000	750,000	900,000	270,000	500,000	42,000	3,500	260,000	50,000	70,000	9,900,000	31,000
Aluminum	2,0.0,000	2,000	660	100,000	10,000	,00,000	270,000	200,000	.2,000	2,200	200,000	20,000	70,000	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	51,000
Other															
Concrete					30,000 <sup>d</sup>										
Boron					500	1,000	300		29						
Silver					1,100	1,000	300		29						
Cadmium					34										
Indium					280										
					280				430						
Magnesium Nickel	210								430						
Rhodium	210								30						
Ruthenium									30						
Samarium									30				67		
Gadolinium					530	950	23						07		
Hafnium					550	950	23							600,000	
11411114111														000,000	

a. Source: DIRS 148240-Dirkmaat (1998, all); DIRS 104377-Dirkmaat (1998, p. 008/016, 009/016); values are rounded to two significant figures.

b. To convert kilograms to pounds, multiply by 2.2046.

c. Blanks indicate none or less than reportable quantities.

d. Low density converters were added to canisters of Three Mile Island Unit 2 fuel and would remain when shipped to the repository.

**Table A-23.** Maximum heat generation for DOE spent nuclear fuel (watts per handling unit).<sup>a,b</sup>

		Maximum heat
	Category and fuel type	generation
1.	Uranium metal	18
2.	Uranium zirconium	90
3.	Uranium molybdenum	4
4.	Intact uranium oxide	1,000
5.	Failed/declad/aluminum clad uranium oxide	800
6.	Uranium aluminide	480
7.	Uranium silicide	1,400
8.	High-integrity thorium/uranium carbide	250
9.	Low-integrity thorium/uranium carbide	37
10.	Nongraphite plutonium/uranium carbide	1,800
11.	Mixed oxide	1,800
12.	Thorium/uranium oxide	600
13.	Uranium zirconium hydride	100
14.	Sodium-bonded	N/A <sup>c</sup>
15.	Naval fuel	4,250
16.	Miscellaneous	1,000

a. Sources: DIRS 104354-Dirkmaat (1997, Attachment, pp. 74 to 77); DIRS 104377-Dirkmaat (1998, Table A.2.2-3); DIRS 156933-Fillmore (2001, all).

Table A-24. Required number of canisters for disposal of DOE spent nuclear fuel. a,b

	Har	nford	INE	EEL	SRS	Na	val
Category	18-inch	25.3-inch	18-inch	24-inch	18-inch	Short	Long
1		440	6		9		
2			8				
3			70				
4	14	20	179	16			
5	1		406		425		
6					750		
7					225		
8			503°				
9			60				
10	2		3				
11	324		43				
12			24	47			
13	3		97				
14 <sup>d</sup>							
15						200	100
16	5		39		2		
<b>Totals</b>	349	460	1,438	<i>63</i>	1,411	200	100

a. Sources: DIRS 104356-Dirkmaat (1997, Attachment, p. 2); Dirkmaat (DIRS 148240-1998, all).

b. Handling unit is a canister.

c. N/A = not applicable. Assumed to be treated and therefore part of high-level radioactive waste inventory (see Section A.2.2.1).

b. INEEL = Idaho National Engineering and Environmental Laboratory; SRS = Savannah River Site.

c. Includes 334 canisters from Fort St. Vrain.

d. Assumed to be treated and therefore part of high-level radioactive waste inventory (see Section A.2.2.1).

Table A-25. Preliminary naval canister design parameters.<sup>a</sup>

Parameter	Short canister	Long canister
Maximum outside diameter (centimeters) <sup>b,c</sup>	169	169
Maximum outer length (centimeters)	475	539
Minimum loaded weight (metric tons) <sup>d</sup>	27	27
Maximum loaded weight (metric tons)	45	45

- a. Source: DIRS 104354-Dirkmaat (1997, Attachment, pp. 86 to 88).
- b. To convert centimeters to inches, multiply by 0.3937.
- c. Right circular cylinder.
- d. To convert metric tons to tons, multiply by 1.1023.

#### A.2.3 HIGH-LEVEL RADIOACTIVE WASTE

High-level radioactive waste is the highly radioactive material resulting from the reprocessing of spent nuclear fuel. DOE stores high-level radioactive waste at the Hanford Site, the Savannah River Site, and the Idaho National Engineering and Environmental Laboratory. Between 1966 and 1972, commercial chemical reprocessing operations at the Nuclear Fuel Services plant near West Valley, New York, generated a small amount of high-level radioactive waste at a site presently owned by the New York State Energy Research and Development Authority. These operations ceased after 1972. In 1980, Congress passed the West Valley Demonstration Project Act, which authorizes DOE to conduct, with the Research and Development Authority, a demonstration of solidification of high-level radioactive waste for disposal and the decontamination and decommissioning of demonstration facilities (DIRS 102588-DOE 1992, Chapter 3). This section addresses defense high-level radioactive waste generated at the DOE sites (Hanford Site, Idaho National Engineering and Environmental Laboratory, and Savannah River Site) and commercial high-level radioactive waste generated at the West Valley Demonstration Project.

## A.2.3.1 Background

In 1985, DOE published a report in response to Section 8 of the Nuclear Waste Policy Act (of 1982) that required the Secretary of Energy to recommend to the President whether defense high-level radioactive waste should be disposed of in a geologic repository along with commercial spent nuclear fuel. That report, *An Evaluation of Commercial Repository Capacity for the Disposal of Defense High-Level Waste* (DIRS 103492-DOE 1985, all), provided the basis, in part, for the President's determination that defense high-level radioactive waste should be disposed of in a geologic repository. Given that determination, DOE decided to allocate 10 percent of the capacity of the first repository for the disposal of DOE spent nuclear fuel (2,333 MTHM) and high-level radioactive waste (4,667 MTHM) (DIRS 104384-Dreyfuss 1995, all; DIRS 104398-Lytle 1995, all).

Calculating the MTHM quantity for spent nuclear fuel is straightforward. It is determined by the actual heavy metal content of the spent fuel. However, an equivalence method for determining the MTHM in defense high-level radioactive waste is necessary because almost all of its heavy metal has been removed. A number of alternative methods for determining MTHM equivalence for high-level radioactive waste have been considered over the years. Four of those methods are described in the following paragraphs.

Historical Method. Table 1-1 of DIRS 103492-DOE (1985) provided a method to estimate the MTHM equivalence for high-level radioactive waste based on comparing the radioactive (curie) equivalence of commercial high-level radioactive waste and defense high-level radioactive waste. The method relies on the relative curie content of a hypothetical (in the early 1980s) canister of defense high-level radioactive waste from the Savannah River, Hanford, or Idaho site, and a hypothetical canister of vitrified waste from reprocessing of high-burnup commercial spent nuclear fuel. Based on commercial high-level radioactive waste containing 2.3 MTHM per canister (heavy metal has not been removed from commercial waste) and defense high-level radioactive waste estimated to contain approximately 22 percent of the

radioactivity of a canister of commercial high-level radioactive waste, defense high-level radioactive waste was estimated to contain the equivalent of 0.5 MTHM per canister. Since 1985, DOE has used this 0.5 MTHM equivalence per canister of defense high-level radioactive waste in its consideration of the potential impacts of the disposal of defense high-level radioactive waste, including the analysis presented in this EIS. With this method, less than 50 percent of the total inventory of high-level radioactive waste could be disposed of in the repository within the 4,667 MTHM allocation for high-level radioactive waste. There has been no determination of which waste would be shipped to the repository, or the order of shipments.

Spent Nuclear Fuel Reprocessed Method. Another method of determining MTHM equivalence, based on the quantity of spent nuclear fuel reprocessed, would be to consider the MTHM in the high-level radioactive waste to be the same as the MTHM in the spent nuclear fuel before it was reprocessed. Using this method, less than 5 percent of the total inventory of high-level radioactive waste could be disposed of in the repository within the 4,667 MTHM allocation for high-level radioactive waste.

Total Radioactivity Method. Another method, the total radioactivity method, would establish equivalence based on a comparison of radioactivity inventory (curies) of defense high-level radioactive waste to that of a standard MTHM of commercial spent nuclear fuel. For this equivalence method the standard spent nuclear fuel characteristics are based on pressurized-water reactor fuel with uranium-235 enrichment of 3.11 percent and 39.65 gigawatt-days per MTHM burnup. Using this method, 100 percent of the total inventory of high-level radioactive waste inventory could be disposed of in the repository within the 4,667 MTHM allocation for high-level radioactive waste.

Radiotoxicity Method. Yet another method, the radiotoxicity method, uses a comparison of the relative radiotoxicity of defense high-level radioactive waste to that of a standard MTHM of commercial spent nuclear fuel, and is thus considered an extension of the total radioactivity method. Radiotoxicity compares the inventory of specific radionuclides to a regulatory release limit for that radionuclide, and uses these relationships to develop an overall radiotoxicity index. For this equivalence, the standard spent nuclear fuel characteristics are based on pressurized-water reactor fuel with uranium-235 enrichment of 3.11 percent, 39.65 gigawatt-days per MTHM burnup. Using this method, 100 percent of the total inventory of high-level radioactive waste could be disposed of in the repository within the 4,667 MTHM allocation for high-level radioactive waste.

A recent report (DIRS 103495-Knecht et al. 1999, all) describes four equivalence calculation methods and notes that, under the Total Radioactivity Method or the Radiotoxicity Method, all DOE high-level radioactive waste could be disposed of under the Proposed Action. Using different equivalence methods would shift the proportion of high-level radioactive waste that could be disposed of between the Proposed Action and Inventory Module 1 analyzed in Chapter 8, but would not change the cumulative impacts analyzed in this EIS. Regardless of the equivalence method used, the EIS analyzes the impacts from disposal of the entire inventory of high-level radioactive waste in inventory Module 1.

#### A.2.3.2 Sources

# A.2.3.2.1 Hanford Site

The Hanford high-level radioactive waste materials discussed in this EIS include tank waste, strontium capsules, and cesium capsules (DIRS 104406-Picha 1997, Table RL-1). DOE has not declared other miscellaneous materials or waste at Hanford, either existing or forecasted, to be candidate high-level radioactive waste streams. Before shipment to the repository, DOE would vitrify the high-level radioactive waste into a borosilicate glass matrix and pour it into stainless-steel canisters.

# A.2.3.2.2 Idaho National Engineering and Environmental Laboratory

The Idaho National Engineering and Environmental Laboratory has proposed three different high-level radioactive waste stream matrices for disposal at the proposed Yucca Mountain Repository—glass, ceramic, and metal. The glass matrix waste stream would come from the Idaho Nuclear Technology and Engineering Center and would consist of wastes generated from the treatment of irradiated nuclear fuels. The ongoing Argonne National Laboratory-West electrometallurgical treatment of DOE sodium-bonded fuels will generate both ceramic and metallic high-level radioactive waste matrices. DOE is developing the *Idaho High-Level Waste and Facilities Disposition Final Environmental Impact Statement* (see DIRS 155100-DOE 1999, all), to support decisions on managing the high-level radioactive waste at the Idaho Nuclear Technology and Engineering Center.

### A.2.3.2.3 Savannah River Site

Savannah River Site high-level radioactive waste consists of wastes generated from the treatment of irradiated nuclear fuels. These wastes include various chemicals, radionuclides, and fission products that DOE maintains in liquid, sludge, and saltcake forms. The Defense Waste Processing Facility at the Savannah River Site mixes the high-level radioactive waste with glass-forming materials, converts it to a durable borosilicate glass waste form, pours it into stainless-steel canisters, and seals the canisters with welded closure plugs (DIRS 104406-Picha 1997, Attachment 4, p. 2).

Another source of high-level radioactive waste at the Savannah River Site is the immobilized plutonium addressed in Section A.2.4.

# A.2.3.2.4 West Valley Demonstration Project

The West Valley Demonstration Project is responsible for solidifying high-level radioactive waste that remains from the commercial spent nuclear fuel reprocessing plant operated by Nuclear Fuel Services. The Project mixes the high-level radioactive waste with glass-forming materials, converts it to a durable borosilicate glass waste form, pours it into stainless-steel canisters, and seals the canisters with welded closure plugs.

### A.2.3.3 Present Status

### A.2.3.3.1 Hanford Site

The Hanford Site stores high-level radioactive waste in underground carbon-steel tanks. This analysis assumed that before vitrification, strontium and cesium capsules currently stored in water basins at Hanford would be blended with the liquid high-level radioactive waste. To date, Hanford has immobilized no high-level radioactive waste. Before shipping waste to a repository, DOE would vitrify it into an acceptable glass form. DOE has scheduled vitrification to begin in 2007 with an estimated completion in 2028.

### A.2.3.3.2 Idaho National Engineering and Environmental Laboratory

Most of the high-level radioactive waste at the Idaho Nuclear Technology and Engineering Center (formerly the Idaho Chemical Processing Plant) is in calcined solids (calcine) stored at the Idaho National Engineering and Environmental Laboratory. The calcine, an interim waste form, is in stainless-steel bins in concrete vaults. Before shipment to a repository, DOE proposes to immobilize the high-level radioactive waste in a vitrified (glass) waste form. The Idaho Nuclear Technology and Engineering Center proposes to implement its vitrification program in 2020 and complete it in 2035 (DIRS 103497-INEEL 1998, pp. A-39 to A-42).

As discussed in Section A.2.2.1, Argonne National Laboratory-West began electrometallurgical treatment of EBR-II reactor fuel in 2000. The ceramic and metallic waste forms being produced will be stored onsite.

#### A.2.3.3.3 Savannah River Site

DOE stores high-level radioactive waste in underground tanks at the Savannah River Site. High-level radioactive waste that has been converted to a borosilicate glass form and DOE projects completion of the vitrification of the stored high-level radioactive waste by 2027 (DIRS 157008-DOE 2001, all).

## A.2.3.3.4 West Valley Demonstration Project

High-level radioactive waste is stored in underground tanks at the West Valley site. High-level radioactive waste that has been converted into a borosilicate glass waste form is stored onsite. West Valley plans to complete its vitrification program by the Fall of 2002 (DIRS 102588-DOE 1992, Chapter 3).

#### A.2.3.4 Final Waste Form

The final waste form for high-level radioactive waste from the Hanford Site, Savannah River Site, Idaho Nuclear Technology and Engineering Center, and West Valley Demonstration Project would be a vitrified glass matrix in a stainless-steel canister.

The waste forms from Argonne National Laboratory-West will be ceramic and metallic waste matrices and will be in stainless-steel canisters similar to those used for Savannah River Site and Idaho Nuclear Technology and Engineering Center glass wastes.

## A.2.3.5 Waste Characteristics

#### A.2.3.5.1 Mass and Volume

Hanford Site. The estimated volume of borosilicate glass generated by high-level radioactive waste disposal actions at Hanford will be 15,700 cubic meters (554,000 cubic feet); the estimated mass of the glass is 44,000 metric tons (48,500 tons) (DIRS 104407-Picha 1998, Attachment 1). The volume calculation assumes that strontium and cesium compounds from capsules currently stored in water basins would be blended with tank wastes before vitrification with no increase in product volume. This volume of glass could require as many as 14,500 canisters, nominally 4.5 meters (15 feet) long with a 0.61-meter (2-foot) diameter (DIRS 104407-Picha 1998, Attachment 1).

*Idaho National Engineering and Environmental Laboratory.* Table A-26 lists the volumes, masses, densities, and estimated number of canisters for the three proposed waste streams.

Savannah River Site. Based on Revision 8 of the High-Level Waste System Plan (DIRS 101904-Davis and Wells 1997, all), the Savannah River Site would generate an estimated 5,978 canisters of high-level radioactive waste (DIRS 104406-Picha 1997, Attachment 1). The canisters have a nominal outside diameter of 0.61 meter (2 feet) and a nominal height of 3 meters (10 feet). They would contain a total of approximately 4,240 cubic meters (150,000 cubic feet) of glass. The estimated total mass of high-level radioactive waste for repository disposal would be 11,600 metric tons (12,800 tons) (DIRS 104406-Picha 1997, Attachment 1). DOE has addressed the additional high-level radioactive waste canisters that DOE

**Table A-26.** Physical characteristics of high-level radioactive waste at the Idaho National Engineering and Environmental Laboratory. <sup>a,b</sup>

Physical quantities	INTEC glass matrix	ANL-W ceramic matrix	ANL-W metal matrix
Volume (cubic meters) <sup>c</sup>	743	60.0	1.2
Mass (kilograms) <sup>d</sup>	1,860,000	144,000	9,000
Density (kilograms per cubic meter)	2,500	2,400	7,750
Number of canisters [range] <sup>e</sup>	1,190	96 [80 - 125]	6 [2 - 10]

- a. Sources: DIRS 104406-Picha (1997, Attachment 1, Table ID-2); DIRS 104389-Goff (1998, all); DIRS 104392-Goff (1998, all).
- b. INTEC = Idaho Nuclear Technology and Engineering Center; ANL-W = Argonne National Laboratory-West.
- c. To convert cubic meters to cubic yards, multiply by 1.3079.
- d. To convert kilograms to pounds, multiply by 2.2046.
- e. Canister would be nominally 3 meters (10 feet) by 0.61 meter (2 feet). Canisters would be filled to approximately 0.625 cubic meter (22 cubic feet).

would generate at the Savannah River Site as a result of immobilizing surplus plutonium (DIRS 118979-DOE 1999, p. 2-29). As discussed in DIRS 118979-DOE (1999, p. 2-29), 101 additional canisters would be required if the assumed one-third of the plutonium is immobilized. If the entire inventory of surplus plutonium was immobilized, 272 additional high-level radioactive waste canisters would be required.

West Valley Demonstration Project. The West Valley Demonstration Project will generate between 260 and 300 canisters of high-level radioactive waste. The canisters have a nominal outside diameter of 0.61 meter (2 feet) and a nominal height of 3 meters (10 feet) (DIRS 104406-Picha 1997, Attachment 1). They will contain approximately 200 cubic meters (7,060 cubic feet) of glass. The estimated total mass of this high-level radioactive waste will be between 540 and 630 metric tons (595 and 694 tons) (DIRS 104413-Picha 1998, p. 3).

*Summary.* Table A-27 summarizes the information in the previous paragraphs to provide the estimated total mass and volume projected to be disposed of at the repository.

**Table A-27.** High-level radioactive waste mass and volume summary.

Parameter	Total <sup>a,b</sup>
Mass	58,000 metric tons
Volume	21,000 cubic meters
Number of canisters	22,147 - 22,280°

- Sources: DIRS 104406-Picha (1997, Attachment 1); DIRS 104407-Picha (1998, Attachment 1).
- b. To convert metric tons to tons, multiply by 1.1023; to convert cubic meters to cubic yards, multiply by 1.3079.
- c. The number of canisters depends on the amount of surplus weapons-usable plutonium immobilized (see Section A.2.4.5.2.1).

### A.2.3.5.2 Amount and Nature of Radioactivity

The following paragraphs present radionuclide inventory information for the individual sites. They present the best available data at varying dates; however, in most cases, the data are conservative because the inventories are for dates earlier than the date of disposal, and additional radioactive decay would occur before disposal. Any differences due to varying amounts of radioactive decay are small.

Hanford Site. Table A-28 lists the estimated radionuclide inventory for Hanford high-level radioactive glass waste, including strontium-90 and cesium-137 currently stored in capsules (DIRS 104406-Picha 1997, Table RL-1). With the exception of hydrogen-3 and carbon-14, this table makes the conservative assumption that 100 percent of a radionuclide in Hanford's 177 tanks and existing capsules is vitrified.

**Table A-28.** Radionuclide distribution for Hanford Site high-level radioactive waste. a,b

		Curies per			Curies per
Radionuclide <sup>c</sup>	Total curies	canister	Radionuclide	Total curies	canister
Hydrogen-3	d		Thorium-229	1.8	$1.3 \times 10^{-4}$
Carbon-14	$9.6 \times 10^{-2}$	$6.6 \times 10^{-6}$	Thorium-230		
Chlorine-36			Thorium-232	2.1	$1.5 \times 10^{-4}$
Nickel-59	$9.3 \times 10^{2}$	$6.4 \times 10^{-2}$	Protactinium-231	$1.6 \times 10^{2}$	$1.1 \times 10^{-2}$
Nickel-63	$9.2 \times 10^4$	6.3	Uranium-232	$1.2 \times 10^{2}$	$8.5 \times 10^{-3}$
Cobalt-60	$1.2 \times 10^4$	$8.5 \times 10^{-1}$	Uranium-233	$4.8 \times 10^{2}$	$3.3 \times 10^{-2}$
Selenium-79	$7.7 \times 10^{2}$	$5.3 \times 10^{-2}$	Uranium-234	$3.5 \times 10^{2}$	$2.4 \times 10^{-2}$
Krypton-85			Uranium-235	$1.5 \times 10^{1}$	$1.0 \times 10^{-3}$
Strontium-90	$9.7 \times 10^{7}$	$6.7 \times 10^3$	Uranium-236	9.6	$6.6 \times 10^{-4}$
Niobium-93m	$2.7 \times 10^{3}$	$1.9 \times 10^{-1}$	Uranium-238	$3.2 \times 10^{2}$	$2.2 \times 10^{-2}$
Niobium-94			Neptunium-237	$1.4 \times 10^{2}$	$9.7 \times 10^{-3}$
Zirconium-93	$3.6 \times 10^{3}$	$2.5 \times 10^{1}$	Plutonium-238	$2.1 \times 10^{3}$	$1.9 \times 10^{-1}$
Technetium-99	$3.3 \times 10^4$	2.3	Plutonium-239	$4.7 \times 10^4$	2.7
Rhodium-101			Plutonium-240	$9.9 \times 10^{3}$	$6.2 \times 10^{-1}$
Rhodium-102			Plutonium-241	$2.3 \times 10^{5}$	$1.6 \times 10^{1}$
Ruthenium-106	$1.0 \times 10^{5}$	7.2	Plutonium-242	1.2	$8.0 \times 10^{-5}$
Palladium-107			Americium-241	$7.0 \times 10^4$	4.8
Tin-126	$1.2 \times 10^{3}$	$8.2 \times 10^{-2}$	Americium-242m		
Iodine-129	$3.2 \times 10^{1}$	$2.2 \times 10^{-3}$	Americium-243	9.3	$6.4 \times 10^{-4}$
Cesium-134	$8.9 \times 10^4$	6.1	Curium-242	$7.7 \times 10^{1}$	$5.3 \times 10^{-3}$
Cesium-135			Curium-243	$1.0 \times 10^{1}$	$6.9 \times 10^{-4}$
Cesium-137	$1.1 \times 10^{8}$	$7.7 \times 10^3$	Curium-244	$2.4 \times 10^{2}$	$1.7 \times 10^{-2}$
Samarium-151	$2.8 \times 10^{6}$	$1.9 \times 10^{2}$	Curium-245		
Lead-210			Curium-246		
Radium-226	$6.3 \times 10^{-2}$	$4.4 \times 10^{-6}$	Curium-247		
Radium-228	$7.7 \times 10^{1}$	$5.3 \times 10^{-3}$	Curium-248		
Actinium-227	$8.8 \times 10^{1}$	$6.0 \times 10^{-3}$	Californium-252		

a. Sources: DIRS 104406-Picha (1997, Table RL-1); DIRS 104407-Picha (1998, Attachment 1).

Consistent with Hanford modeling for the Integrated Data Base (DIRS 101815-DOE 1997, p. 2-24), pretreatment and vitrification would separate hydrogen-3 and carbon-14 from the high-level radioactive waste stream such that essentially 0.0 percent and 0.002 percent of each, respectively, would be present in the glass. A large portion of iodine-129 could also be separated, but the analysis assumed a conservative 50-percent retention (DIRS 104407-Picha 1998, Attachment 1). Table A-28 uses the estimated number of canisters (14,500) to develop the curies-per-canister value.

Idaho National Engineering and Environmental Laboratory. Table A-29 contains a baseline radionuclide distribution for the three Idaho National Engineering and Environmental Laboratory highlevel radioactive waste streams. For each waste stream, the total radionuclide inventory is provided, as is the worst-case value for curies per canister. For Idaho Nuclear Technology and Engineering Center glass, the calculated inventories are decayed to 2035. For Argonne National Laboratory-West waste matrices, the calculated inventories are decayed to 2000.

Savannah River Site. The Waste Qualification Report details the projected radionuclide distribution in the high-level radioactive waste from the Savannah River Site (DIRS 101908-Plodinec and Marra 1994, p. 10). Table A-30 lists the quantities of individual radionuclides decayed to 2015. The curie-percanister values were obtained by dividing the total radionuclide projection by the expected number of canisters (5,978).

b. Decayed to January 1, 1994.

c. Half-lives are listed in Tables A-11 and footnote g of Table A-21, with the exception of lead-210 = 23 years, rhodium-101 = 3.3 years, radium- $226 = 1.6 \times 10^3$  years, and radium 228 = 5.7 years.

d. -- = not found in appreciable quantities.

**Table A-29.** Radionuclide distribution for Idaho National Engineering and Environmental Laboratory high-level radioactive waste. a,b

	INTE	C glass	ANL-W c	eramic <sup>d</sup>	ANL	W metal <sup>d</sup>
Radionuclides <sup>c</sup>	Total curies for 2035	Curies per canister <sup>d</sup>	Total curies for 2000	Curies per canister <sup>e</sup>	Total curies for 2000	Curies per canister <sup>e</sup>
Hydrogen-3	$3.6 \times 10^{3}$	4.3	f			
Carbon-14	2.8×10 <sup>-2</sup>	8.3×10 <sup>-5</sup>			4.3	4.3
Chlorine-36	2.6×10	0.5×10			4.5	4.5
Cobalt-60	$3.2 \times 10^{1}$	3.6×10 <sup>-2</sup>			$3.2 \times 10^3$	$3.2 \times 10^3$
Nickel-59	3.2×10 				$1.1 \times 10^{1}$	$1.1 \times 10^{1}$
					$4.1 \times 10^{2}$	$3.9 \times 10^{2}$
Nickel-63					4.1×10	3.9×10 
Selenium-79						
Krypton-85	$7.0 \times 10^6$	$1.2 \times 10^4$	$7.1 \times 10^5$	$4.7 \times 10^4$		
Strontium-90	$4.7 \times 10^{2}$	1.4			$2.9 \times 10^{1}$	$2.9 \times 10^{1}$
Niobium-93 Niobium-94	$5.4 \times 10^{-3}$	1.4 1.6×10 <sup>-5</sup>			2.9×10	2.9×10
		1.6×10				
Zirconium-93	 2 410 <sup>3</sup>	9.9			1 2 102	1 2102
Technetium-99	$3.4 \times 10^{3}$	9.9			$1.3 \times 10^2$	$1.3 \times 10^2$
Rhodium-101	2 0 . 10-5	2 2 10-8				
Rhodium-102	$2.0 \times 10^{-5}$	$2.2 \times 10^{-8}$			104	 2 1 10 <sup>4</sup>
Ruthenium-106	1.0×10 <sup>-9</sup>	$8.7 \times 10^{-13}$			$2.1 \times 10^4$	$2.1 \times 10^4$
Palladium-107		 2 < 10-1				
Tin-126	$8.9 \times 10^{1}$	2.6×10 <sup>-1</sup>			2.8	2.1
Iodine-129	5.6	1.7×10 <sup>-2</sup>	$3.4 \times 10^{-1}$	$1.8 \times 10^{-2}$		
Cesium-134	$3.3 \times 10^{-2}$	3.6×10 <sup>-5</sup>	$7.9 \times 10^3$	$5.1 \times 10^2$		
Cesium-135	$1.6 \times 10^{2}$	2.5×10 <sup>-1</sup>	$1.6 \times 10^{1}$	8.8×10 <sup>-1</sup>		
Cesium-137	$6.0 \times 10^6$	$1.2 \times 10^4$	$8.5 \times 10^{5}$	$5.3 \times 10^4$		
Samarium-151						
Lead-210	2	5	5			
Radium-226	$9.7 \times 10^{-3}$	$7.2 \times 10^{-5}$	$3.0 \times 10^{-5}$	$2.1 \times 10^{-6}$		
Radium-228						
Actinium-227						
Thorium-229	1					
Thorium-230	$4.0 \times 10^{-1}$	$2.8 \times 10^{-3}$	$4.7 \times 10^{-3}$	$8.9 \times 10^{-4}$		
Thorium-232	$9.9 \times 10^{-8}$	$5.0 \times 10^{-10}$	$2.3 \times 10^{-9}$	1.3×10 <sup>-10</sup>		
Protactinium-231						
Uranium-232	$4.6 \times 10^{-3}$	$5.2 \times 10^{-6}$	$2.6 \times 10^{-3}$	$1.8 \times 10^{-4}$	1.2×10 <sup>-4</sup>	1.2×10 <sup>-4</sup>
Uranium-233	$1.3 \times 10^{-3}$	$6.1 \times 10^{-6}$	$2.0 \times 10^{-4}$	$1.4 \times 10^{-5}$	5.8×10 <sup>-5</sup>	5.8×10 <sup>-5</sup>
Uranium-234	$1.0 \times 10^{2}$	1.1×10 <sup>-1</sup>	2.8	$1.9 \times 10^{-1}$	$7.7 \times 10^{-1}$	$7.7 \times 10^{-1}$
Uranium-235	$5.9 \times 10^{-1}$	$6.6 \times 10^{-4}$	8.8×10 <sup>-2</sup>	$5.9 \times 10^{-3}$	$2.5 \times 10^{-2}$	$2.5 \times 10^{-2}$
Uranium-236	1.5	$1.7 \times 10^{-3}$	$6.3 \times 10^{-2}$	$4.2 \times 10^{-3}$	$1.8 \times 10^{-2}$	$1.8 \times 10^{-2}$
Uranium-238	$2.9 \times 10^{-2}$	$3.3\times10^{-5}$	$2.8 \times 10^{-1}$	$4.9 \times 10^{-3}$	$9.7 \times 10^{-2}$	$8.8 \times 10^{-2}$
Neptunium-237	6.3	$2.8 \times 10^{-2}$	1.3	$5.8 \times 10^{-2}$	$2.4 \times 10^{-5}$	2.3×10 <sup>-5</sup>
Plutonium-238	$9.0 \times 10^4$	$1.0 \times 10^{2}$	$3.6 \times 10^{2}$	$2.9 \times 10^{1}$	$6.6 \times 10^{-3}$	$6.6 \times 10^{-3}$
Plutonium-239	$1.8 \times 10^{3}$	2.0	$1.7 \times 10^4$	$8.1 \times 10^{2}$	$3.3 \times 10^{-1}$	$3.3\times10^{-1}$
Plutonium-240	$1.6 \times 10^{3}$	1.8	$1.5 \times 10^{3}$	$6.9 \times 10^{1}$	$2.9 \times 10^{-2}$	$2.9 \times 10^{-2}$
Plutonium-241	$1.9 \times 10^4$	$2.2 \times 10^{1}$	$1.1 \times 10^4$	$1.3 \times 10^{3}$	$1.9 \times 10^{-1}$	$1.9 \times 10^{-1}$
Plutonium-242	3.4	$3.8 \times 10^{-3}$	$1.2 \times 10^{-1}$	$2.3 \times 10^{-2}$	$2.0 \times 10^{-6}$	$2.0 \times 10^{-6}$
Americium-241	$1.3 \times 10^4$	$1.4 \times 10^{1}$	$1.6 \times 10^{3}$	$3.4 \times 10^{1}$	$3.1 \times 10^{-2}$	$2.1 \times 10^{-2}$
Americium-242/242m	$1.5 \times 10^{-2}$	$9.4 \times 10^{-5}$	$1.4 \times 10^{1}$	$2.1 \times 10^{-1}$	$2.7 \times 10^{-4}$	$2.1 \times 10^{-4}$
Americium-243	$1.4 \times 10^{-2}$	1.1×10 <sup>-4</sup>	2.8×10 <sup>-1</sup>	1.9×10 <sup>-2</sup>	4.8×10 <sup>-6</sup>	4.8×10 <sup>-6</sup>
Curium-242	1.2×10 <sup>-2</sup>	$7.7 \times 10^{-5}$	$1.2 \times 10^{1}$	1.8×10 <sup>-1</sup>	2.3×10 <sup>-4</sup>	1.8×10 <sup>-4</sup>
Curium-243	4.7×10 <sup>-4</sup>	$3.4 \times 10^{-6}$	1.6×10 <sup>-1</sup>	3.1×10 <sup>-3</sup>	$3.0 \times 10^{-6}$	2.1×10 <sup>-6</sup>
Curium-244	1.0×10 <sup>-2</sup>	$7.7 \times 10^{-5}$	1.9	1.3×10 <sup>-1</sup>	3.1×10 <sup>-5</sup>	3.1×10 <sup>-5</sup>
Curium-245	3.7×10 <sup>-6</sup>	2.8×10 <sup>-8</sup>	6.8×10 <sup>-5</sup>	4.7×10 <sup>-6</sup>	1.1×10 <sup>-9</sup>	1.1×10 <sup>-9</sup>
Curium-246	8.7×10 <sup>-8</sup>	$6.6 \times 10^{-10}$	$4.2 \times 10^{-7}$	2.9×10 <sup>-8</sup>	$7.1 \times 10^{-12}$	$7.1 \times 10^{-12}$
Curium-247	$3.1 \times 10^{-14}$	$2.4 \times 10^{-16}$	$2.4 \times 10^{-13}$	1.6×10 <sup>-14</sup>	$4.0 \times 10^{-18}$	$4.0 \times 10^{-18}$
Curium-248	9.4×10 <sup>-15</sup>	$7.2 \times 10^{-17}$	$2.6 \times 10^{-14}$	1.8×10 <sup>-15</sup>	$4.4 \times 10^{-19}$	$4.4 \times 10^{-19}$
Californium-252			$6.5 \times 10^{-19}$	1.6×10 <sup>-19</sup>		

a. Sources: DIRS 104406-Picha (1997, Table ID-2); DIRS 104389-Goff (1998, all).

b. INTEC = Idaho Nuclear Technology and Engineering Center; ANL-W = Argonne National Laboratory-West.

c. Half-lives are listed in Tables A-11, footnote g of Table A-21, and footnote c of Table A-28.

d. Matrices based on treating all sodium-bonded fuels. Curie values based on calculated data from stored material.

e. Curie per canister values were provided as worst case rather than a homogenous mixture.

<sup>. -- =</sup> not found in appreciable quantities.

**Table A-30.** Radionuclide distribution for Savannah River Site high-level radioactive waste (2015).<sup>a</sup>

Radionuclideb         (curies)         canister         Radionuclide         (curies)         Curies per canister           Hydrogen-3        c          Thorium-229             Carbon-14           Thorium-230         2.4×10 <sup>-2</sup> 4.0×10 <sup>-6</sup> Chlorine-36           Thorium-232             Nickel-59         1.1×10 <sup>2</sup> 1.8×10 <sup>-2</sup> Protactinium-231             Nickel-63         1.2×10 <sup>4</sup> 2.1         Uranium-232             Cobalt-60 <sup>c</sup> 4.5×10 <sup>1</sup> Uranium-233             Selenium-79         1.1×10 <sup>3</sup> 1.8×10 <sup>-1</sup> Uranium-234         1.6×10 <sup>2</sup> 2.7×10 <sup>-2</sup> Krypton-85           Uranium-235              Strontium-90         1.7×10 <sup>8</sup> 2.9×10 <sup>4</sup> Uranium-236              Niobium-93m         1.3×10 <sup>4</sup> 2.2         Uranium-238         5.0×10 <sup>1</sup> 8.3×10 <sup>-3</sup> Niobium-93         3.0×10 <sup>4</sup> 5.0         Plutonium-237 <t< th=""></t<>
Carbon-14 Thorium-230 $2.4 \times 10^{-2}$ $4.0 \times 10^{-6}$ Chlorine-36 Thorium-232 Nickel-59 $1.1 \times 10^2$ $1.8 \times 10^{-2}$ Protactinium-231 Nickel-63 $1.2 \times 10^4$ $2.1$ Uranium-232 Cobalt-60° $4.5 \times 10^1$ Uranium-233 Selenium-79 $1.1 \times 10^3$ $1.8 \times 10^{-1}$ Uranium-234 $1.6 \times 10^2$ $2.7 \times 10^{-2}$ Krypton-85 Uranium-235 Strontium-90 $1.7 \times 10^8$ $2.9 \times 10^4$ Uranium-236 Niobium-93m $1.3 \times 10^4$ $2.2$ Uranium-238 $5.0 \times 10^1$ $8.3 \times 10^{-3}$ Niobium-94 Neptunium-237 $4.1 \times 10^2$ $6.8 \times 10^{-2}$ Zirconium-93 $3.0 \times 10^4$ $5.0$ Plutonium-238 $3.0 \times 10^6$ $5.0 \times 10^2$ Technetium-99 $1.5 \times 10^4$ $2.5$ Plutonium-239 $3.7 \times 10^4$ $6.2$ Rhodium-101 Plutonium-240 $2.5 \times 10^4$ $4.1$ Rhodium-102 Plutonium-241 $3.3 \times 10^6$ $5.4 \times 10^2$ Ruthenium-106 <sup>d</sup> $2.4$ Plutonium-242 $3.5 \times 10^1$ $5.8 \times 10^{-3}$ Palladium-107 $7.3 \times 10^1$ $1.2 \times 10^{-2}$ Americium-241 $1.6 \times 10^5$ $2.6 \times 10^1$ Tin-126 $2.6 \times 10^3$ $4.3 \times 10^{-1}$ Americium-242m
Carbon-14           Thorium-230 $2.4 \times 10^{-2}$ $4.0 \times 10^{-6}$ Chlorine-36           Thorium-232             Nickel-59 $1.1 \times 10^2$ $1.8 \times 10^{-2}$ Protactinium-231             Nickel-63 $1.2 \times 10^4$ $2.1$ Uranium-232             Cobalt-60° $4.5 \times 10^1$ Uranium-233             Selenium-79 $1.1 \times 10^3$ $1.8 \times 10^{-1}$ Uranium-234 $1.6 \times 10^2$ $2.7 \times 10^{-2}$ Krypton-85           Uranium-235             Strontium-90 $1.7 \times 10^8$ $2.9 \times 10^4$ Uranium-236             Niobium-93m $1.3 \times 10^4$ $2.2$ Uranium-238 $5.0 \times 10^1$ $8.3 \times 10^{-3}$ Niobium-94           Neptunium-237 $4.1 \times 10^2$ $6.8 \times 10^{-2}$ Zirconium-93 $3.0 \times 10^4$ $5.0$ Plutonium-238 $3.0 \times 10^6$ $5.0 \times 10^2$ Technetium-99 $1.5 \times 10^4$ $2$
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Cesium-137 $1.5 \times 10^8$ $2.4 \times 10^4$ Curium-244 $4.9 \times 10^5$ $8.3 \times 10^1$
Samarium-151 $3.3 \times 10^6$ $5.5 \times 10^2$ Curium-245
Lead-210 Curium-246
Radium-226 Curium-247
Radium-228 Curium-248
Actinium-227 Californium-252

a. Sources: DIRS 101908-Plodinec and Marra (1994, p. 10); DIRS 104403-Pearson (1998, all).

West Valley Demonstration Project. DOE used the ORIGEN2 computer code to estimate the radionuclide inventory for the West Valley Demonstration Project, simulating each Nuclear Fuel Services irradiated fuel campaign. A detailed description of the development of these estimates is in the West Valley Demonstration Project Waste Qualification Report (DIRS 103500-WVNS n.d., WQR-1.2, Appendix 1, Rev. 1). Table A-31 lists the estimated activity by nuclide and provides the total curies, as well as the curies per canister, based on 260 canisters.

#### A.2.3.5.3 Chemical Composition

Hanford Site. The Integrated Data Base (DIRS 101815-DOE 1997, p. 2-29) provides the best available information for the proposed representative chemical composition of future high-level radioactive waste glass from Hanford. Table A-32 combines the percentages by weight of chemical constituents obtained from the Integrated Data Base with the estimated mass to present the expected chemical composition of the glass in terms of mass per chemical compound.

### Idaho National Engineering and Environmental Laboratory

Idaho Nuclear Technology and Engineering Center Glass Matrix. This waste stream is composed of three primary sources—zirconium calcine, aluminum calcine, and sodium-bearing waste.

b. Half-lives are listed in Tables A-11, footnote g of Table A-21, and footnote c of Table A-28.

c. -- = not found in appreciable quantities.

d. Total curie content not provided for these nuclides; curie per canister values provided for 10 years after production.

**Table A-31.** Radionuclide distribution for West Valley Demonstration Project high-level radioactive waste (2015).<sup>a</sup>

		Curies per			Curies per
Radionuclide <sup>b</sup>	Total curies	canister	Radionuclide	Total curies	canister
Hydrogen-3	$2.0 \times 10^{1}$	$7.8 \times 10^{-2}$	Thorium-229	2.3×10 <sup>-1</sup>	$8.9 \times 10^{-4}$
Carbon-14	$1.4 \times 10^{2}$	$5.3 \times 10^{-1}$	Thorium-230	$6.0 \times 10^{-2}$	$2.3 \times 10^{-4}$
Chlorine-36	°		Thorium-232	1.6	$6.3 \times 10^{-3}$
Nickel-59	$1.1 \times 10^{2}$	$4.1 \times 10^{-1}$	Protactinium-231	$1.5 \times 10^{1}$	$5.9 \times 10^{-2}$
Nickel-63	$7.1 \times 10^{3}$	$2.7 \times 10^{1}$	Uranium-232	5.9	$2.3 \times 10^{-2}$
Cobalt-60	$2.9 \times 10^{1}$	$1.1 \times 10^{-1}$	Uranium-233	9.5	$3.7 \times 10^{-2}$
Selenium-79	$6.0 \times 10^{1}$	$2.3 \times 10^{-1}$	Uranium-234	5.0	$1.9 \times 10^{-2}$
Krypton-85			Uranium-235	$1.0 \times 10^{-1}$	$3.9 \times 10^{-4}$
Strontium-90	$3.7 \times 10^{6}$	$1.4 \times 10^4$	Uranium-236	$3.0 \times 10^{-1}$	$1.1 \times 10^{-3}$
Niobium-93m	$2.5 \times 10^{2}$	$9.5 \times 10^{-1}$	Uranium-238	$8.5 \times 10^{-1}$	$3.3 \times 10^{-3}$
Niobium-94			Neptunium-237	$2.4 \times 10^{1}$	$9.2 \times 10^{-2}$
Zirconium-93	$2.7 \times 10^{2}$	1.1	Plutonium-238	$7.0 \times 10^{3}$	$2.7 \times 10^{1}$
Technetium-99	$1.7 \times 10^{3}$	6.5	Plutonium-239	$1.7 \times 10^{3}$	6.4
Rhodium-101			Plutonium-240	$1.2 \times 10^3$	4.7
Rhodium-102			Plutonium-241	$2.5 \times 10^4$	$9.5 \times 10^{1}$
Ruthenium-106	$5.0 \times 10^{-7}$	$1.9 \times 10^{-9}$	Plutonium-242	1.7	$6.4 \times 10^{-3}$
Palladium-107	$1.1 \times 10^{1}$	$4.2 \times 10^{-2}$	Americium-241	$5.3 \times 10^4$	$2.0 \times 10^{2}$
Tin-126	$1.0 \times 10^{2}$	$4.0 \times 10^{-1}$	Americium-242m	$2.7 \times 10^{2}$	1.0
Iodine-129	$2.1 \times 10^{-1}$	$8.1 \times 10^{-4}$	Americium-243	$3.5 \times 10^{2}$	1.3
Cesium-134	1.2	$4.4 \times 10^{-3}$	Curium-242	$2.2 \times 10^{2}$	$8.4 \times 10^{-1}$
Cesium-135	$1.6 \times 10^{2}$	$6.2 \times 10^{-1}$	Curium-243	$7.3 \times 10^{1}$	$2.8 \times 10^{-1}$
Cesium-137	$4.1 \times 10^{6}$	$1.6 \times 10^4$	Curium-244	$2.9 \times 10^{3}$	$1.1 \times 10^{1}$
Samarium-151	$7.0 \times 10^4$	$2.7 \times 10^{2}$	Curium-245	$8.8 \times 10^{-1}$	$3.4 \times 10^{-3}$
Lead-210			Curium-246	$1.0 \times 10^{-1}$	$3.9 \times 10^{-4}$
Radium-226			Curium-247		
Radium-228	1.6	$6.3 \times 10^{-3}$	Curium-248		
Actinium-227	$1.2 \times 10^{1}$	$4.6 \times 10^{-2}$	Californium-252		

a. Source: DIRS 103500-WVNS (n.d., WQR-1.2, Appendix 1, Rev. 1).

**Table A-32.** Expected chemical composition of Hanford high-level radioactive waste glass (kilograms). a,b

Compound	Mass	Compound	Mass
Aluminum oxide	4,100,000	Sodium oxide	5,190,000
Boron oxide	3,090,000	Sodium sulfate	44,000
Bismuth trioxide	510,000	Nickel monoxide	480,000
Calcium oxide	370,000	Phosphorous pentaoxide	690,000
Ceric oxide	500,000	Lead monoxide	62,000
Chromic oxide	160,000	Silicon oxide	20,300,000
Ferric oxide	1,980,000	Strontium oxide	79,000
Potassium oxide	75,000	Thorium dioxide	4,400
Lanthanum oxide	48,000	Uranium oxide	2,940,000
Lithium oxide	880,000	Zirconium dioxide	1,630,000
Manganese dioxide	510,000	Other	75,000
Sodium fluoride	280,000	Total	44,000,000

a. Sources: DIRS 101815-DOE (1997, p. 2-29); DIRS 104407-Picha (1998, Attachment 1).

The distribution of these sources is 55 percent, 15 percent, and 30 percent, respectively (DIRS 104395-Heiser 1998, all). Table A-33 lists the chemical composition of the total waste stream.

b. Half-lives are listed in Tables A-11, A-21, and A-28.

c. -- = not found in appreciable quantities.

b. To convert kilograms to pounds, multiply by 2.2046.

**Table A-33.** Expected glass matrix chemical composition at Idaho Nuclear Technology and Engineering Center (kilograms). a,b

Compound or element	Mass	Compound or element	Mass
Aluminum oxide	130,000	Silicon oxide	1,020,000
Ammoniummolybdophosphate	26,000	Zirconium dioxide	18,000
Boron oxide	200,000	Arsenic	100
Calcium fluoride	140,000	Cadmium	42,000
Calcium oxide	4,100	Chromium	14,000
Ceric oxide	300	Mercury <sup>c</sup>	200
Ferric oxide	800	Nickel	1,400
Sodium oxide	250,000	Lead	1,800
Phosphorous pentaoxide	1,000	$Total^d$	1,860,000

- a. Sources: DIRS 104406-Picha (1997, Table ID-3); DIRS 104395-Heiser (1998, all).
- b. Masses are rounded to the nearest 100 kilograms; to convert kilograms to pounds, multiply by 2.2046.
- c. Assumes only 0.1 percent capture of original mercury in the feed materials.
- d. Trace amounts of antimony, beryllium, barium, selenium, silver, and thallium were also reported.

Argonne National Laboratory-West Ceramic and Metal Matrices. Electrometallurgical processing of DOE spent nuclear fuel containing thermal-bond sodium results in two high-level radioactive waste forms for repository disposal. The first form is a glass-bonded ceramic composite. It stabilizes the alkali, alkaline earth, lanthanide, halide, and transuranic materials in processed spent nuclear fuel. These elements are present as halides after fuel treatment. For disposal, these compounds are stabilized in a zeolite-based material (DIRS 104389-Goff 1998, all).

The chemical formula for zeolite-4A, the typical starting material, is  $Na_{12}[(AlO_2)_{12}(SiO_2)_{12}]$ . In the waste form, zeolite contains approximately 10 to 12 percent of the halide compounds by weight. The zeolite mixture typically is combined with 25-percent glass frit by weight, placed in a stainless-steel container, and processed into a solid monolith. The zeolite is converted to the mineral sodalite in the process (DIRS 104389-Goff 1998, all). Table A-34 lists the composition of the waste form.

**Table A-34.** Ceramic waste matrix chemical composition at Argonne National Laboratory-West (kilograms).<sup>a,b</sup>

Component	Mass	Component	Mass
Zeolite-4A	92,000	Potassium iodide	10
Silicon oxide	24,000	Cesium chloride	160
Boron oxide	6,800	Barium chloride	70
Aluminum oxide	2,500	Lanthium chloride	90
Sodium oxide	2,700	Ceric chloride	140
Potassium oxide	140	Praseodymium chloride	70
Lithium-potassium chloride	13,000	Neodymium chloride	240
Sodium chloride	980	Samarium chloride	40
Rubidium chloride	20	Yttrium chloride	60
Strontium chloride	70	$Total^c$	144,000

- a. Source: DIRS 104389-Goff (1998, all), DIRS 104392-Goff (1998, all).
- b. To convert kilograms to pounds, multiply by 2.2046.
- c. Includes trace amounts of potassium bromide and europium chloride.

The halide composition would depend on the fuel processed. The final bulk composition of the ceramic waste form by weight percentages would be 25 percent glass, 63 to 65 percent zeolite-4A, and 10 to 12 percent halide salts.

Table A-35 lists the estimated composition of the second high-level radioactive waste form, which is a metal matrix waste form. The table combines percentage weight distribution with the total expected mass of the metal waste form to achieve a distributed mass by element (DIRS 104389-Goff 1998, all).

**Table A-35.** Expected metal waste matrix chemical composition at Argonne National Laboratory-West (kilograms).<sup>a</sup>

Component	Mass
Iron	4,200
Chromium	1,500
Nickel	1,100
Manganese	180
Molybdenum	220
Silicon	90
Zirconium	1,400
NMFPs <sup>b</sup>	360
Others <sup>c</sup>	20
Total	9,000

- a. Source: DIRS 104389-Goff (1998, all); to convert kilograms to pounds, multiply by 2.2046.
- NMFPs = Noble metal fission products; includes silver, niobium, palladium, rhodium, ruthenium, antimony, tin, tantalum, technetium, and cobalt in small amounts.
- Others include trace amounts of carbon, phosphorus, and sulfur.

Savannah River Site. Fowler et al. (DIRS 101829-1995, p. 4) describes the chemical composition of the Defense Waste Processing Facility glass in detail. Table A-36 lists the distributed mass of the chemical constituents that comprise the current design-basis glass for the Savannah River Site. These values are based on a total mass of the glass of 11,600 metric tons (12,800 tons) (DIRS 104406-Picha 1997, Attachment 1).

West Valley Demonstration Project. The West Valley Demonstration Project will produce a single type of vitrified high-level radioactive waste. West Valley Nuclear Services Company provides a target composition for all chemical constituents in the high-level radioactive waste (DIRS 103500-WVNS n.d., WQR-1.1, Rev. 1, p. 7). Table A-37 lists the expected chemical composition based on this target composition and the upper range of the projected total glass mass, 630 metric tons (694 tons).

#### A.2.3.5.4 Thermal Output

Hanford Site. The estimated total thermal power from radioactive decay in the 14,500 reference canisters would be 1,190 kilowatts (as of January 1, 1994). This total heat load equates to an average power of 82 watts per canister. These values represent the hypothetical situation in which washed sludges from 177 tanks, cesium concentrates from the decontamination of low-level supernates, and strontium and cesium materials from capsules would be uniformly blended before vitrification. Realistically, uniform blending would not be likely. Current planning calls for merging all capsule materials with tank wastes from 2013 through 2016, which would create much hotter canisters during these years. In the extreme, the nonuniform blending of cesium concentrates and capsule materials into a relatively small volume of sludge waste could produce a few canisters with specific powers as high as 1,500 watts, which is the expected maximum for the nominally 4.5-meter (15-foot) Hanford canisters in the Civilian Radioactive Waste Management System Baseline (DIRS 104406-Picha 1997, Attachment 1, p. 2; DIRS 104476-Taylor 1997, all).

*Idaho National Engineering and Environmental Laboratory.* The Laboratory has three proposed high-level radioactive waste streams. Table A-38 lists the thermal output of these waste streams per waste canister.

**Table A-36.** Expected Savannah River Site high-level radioactive waste chemical composition (kilograms). <sup>a,b</sup>

Glass component	Mass	Glass component	Mass
Aluminum oxide	460,000	Sodium chloride	22,000
Barium sulfate	31,000	Neodymium	13,000
Calcium oxide	110,000	Nickel monoxide	100,000
Calcium sulfate	9,300	Neptunium	100
Cadmium	140	Promethium	210
Cerium	6,800	Praseodymium	3,300
Chromic oxide	14,000	Rubidium	120
Cesium oxide	14,000	Selenium	270
Copper oxide	51,000	Silicon oxide	5,800,000
Europium	200	Samarium	2,200
Ferric oxide	1,200,000	Tin	120
Potassium oxide	450,000	Tellurium	2,200
Lanthanum	3,500	Thorium dioxide	22,000
Lithium oxide	510,000	Titanium dioxide	100,000
Magnesium oxide	160,000	Uranium oxide	250,000
Manganese oxide	230,000	Zirconium	13,000
Molybdenum	14,000	Other <sup>c</sup>	58,000
Sodium oxide	1,000,000		
Sodium sulfate	12,000	Total	11,600,000

a. Sources: DIRS 101829-Fowler et al. (1995, p. 4); DIRS 104406-Picha (1997, Attachment 1).

**Table A-37.** Expected West Valley Demonstration Project chemical composition (kilograms). a,b

Compound	Mass	Compound	Mass
Aluminum oxide	38,000	Nickel monoxide	1,600
Boron oxide	82,000	Phosphorous pentaoxide	7,600
Barium oxide	1,000	Rubidium oxide	500
Calcium oxide	3,000	Silicon oxide	260,000
Ceric oxide	2,000	Strontium oxide	100
Chromic oxide	900	Thorium dioxide	23,000
Ferric oxide	76,000	Titanium dioxide	4,300
Potassium oxide	32,000	Uranium oxide	3,000
Lithium oxide	24,000	Zinc oxide	100
Magnesium oxide	5,600	Zirconium dioxide	7,100
Manganese oxide	5,200	Others	3,900
Sodium oxide	51,000		
Neodymium oxide	900	Total	630,000

a. Sources: DIRS 103500-WVNS (n.d., WQR-1.1, Rev. 1, p. 7); DIRS 104413-Picha (1998, p. 3).

**Table A-38.** Idaho National Engineering and Environmental Laboratory waste stream thermal output (watts). a,b

Output per waste canister	INTEC glass matrix	ANL-W ceramic matrix	ANL-W metal matrix
Average <sup>c</sup>	7.1	160	170
Worst case d	180	620	410

a. Source: DIRS 104406-Picha (1997, Attachment 1, p. 2).

b. To convert kilograms to pounds, multiply by 2.2046.

c. Includes trace amounts of silver, americium, cobalt, and antimony.

b. To convert kilograms to pounds, multiply by 2.2046.

b. INTEC = Idaho Nuclear Technology and Engineering Center; ANL-W = Argonne National Laboratory-West.

c. Based on average case; 2035 used as base year for Idaho Nuclear Technology and Engineering Center glass and 2000 for ANL-W matrices.

Based on worst case; 2020 used as base year for Idaho Nuclear Technology and Engineering Center glass and 2000 for ANL-W matrices.

*Savannah River Site*. The radionuclide inventories reported for the Savannah River Site high-level radioactive waste in Section A.2.3.5.2 were used to calculate projected heat generation rates for single canisters.

For the design-basis waste form, the heat generation rates 10 and 20 years after production are 465 and 302 watts per canister, respectively (DIRS 101909-Plodinec, Moore, and Marra 1993, pp. 8 and 9).

West Valley Demonstration Project. West Valley has calculated heat generation rates for a nominal West Valley canister after several different decay times (DIRS 103500-WVNS n.d., WQR-3.8, Rev. 5, 6/29/00, p. 3). In the nominal case, the ORIGEN2-computed heat generation rate was 239 watts at the calculational base time in 1988. The heat generation rate would decrease continuously from 239 watts to about 155 watts after 19 years of additional decay.

# A.2.3.5.5 Quantity of Waste Per Canister

Table A-39 lists the estimated mass of glass per waste canister for each high-level radioactive waste stream.

Table A-3	<b>9.</b> Approximate m	ass of high-level radioad	ctive waste glass j	per canister (kilograms). <sup>a</sup>

Waste stream <sup>b</sup>	Mass per canister	Source
Hanford	3,040	DIRS 104406-Picha (1997, Attachment 1, p. 2)
INEEL		
INTEC	1,560	DIRS 104406-Picha (1997, Attachment 1, p. 2)
ANL-W ceramic <sup>c</sup>	960 - 1,500	DIRS 104389-Goff (1998, Attachment, p. 5)
ANL-W metal <sup>c</sup>	1,500 - 4,850	DIRS 104389-Goff (1998, Attachment, p. 5)
Savannah River Site	2,000	DIRS 104403-Pearson (1998, all)
WVDP	2,000	DIRS 104406-Picha (1997, Attachment 1, p. 2)

a. To convert kilograms to pounds, multiply by 2.2046.

### A.2.3.5.6 High-Level Radioactive Waste Canister Parameters

*Hanford Site.* Table A-40 lists preliminary physical parameters for a standard canister used for high-level radioactive wastes from the Hanford Site (DIRS 104406-Picha 1997, Table RL-3).

Idaho National Engineering and Environmental Laboratory. The Idaho Nuclear Technology and Engineering Center would use stainless-steel canisters identical in design to those used at the Savannah River Site in the Defense Waste Processing Facility. A similar canister would also be used to contain the ceramic and metal waste matrices resulting from the high-level radioactive waste processing at Argonne National Laboratory-West (DIRS 104406-Picha 1997, Table ID-1).

Savannah River Site. The fabrication specifications of the Defense Waste Processing Facility high-level radioactive waste canisters are described in detail in DIRS 101854-Marra, Harbour, and Plodinec (1995, all). The 3-meter (10-foot) long canisters are fabricated from four basic pieces of A240 304L austenitic stainless steel—the main cylinder, the bottom head, the top head, and a nozzle.

b. INEEL = Idaho National Engineering and Environmental Laboratory; INTEC = Idaho Nuclear technology and Engineering Center; ANL-W = Argonne National Laboratory-West; WVDP = West Valley Demonstration Project.

c. These values are estimates. ANL-W is evaluating waste package configurations compatible with existing storage and remote hot cell facilities. The geometries would be compatible with the Defense Waste Processing Facility high-level radioactive waste canister.

**Table A-40.** Parameters of the proposed standard canister for Hanford high-level radioactive waste disposal.<sup>a</sup>

Parameter	Value <sup>b</sup>	Comments <sup>c</sup>
Length	4.50 meters	1.5 meters longer than DWPF and WVDP canisters - nominal 4.5-meter length
Nominal outer diameter	0.61 meter	Same as DWPF and WVDP canisters
Material	304L stainless steel	Same as DWPF and WVDP canisters
Canister weight	720 kilograms	
Dished bottom	Yes	Same as DWPF and WVDP
Available volume	1.2 cubic meters	
Nominal percent fill	90 percent	Provides approximately same void volume as WVDP canister
Glass volume	1.1 cubic meters	

a. Source: DIRS 104406-Picha (1997, Table RL-3).

West Valley Demonstration Project. The West Valley canister is designed, fabricated, and handled in accordance with the specifications in the West Valley Demonstration Project Waste Qualification Report (DIRS 103500-WVNS n.d., WQR-2.2). The West Valley canisters are also 3 meters (10 feet) long and fabricated from four principal 304L austenitic stainless-steel components.

#### A.2.3.5.7 Nonstandard Packages

Each site that would ship high-level radioactive waste to the repository has provided additional data on an estimate of nonstandard packages for possible inclusion in the candidate waste material. The mass, volume, and radioactivity of potential nonstandard packages would be dominated by failed or spent melters from the vitrification facilities. Final disposition plans for these melters are in development and vary from site to site. The EIS used the following assumptions to estimate the potential inventory.

Hanford Site. DOE could need to ship such nonstandard high-level radioactive waste packages as failed melters and failed contaminated high-level radioactive waste processing equipment to the repository. For this EIS, the estimated volume of nonstandard packages available for shipment to the repository from the Hanford Site would be equivalent to that described below for the Savannah River Site.

Idaho National Engineering and Environmental Laboratory. DOE proposes to treat and dispose of nonstandard packages under existing regulations. However, to bound the number of failed melters the Idaho National Engineering and Environmental Laboratory could ship to the repository, this EIS uses the same ratio of failed melters to the number of canisters produced as the Savannah River Site (DIRS 104401-Palmer 1997, p. 2). The Idaho National Engineering and Environmental Laboratory would produce approximately 20 percent of the number of canisters produced at the Savannah River Site, which assumes 10 failed melters. Therefore, the Idaho National Engineering and Environmental Laboratory assumes two failed melters. The volumes and other parameters would then be twice the values listed in Table A-41 for an individual melter.

*Savannah River Site.* Table A-41 lists the estimated parameters of nonstandard packages for repository shipment from the Savannah River Site.

West Valley Demonstration Project. The West Valley Demonstration Project anticipates that it would send only one melter to the repository at the end of the waste solidification campaign. It would be disposed of as a nonstandard package. Table A-42 lists the estimated parameters of nonstandard packages from the West Valley Demonstration Project.

b. To convert meters to feet, multiply by 3.2808; to convert centimeters to inches, multiply by 0.3937; to convert kilograms to tons, multiply by 0.0011023; to convert cubic meters to cubic feet, multiply by 35.314.

c. DWPF = Defense Waste Processing Facility; WVDP = West Valley Demonstration Project.

Table A-41. Parameters of nonstandard packages from Savannah River Site.<sup>a</sup>

Parameter	Value
Volume	10 melters based on current planning to 2021
Activity	4.5 equivalent DWPF <sup>b</sup> canisters for each melter
Mass	1,000 metric tons <sup>c</sup> for 10 melters (filled melter: 100 metric tons)
Chemical composition	Glass (see Section A.2.3.5.3)  Melter – Refractory brick Aluminum Stainless steel Inconel
Quantity per disposal package	1 melter per disposal package
Thermal generation	4.5 times the heat generation of a single canister for each melter

a. Source: DIRS 104402-Pearson (1997, Attachment 1, pp. 3 and 4).

Table A-42. Parameters of nonstandard packages from West Valley Demonstration Project.<sup>a</sup>

Parameter	Value <sup>b</sup>
Volume	1 melter (24 cubic meters)
Activity	1.1 equivalent West Valley canisters
Mass	52 metric tons
Chemical composition	Melter refractories (38 metric tons) Inconel (11 metric tons) Stainless steel (1.6 metric tons) Glass (see Table A-37)
Quantity per disposal package	1 melter per package
Thermal generator	1.1 times the heat generation of a single canister (see Section A.2.3.5.4)

a. Source: DIRS 104418-Rowland (1997, all).

### A.2.4 SURPLUS WEAPONS-USABLE PLUTONIUM

### A.2.4.1 Background

The President has declared an amount of weapons-usable plutonium to be surplus to national security needs (DIRS 118979-DOE 1999, p. 1-1). This material includes the following:

- Plutonium in various forms (metal, oxide, etc.)
- Nuclear weapons components
- Materials that DOE could process in the future to produce purified plutonium
- Plutonium residues that DOE previously saved for future recovery of purified plutonium

These materials are currently stored at various facilities throughout the United States. DOE would draw the specific surplus weapons-usable plutonium it ultimately disposed of from the larger inventory primarily stored at these sites.

DOE could process the surplus weapons-usable plutonium as two material streams. One stream would be an immobilized plutonium ceramic form that DOE would dispose of using a can-in-canister technique with high-level radioactive waste. The second stream would be mixed uranium and plutonium oxide fuel assemblies that would be used for power production in light-water reactors and disposed of as commercial spent nuclear fuel. The Surplus Plutonium Disposition Final Environmental Impact

b. DWPF = Defense Waste Processing Facility.

c. To convert metric tons to tons, multiply by 1.1023.

b. To convert cubic meters to cubic feet, multiply by 35.314; to convert metric tons to tons, multiply by 1.1023.

Statement (DIRS 118979-DOE 1999, p. 1-1) evaluates the quantity of plutonium processed in each stream. This EIS assumes that approximately one-third of the surplus weapons-usable plutonium would be immobilized and two-thirds would be made into mixed-oxide commercial nuclear fuel. The actual split could include the immobilization of up to the entire inventory of plutonium addressed in DIRS 118979-DOE (1999, p. 1-1).

#### A.2.4.2 Sources

DOE would produce the immobilized plutonium and/or mixed-oxide fuel at the Savannah River Site as determined in a Record of Decision for the Surplus Plutonium Disposition Final Environmental Impact Statement (65 FR 1608; January 11, 2000). The Department analyzed the potential environmental impacts of using mixed-oxide fuel in six commercial light-water reactors in which it proposes to irradiate the mixed-oxide fuel: both units at Catawba in South Carolina; both units at McGuire in North Carolina; and both units at North Anna Power Station in Virginia (65 FR 1608, January 11, 2000). Subsequently, the Department has decided to pursue irradiation of mixed-oxide fuel at only the Catawba and McGuire units.

## A.2.4.3 Present Storage and Generation Status

DOE suspended planning and work activities for the immobilized plutonium program in April 2001. For planning purposes, immobilized plutonium production could start in 2012. DOE has not determined an immobilized plutonium production completion date.

The immobilization of one-third of the plutonium would produce an estimated 101 additional canisters of high-level radioactive waste, which the production location would store until shipment to the repository. The immobilization of the full considered inventory of plutonium would produce an estimated 272 additional canisters of high-level radioactive waste. This EIS assumes that the production location would be the Savannah River Site and, therefore, used the physical dimensions of the Defense Waste Processing Facility canisters to calculate these values (DIRS 118979-DOE 1999, pp. 2-26 and 2-27).

Commercial light-water reactors would use mixed-oxide fuel assemblies for power production starting as early as 2007. This fuel would replace the low-enriched uranium fuel that normally would be in the reactors. After the fuel assemblies were discharged from the reactors as spent mixed-oxide fuel, the reactor sites would store them until shipment to the repository.

#### A.2.4.4 Final Waste Form

The final waste form would be immobilized plutonium or spent mixed-oxide fuel. Section A.2.4.5 discusses the characteristics of these materials. The spent mixed-oxide fuel discussed here has different characteristics than the mixed-oxide fuel included in the National Spent Fuel Program (DIRS 153072-Wheatley 2000, all) and described in Section A.2.2.

#### A.2.4.5 Material Characteristics

### A.2.4.5.1 Mixed-Oxide Fuel

**A.2.4.5.1.1 Mass and Volume.** The EIS on surplus weapons-usable plutonium disposition (DIRS 118979-DOE 1999, p. 1-9) evaluates the disposal of two-thirds of the plutonium as mixed-oxide fuel. The amount of plutonium and uranium measured in metric tons of heavy metal going to a repository would depend on the average percentage of plutonium in the fuel. The percentage of plutonium would be influenced by the fuel design. DOE has chosen pressurized-water reactors for the proposed irradiation of these assemblies. For pressurized-water reactors, the expected average plutonium percentages would be

approximately 4.6 percent; however, they could range between 3.5 and 6 percent (DIRS 104422-Stevenson 1997, pp. 5 and 6). Table A-43 lists estimates and ranges for the total metric tons of heavy metal (uranium and plutonium) that would result from disposing of two-thirds of the plutonium in mixed-oxide fuel. The table also lists a corresponding estimate for the number of assemblies required, based on using the typical assemblies described in Section A.2.1.4. The ranges of metric tons of heavy metal account for the proposed range in potential plutonium percentage.

**Table A-43.** Estimated spent nuclear fuel quantities for disposition of two-thirds of the surplus weapons-usable plutonium in mixed-oxide fuel.<sup>a,b</sup>

	Plutonium	Best estimate	Assemblies	Range
Reactor and fuel type	percentage	(MTHM)	required	(MTHM)
Pressurized-water reactor	4.56	700	1,500	500-900

a. Source: DIRS 104422-Stevenson (1997, pp. 5 and 6).

DOE assumed that each spent mixed-oxide assembly irradiated and disposed of would replace an energy-equivalent, low-enriched uranium assembly originally intended for the repository. The mixed-oxide assemblies would be part of the 63,000 metric tons (69,000 tons) that comprise the commercial spent nuclear fuel disposal amount in the Proposed Action (DIRS 104405-Person 1998, all). DOE also assumes that the average burnup levels for the pressurized-water reactor would be the same as that for the energy-equivalent, low-enriched uranium fuel. Table A-44 lists the assumed burnup levels and the amount of heavy metal in an assembly.

Table A-44. Assumed design parameters for typical mixed-oxide assembly.<sup>a</sup>

Parameter	Pressurized-water reactor
Mixed-oxide and low-enriched uranium burnup (MWd/MTHM) <sup>b</sup>	45,000
Mixed-oxide assembly mass (kilograms <sup>c</sup> of heavy metal)	450
Mixed-oxide assembly percentage of plutonium	4.56

a. Source: DIRS 104422-Stevenson (1997, p. 7).

The analysis assumed that the mixed-oxide spent nuclear fuel would replace the low-enriched uranium fuel. Because of the similarities in the two fuel types, impacts to the repository would be small. Nuclear criticality, radionuclide release rates, and heat generation comparisons are evaluated in DIRS 104422-Stevenson (1997, pp. 35 to 37).

**A.2.4.5.1.2** Amount and Nature of Radioactivity. Tables A-45 and A-46 list isotopic composition data for spent mixed-oxide fuel assemblies. The tables reflect SCALE data files from an Oak Ridge National Laboratory report used with computer simulation to project the characteristics of spent mixed-oxide fuel in pressurized-water reactors (DIRS 100976-Murphy 1997, Volume 3, Appendix B). The tables summarize data for two different potential fuel assemblies: a typical pressurized-water reactor, and a high-burnup pressurized-water reactor. A high burnup pressurized-water assembly would be irradiated for three cycles in comparison to the two cycles for the typical assemblies. For each of these assemblies, the tables provide radioactivity data for the common set of nuclides used in this EIS for the assumed 5-year minimum cooling time.

**A.2.4.5.1.3 Chemical Composition.** Tables A-47 and A-48 list the elemental distributions for the typical and high-burnup pressurized-water reactor spent mixed-oxide fuel assemblies.

**A.2.4.5.1.4 Thermal Output.** Table A-49 lists the decay heat from the representative mixed-oxide spent fuel assemblies at a range of times after discharge.

b. MTHM = metric tons of heavy metal; to convert metric tons to tons, multiply by 1.1023.

b. MWd/MTHM = megawatt days per metric ton of heavy metal; to convert metric tons to tons, multiply by 1.1023.

c. To convert kilograms to pounds, multiply by 2.2046.

Table A-45. Radionuclide activity for typical pressurized-water reactor spent mixed-oxide assembly.<sup>a</sup>

Radionuclide <sup>b</sup>	Curies per assembly	Radionuclide <sup>b</sup>	Curies per assembly
Hydrogen-3	$2.0 \times 10^{2}$	Samarium-151	$5.3 \times 10^2$
Carbon-14	$3.4 \times 10^{-1}$	Uranium-234	$4.9 \times 10^{-2}$
Cobalt-60	$1.7 \times 10^3$	Uranium-235	$1.0 \times 10^{-3}$
Nickel-59	1.1	Uranium-236	$6.4 \times 10^{-3}$
Nickel-63	$1.4 \times 10^2$	Uranium-238	$1.4 \times 10^{-1}$
Krypton-85	$1.9 \times 10^{3}$	Plutonium-238	$1.2 \times 10^{3}$
Strontium-90	$1.7 \times 10^4$	Plutonium-239	$6.6 \times 10^{2}$
Zirconium-93	$6.5 \times 10^{-2}$	Plutonium-240	$8.6 \times 10^{2}$
Niobium-93m	$2.8 \times 10^{1}$	Plutonium-241	$2.0 \times 10^{5}$
Niobium-94	$6.8 \times 10^{-1}$	Americium-241	$2.2 \times 10^{3}$
Technetium-99	6.3	Americium-242/242m	$3.4 \times 10^{1}$
Ruthenium-106	$1.6 \times 10^4$	Americium-243	$2.4 \times 10^{1}$
Iodine-129	$2.1 \times 10^{-2}$	Curium-242	$6.0 \times 10^{1}$
Cesium-134	$1.4 \times 10^4$	Curium-243	$3.2 \times 10^{1}$
Cesium-137	4.7×10 <sup>4</sup>	Curium-244	$2.6 \times 10^3$

a. Source: DIRS 100976-Murphy (1997, Appendix B).

**Table A-46.** Radionuclide activity for high-burnup pressurized-water reactor spent mixed-oxide assembly.<sup>a,b</sup>

Radionuclide <sup>b</sup>	Curies per assembly	Radionuclide <sup>b</sup>	Curies per assembly
Hydrogen-3	$2.9 \times 10^{2}$	Uranium-234	6.8×10 <sup>-2</sup>
Carbon-14	$5.4 \times 10^{-1}$	Uranium-235	$6.7 \times 10^{-4}$
Cobalt-60	$2.4 \times 10^{3}$	Uranium-236	$7.7 \times 10^{-3}$
Nickel-59	1.7	Uranium-238	1.5×10 <sup>-1</sup>
Nickel-63	$2.3 \times 10^{2}$	Plutonium-238	$2.7 \times 10^3$
Krypton-85	$2.6 \times 10^{3}$	Plutonium-239	$4.6 \times 10^{2}$
Strontium-90	$2.4 \times 10^4$	Plutonium-240	$8.8 \times 10^{2}$
Niobium-93m	$3.9 \times 10^{1}$	Plutonium-241	$2.2 \times 10^{5}$
Niobium-94	$9.8 \times 10^{-1}$	Americium-241	$2.5 \times 10^{3}$
Technetium-99	9.0	Americium-242/242m	$4.9 \times 10^{1}$
Ruthenium-106	$1.8 \times 10^4$	Americium-243	$5.6 \times 10^{1}$
Iodine-129	$3.0 \times 10^{-2}$	Curium-242	$1.0 \times 10^{2}$
Cesium-134	$2.5 \times 10^4$	Curium-243	$8.5 \times 10^{1}$
Cesium-137	$7.0 \times 10^4$	Curium-244	$8.9 \times 10^{3}$
Samarium-151	$5.4 \times 10^2$		

a. Sources: DIRS 100976-Murphy (1997, Volume 3, Appendix B).

**A.2.4.5.1.5 Physical Parameters.** Because the mixed-oxide fuel would replace low-enriched uranium fuel in existing reactors, Section A.2.1.5.5 describes the physical parameters, with the exception of uranium and plutonium content, which are listed in Table A-44.

## A.2.4.5.2 Immobilized Plutonium

DOE has not yet determined the total quantity of plutonium for immobilization. The Department assumes that two-thirds of the considered inventory is "clean" metal suitable for use in mixed-oxide fuel, and that it could dispose of this material by burning it in reactors (DIRS 118979-DOE 1999, p. 1-1). The remaining surplus plutonium would require considerable additional chemical processing to make it suitable for reactor use. This EIS evaluates two cases, one in which DOE immobilizes only the "impure" materials (base case) and a second in which it immobilizes the entire considered surplus inventory. The base case is evaluated for the Proposed Action because it is DOE's preferred alternative (DIRS 118979-

b. Half-lives are listed in Table A-11.

b. Half-lives are listed in Table A-11.

**Table A-47.** Elemental distribution of typical burn-up pressurized-water reactor spent mixed-oxide assembly.<sup>a</sup>

Element	Grams per assembly <sup>b</sup>	Percent <sup>c</sup>	Element	Grams per assembly	Percent
Americium	770	0.12	Palladium	1,200	0.19
Barium	750	0.12	Phosphorus	140	0.02
Carbon	67	0.12	Plutonium	17,000	2.59
Cerium	1,100	0.16	Praseodymium	500	0.08
Cesium	1,500	0.23	Rhodium	360	0.05
Chromium	2,300	0.36	Rubidium	91	0.01
Europium	90	0.01	Ruthenium	1,300	0.20
Iodine	150	0.02	Samarium	440	0.07
Iron	4,600	0.71	Silicon	66	0.01
Krypton	100	0.02	Strontium	210	0.03
Lanthanum	540	0.08	Technetium	370	0.06
Manganese	110	0.02	Tellurium	260	0.04
Molybdenum	1.700	0.27	Tin	1900	0.28
Neodymium	1,700	0.26	Uranium	428,000	65.92
Neptunium	72	0.01	Xenon	2500	0.38
Nickel	4,400	0.68	Yttrium	110	0.02
Niobium	330	0.05	Zirconium	111,000	17.10
Oxygen	62,000	9.56	Totals	648,000	99.73

a. Source: DIRS 104399-Murphy (1998, all).

**Table A-48.** Elemental distribution of high burn-up pressurized-water reactor spent mixed-oxide assembly.<sup>a</sup>

T-1	Grams per	D		Grams per	
Element	assembly <sup>b</sup>	Percent <sup>c</sup>	Element	assembly	Percent
Americium	1,000	0.16	Palladium	2,000	0.30
Barium	1,200	0.18	Phosphorus	140	0.02
Carbon	70	0.01	Plutonium	14,000	2.22
Cerium	1,600	0.24	Praseodymium	750	0.11
Cesium	2,100	0.33	Rhodium	460	0.07
Chromium	2,300	0.36	Rubidium	140	0.02
Europium	140	0.02	Ruthenium	2,000	0.31
Iodine	220	0.03	Samarium	630	0.10
Iron	4,600	0.71	Silicon	66	0.01
Krypton	150	0.02	Strontium	300	0.05
Lanthanum	810	0.12	Technetium	520	0.08
Manganese	100	0.02	Tellurium	390	0.06
Molybdenum	2,500	0.39	Tin	1,900	0.29
Neodymium	2,500	0.39	Uranium	421,000	64.84
Neptunium	93	0.01	Xenon	3,700	0.57
Nickel	4,400	0.68	Yttrium	170	0.03
Niobium	330	0.05	Zirconium	111,000	17.10
Oxygen	62,000	9.56	Totals	646,000	99.46

a. Source: DIRS 104399-Murphy (1998, all).

DOE 1999, p. 1-1). The EIS evaluates the second case for potential cumulative impacts (Modules 1 and 2) because it would conservatively predict the largest number of required high-level radioactive waste canisters.

b. To convert grams to ounces, multiply by 0.035274.

c. Table includes only elements that constitute at least 0.01 percent of the total; therefore, total is slightly less than 100 percent.

b. To convert grams to ounces, multiply by 0.035274.

c. Table includes only elements that constitute at least 0.01 percent of the total; therefore, total is slightly less than 100 percent.

**Table A-49.** Mixed-oxide spent nuclear fuel thermal profile (watts per assembly).<sup>a</sup>

Years	Typical PWR <sup>b</sup>	High-burnup PWR
1	6,100	8,000
5	1,000	1,600
10	670	1,100
15	610	970
30	540	780
100	370	430
300	240	260
1,000	110	110
3,000	42	38
10,000	25	22
30,000	10	7.9
100,000	1.5	1.3
250,000	0.5	0.6

a. Source: DIRS 100976-Murphy (1997, Volume 3, Appendix B).

**A.2.4.5.2.1 Mass and Volume.** In DOE's preferred disposition alternative, immobilized plutonium would arrive at the repository in canisters of vitrified high-level radioactive waste that would be externally identical to standard canisters from the Defense Waste Processing Facility at the Savannah River Site. Smaller cans containing immobilized plutonium in ceramic disks would be embedded in each canister of high-level radioactive waste glass. This is the *can-in-canister* concept. Because the design of the can-in-canister is not final, DOE has not determined final waste loadings per canister, volume displaced by the cans, or other specifications. DOE estimates that each canister would contain 28 cans, but has not yet finalized the actual number. One of the limitations on the number of cans is determined by the ability to ensure that the high-level radioactive waste glass would fill completely around the cans; increasing the volume that the cans would occupy in a canister could increase the difficulty of achieving this.

Marra, Harbour, and Plodinec (DIRS 101854-1995, p. 2) describes the volume of a high-level radioactive waste canister. Each canister has a design capacity of 2,000 kilograms (4,400 pounds) of high-level radioactive waste glass. A nominal glass density of 2.7 grams per cubic centimeter (0.10 pound per cubic inch) yields a design glass volume of 620 liters (22 cubic feet). The 28 cans containing plutonium would displace 68 liters (2.4 cubic feet), or about 11 percent of the available volume. The rack holding the cans would displace about an additional 1 percent of the available volume, yielding a total displacement of about 12 percent.

Table A-50 lists the number of high-level radioactive waste canisters required to dispose of immobilized surplus plutonium using the loading and volumetric assumptions given above for both the base and full inventory cases. It also lists the number of additional canisters DOE would have to produce (in addition to those the high-level radioactive waste producer would already have produced) due to the displacement of high-level radioactive waste glass by the plutonium-containing canisters. The total number of required canisters would be a function of both the number of cans in each canister and the plutonium loading of the immobilization form. The number of additional canisters would depend only on the plutonium loading of the immobilization form.

**A.2.4.5.2.2 Amount and Nature of Radioactivity.** Assuming the current 10.5-percent plutonium loading in the ceramic, the expected isotopic composition of the various materials in the feedstream for ceramic production, and the nominal quantity of ceramic in each canister, Stevenson (DIRS 104422-1997, p. 49) calculated the activity of the immobilized material in each high-level radioactive waste canister.

b. PWR = pressurized-water reactor.

**Table A-50.** Number of canisters required for immobilized plutonium disposition. <sup>a,b</sup>

Canisters	Base case	Full inventory case
Containing plutonium	670	1,820
In excess of those required for DWPF <sup>c</sup> (12% of total canisters)	101	272
Additional <sup>d</sup>	1.7%	4.5%

- a. Source: DIRS 118979-DOE (1999, p. 2-29).
- b. Assumes displacement of 12 percent of the high-level radioactive waste glass by plutonium cans and rack.
- c. DWPF = Defense Waste Processing Facility.
- d. As percentage of total planned DWPF canisters (about 6,000).

The figures do not include the radioactivity of the vitrified high-level radioactive waste that would surround the cans of immobilized plutonium. Calculation of the total radioactivity of a canister requires the subtraction of approximately 12 percent from the radioactivity of a full high-level radioactive waste canister to account for the displacement of the immobilized plutonium and its rack. Those reduced numbers, added to the appropriate figures in Table A-51, produce the total activity of a plutonium-containing high-level radioactive waste canister.

**Table A-51.** Average total radioactivity of immobilized plutonium ceramic in a single canister in 2010 (curies).<sup>a,b</sup>

Radionuclide <sup>c</sup>	Base case	Full inventory
Plutonium-238	120	60
Plutonium-239	1,600	1,700
Plutonium-240	550	430
Plutonium-241	4,700	2,800
Plutonium-242	0.098	0.046
Americium-241	720	430
Uranium-234	$< 0.000015^{d}$	< 0.000005
Uranium-235	0.0024	< 0.0011
Uranium-238	0.019	0.019
Thorium-232	< 0.00003	< 0.00003
Totals	7,700	5,400

- a. Source: DIRS 104422-Stevenson (1997, p. 49).
- b. Assumes 10.5 percent of plutonium by weight in ceramic form and 1:2 molar ratio of plutonium to uranium. These values account only for the radioactivity in the immobilized form; they do not include that in the surrounding high-level radioactive waste glass.
- c. Half-lives are listed in Table A-11.
- d. < = less than.

Values for the base case and the full inventory case are different because the plutonium in the base case contains more transuranic radionuclides, other than plutonium-239, than does the remainder of the plutonium. Thus, the "other" transuranic radionuclides are diluted in the full inventory case. From a thermal output and radiological impact standpoint, the base case is a more severe condition and, therefore, DOE has used it for the Proposed Action analysis.

Section A.2.3.5.2 contains information on the radioactivity contained in a standard Defense Waste Processing Facility high-level radioactive waste canister.

**A.2.4.5.2.3 Chemical Composition.** The current design for a ceramic immobilization form is a multiphase titanate ceramic, with a target bulk composition listed in Table A-52. The neutron absorbers, hafnium and gadolinium, are each present at a 1-to-1 atomic ratio to plutonium, and the atomic ratio of

uranium to plutonium is approximately 2-to-1. For the base case, the presence of impurities in some categories of surplus weapons-usable plutonium would result in the presence of a few weight percent of other nonradioactive oxides in some of the actual ceramic; Table A-52 does not list these impurities (DIRS 104422-Stevenson 1997, p. 51).

**Table A-52.** Chemical composition of baseline ceramic immobilization form.<sup>a</sup>

Oxide	Approximate percent by weight
Titanium oxide	36
Hafnium oxide	10
Calcium oxide	10
Gadolinium oxide	8
Plutonium oxide	12
Uranium oxide	24

a. Source: DIRS 104422-Stevenson (1997, p. 51).

The ceramic phase assemblage is mostly Hf-pyrochlore [(CaGd)(Gd,Pu,U,Hf)Ti<sub>2</sub>O<sub>7</sub>], with subsidiary Hf-zirconolite [(CaGd)(Gd,Pu,U,Hf)Ti<sub>2</sub>O<sub>7</sub>)], and minor amounts of brannerite [(U,Pu,Gd)Ti<sub>2</sub>O<sub>6</sub>] and rutile [(Ti,Hf)O<sub>2</sub>]. Pyrochlore and zirconolite differ in their crystalline structures. The presence of silicon as an impurity in the plutonium could lead to the formation of a minor amount of a silicate glass phase in the ceramic. This phase could contain a trace amount of the immobilized plutonium. Some residual plutonium oxide (less than 0.5 percent of the total quantity of plutonium) could also be present. The residual plutonium oxide contains uranium with smaller amounts of gadolinium and hafnium as a result of partial reaction with the other constituents of the ceramic (DIRS 104422-Stevenson 1997, p. 51). Section A.2.3.5.3 describes the chemical composition of the high-level radioactive waste glass surrounding the plutonium-containing cans.

**A.2.4.5.2.4 Thermal Output.** DIRS 104422-Stevenson (1997, p. 49) has presented the heat generation of the immobilized ceramic. These figures represent only the heat from the ceramic; they do not account for the heat from the surrounding high-level radioactive waste glass. The total heat from a Defense Waste Processing Facility canister containing high-level radioactive waste and immobilized plutonium would be the value listed in Table A-53 combined with 88 percent of the value listed in Section A.2.3.5.4 for the heat from a Defense Waste Processing Facility canister.

**Table A-53.** Thermal generation from immobilized plutonium ceramic in a single canister in 2010 (watts per canister).<sup>a</sup>

Case	Thermal production
Base case	8.6
Full inventory case	7.0

a. Source: DIRS 104422-Stevenson (1997, p. 49).

#### A.2.5 COMMERCIAL GREATER-THAN-CLASS-C LOW-LEVEL WASTE

# A.2.5.1 Background

Title 10 of the Code of Federal Regulations, Part 61 (10 CFR Part 61), establishes disposal requirements for three classes of waste—A, B, and C—suitable for near-surface disposal. Class C has the highest level of radioactivity and therefore the most rigorous disposal specifications. Wastes with concentrations

b. To convert metric tons to tons, multiply by 1.1023.

above Class C limits (listed in 10 CFR 61.55 Tables 1 and 2 for long and short half-life radionuclides, respectively) are called Greater-Than-Class-C low-level waste, and are not generally suitable for near-surface disposal (DIRS 101798-DOE 1994, all).

Commercial nuclear powerplants, research reactors, radioisotope manufacturers, and other manufacturing and research institutions generate waste that exceeds the Nuclear Regulatory Commission Class C shallow-land-burial disposal limits. Public Law 99-240 assigns the Federal Government, specifically DOE, the responsibility for disposing of this Greater-Than-Class-C waste. DOE could use a number of techniques for the disposal of these wastes, including engineered near-surface disposal, deep borehole disposal, intermediate-depth burial, and disposal in a deep geologic repository (DIRS 101798-DOE 1994, all).

The activities of nuclear electric utilities and other radioactive waste generators to date have produced relatively small quantities of Greater-Than-Class-C waste. As the utilities take their reactors out of service and decommission them, they could generate more waste of this type (DIRS 101798-DOE 1994, all).

Greater-Than-Class-C waste could include the following materials:

- Nuclear powerplant operating wastes
- Nuclear powerplant decommissioning wastes
- Sealed radioisotope sources that exceed Class C limits for waste classification
- DOE-held Greater-Than-Class-C waste (addressed in Section A.2.6)
- Greater-Than-Class-C waste from other generators

This section describes the quantities and characteristics of these waste types.

### A.2.5.2 Sources

Sources or categories of Greater-Than-Class-C waste include:

- DOE facilities (addressed in Section A.2.6)
- Nuclear utilities
- Sealed sources
- Other generators

Nuclear utility waste includes activated metals and process wastes from commercial nuclear powerplants. Sealed sources are radioactive materials in small metallic capsules used in measurement and calibration devices. Other generator wastes consist of sludge, activated metals, and other wastes from radionuclide manufacturers, commercial research, sealed-source manufacturers, and similar operations. The decommissioning of light-water reactors probably will generate additional Greater-Than-Class-C waste. Some internal reactor components will exceed Class C disposal limits.

### A.2.5.3 Present Status

Nuclear utilities store their Greater-Than-Class-C waste at the generator site, where it will remain until a disposal option becomes available.

Sealed sources are held by a Nuclear Regulatory Commission or Agreement State licensee. Current DOE sealed-source management plans call for the licensees to store their sealed-source wastes until a disposal option becomes available. If storage by a licensee became physically or financially impossible and a threat to public health and safety, the Nuclear Regulatory Commission would determine if the source was

a candidate for DOE storage. At that time, the Commission could request that DOE accept the source for storage, reuse, or recycling. The inventory projections do not include such a transfer of material.

In 1993, there were 13 identified "other generators" of Greater-Than-Class-C waste (DIRS 101798-DOE 1994, Appendix D), which were categorized into seven business types:

- Carbon-14 user
- Industrial research and development
- Irradiation laboratory
- Fuel fabricator
- University reactor
- Sealed-source manufacturer
- Nonmedical academic institution

These generators store their wastes at their sites and will continue to do so until a disposal site becomes operational.

### A.2.5.4 Final Waste Form

The final disposition method for Greater-Than-Class-C waste is not known. If DOE was to place such waste in a repository, it is assumed that it would be placed in a disposal canister before shipment. The EIS assumes the use of a canister similar to the naval canister, which is described in Section A.2.2.5.6, for all shipments by rail and a package similar to the high-level radioactive waste canisters for all shipments by truck.

#### A.2.5.5 Waste Characteristics

Table A-54 lists existing and projected volumes for the three Greater-Than-Class-C waste generator sources. DOE conservatively projects the volume of nuclear utility wastes to 2055 because that date would include the majority of this waste from the decontamination and decommissioning of commercial nuclear reactors. The projected volumes conservatively reflect the highest potential volume and activity based on inventories, surveys, and industry production rates. DOE projects the other two generator sources (sealed sources and other generators) to 2035 (DIRS 101798-DOE 1994, all).

**Table A-54.** Greater-Than-Class-C waste volume by generator source (cubic meters). a,b

	1993	Projected
Source	volume	volume
Nuclear electric utility	26	1,300
Sealed sources	39	240
Other generators	74	470
Totals	139	2,010

- a. Source: DIRS 101798-DOE (1994, all).
- b. To convert cubic meters to cubic feet, multiply by 35.314.

The data concerning the volumes and projections are from Greater-Than-Class-C Waste Characterization: Estimated Volumes, Radionuclide Activities, and Other Characteristics (DIRS 101798-DOE 1994, Appendix A-1), which provides detailed radioactivity reports for such waste currently stored at nuclear utilities. Table A-55 summarizes the radioactivity data for the primary radionuclides in the waste, projected to 2055.

**Table A-55.** Commercial Greater-Than-Class-C waste radioactivity (curies) by nuclide (projected to 2055).<sup>a</sup>

Nuclide <sup>b</sup>	Radioactivity
Carbon-14	$6.8 \times 10^4$
Cobalt-60	$3.3 \times 10^{7}$
Iron-55	$1.8 \times 10^{7}$
Hydrogen-3	$1.2 \times 10^4$
Manganese-54	$3.2 \times 10^4$
Niobium-94	$9.8 \times 10^{2}$
Nickel-59	$2.5 \times 10^{5}$
Nickel-63	$3.7 \times 10^{7}$
Transuranics	$2.0 \times 10^{3}$
Total	8.8×10 <sup>7</sup>

a. Source: DIRS 101798-DOE (1994, Appendix A-1).

Appendix B of DIRS 101798-DOE (1994) provides detailed radioactivity reports for the sealed sources, which could be candidate wastes for the repository. Table A-56 summarizes the radioactivity data for the radionuclides in these sources, projected to 2035.

**Table A-56.** Sealed-source Greater-Than-Class-C waste radioactivity (curies) by nuclide (projected to 2035).<sup>a</sup>

Nuclide <sup>b</sup>	Radioactivity
Americium-241	$8.0 \times 10^4$
Curium-244	$1.6 \times 10^2$
Cesium-137	$4.0 \times 10^{7}$
Plutonium-238	$1.6 \times 10^4$
Plutonium-239	$1.1 \times 10^5$
Plutonium-241	$2.8 \times 10^{1}$
Technetium-99	$5.8 \times 10^3$
Uranium-238	$5.7 \times 10^{1}$
Total	4.2×10 <sup>7</sup>

a. Source: DIRS 101798-DOE (1994, Appendix A-1).

DIRS 101798-DOE (1994, Section 5) also identifies the 13 other generators and the current and projected volumes and total radioactivity of Greater-Than-Class-C waste held by each. It does not provide specific radionuclide activity by nuclide. DOE used the data to derive a distribution, by user business type, of the specific nuclides that comprise the total radioactivity. Table A-57 lists this distributed radioactivity for other generators.

A detailed chemical composition by weight percentage for current Greater-Than-Class-C waste is not available. However, Table A-58 lists the typical composition of such wastes by generator.

The heat generation rates or thermal profiles for this waste type are not included in the source documentation. However, the contribution to the total thermal load at the repository from the Greater-Than-Class-C radioactive waste would be very small in comparison to commercial spent nuclear fuel or high-level radioactive waste.

b. Half-lives are listed in Table A-11.

b. Half-lives are listed in Table A-11.

**Table A-57.** Other generator Greater-Than-Class-C waste radioactivity (in curies) by nuclide (projected to 2035).<sup>a</sup>

Nuclide <sup>b</sup>	Radioactivity
Carbon-14	$7.7 \times 10^3$
Transuranic	$2.2 \times 10^{3}$
Cobalt-60	$1.5 \times 10^2$
Nickel-63	$1.5 \times 10^2$
Americium-241	$2.4 \times 10^{3}$
Cesium-137	$6.6 \times 10^{1}$
Technetium-99	$5.1 \times 10^{-2}$
Total <sup>c</sup>	$1.3 \times 10^4$

- a. Source: Derived from DIRS 101798-DOE (1994, Appendix D).
- b. Half-lives are listed in Table A-11.
- c. Total differs from sum of values due to rounding.

**Table A-58.** Typical chemical composition of Greater-Than-Class-C wastes.<sup>a</sup>

Source	Typical composition
Nuclear electric utility	Stainless steel-304, and zirconium
	alloys
Sealed sources	Stainless steel-304 (source material
	has very small mass contribution)
Other generators	Various materials

a. Source: DIRS 101798-DOE (1994, all).

#### A.2.6 SPECIAL-PERFORMANCE-ASSESSMENT-REQUIRED LOW-LEVEL WASTE

## A.2.6.1 Background

DOE production reactors, research reactors, reprocessing facilities, and research and development activities generate wastes that exceed the Nuclear Regulatory Commission Class C shallow-land-burial disposal limits. The Department is responsible for the safe disposal of such waste, and could use a number of techniques such as engineered near-surface disposal, deep borehole disposal, intermediate-depth burial, or disposal in a deep geologic repository. These wastes have been designated as Special-Performance-Assessment Required wastes.

DOE Special-Performance-Assessment-Required waste could include the following materials:

- Production reactor operating wastes
- Production and research reactor decommissioning wastes
- Non-fuel-bearing components of naval reactors
- Sealed radioisotope sources that exceed Class C limits for waste classification
- DOE isotope production-related wastes
- Research reactor fuel assembly hardware

#### A.2.6.2 Sources

DOE has identified Special-Performance-Assessment-Required waste inventories at several locations. Table A-59 lists the generators and amounts of these wastes. These amounts include current and projected inventories. These inventories are subject to revision as DOE develops its site material management programs and facility decommissioning plans.

**Table A-59.** Estimated Special-Performance-Assessment-Required low-level waste volume and mass by generator source.<sup>a</sup>

Source <sup>b</sup>	Volume (cubic meters) <sup>c</sup>	Mass (kilograms) <sup>d</sup>
Hanford	20	360,000
INEEL <sup>e</sup>	20	280,000
ORNL	2,900	4,700,000
WVDP	550	5,200,000
ANL-E	1	230
Naval Reactors Facility at INEEL	500	2,500,000
Totals (rounded)	3,990	13,000,000

- a. Source: DIRS 104411-Picha (1998, all).
- INEEL = Idaho National Engineering and Environmental Laboratory (including Argonne National Laboratory-West); ORNL = Oak Ridge National Laboratory; WVDP = West Valley Demonstration Project; ANL-E = Argonne National Laboratory-East.
- c. To convert cubic meters to cubic yards, multiply by 1.3079.
- d. To convert kilograms to pounds, multiply by 2.2046.
- e. Includes Argonne National Laboratory-West.

#### A.2.6.3 Present Status

DOE stores its Special-Performance-Assessment-Required waste at the generator sites listed in Table A-59. Tables A-60 through A-63 list the waste inventories at the individual sites. For radionuclides, these tables include only the reported isotopes with inventories greater than  $1 \times 10^{-5}$  curies. Table A-64 lists the chemical composition of this material at each site.

**Table A-60.** Hanford Special-Performance-Assessment-Required low-level waste radioactivity by nuclide (curies).<sup>a</sup>

Nuclide <sup>b</sup>	Radioactivity
Cesium-137	6.0×10 <sup>4</sup>
Strontium-90	$6.0 \times 10^4$

- a. Source: DIRS 104411-Picha (1998, all).
- b. Half-lives are listed in Table A-11.

**Table A-61.** Idaho National Engineering and Environmental Laboratory (including Argonne National Laboratory-West) Special-Performance-Assessment-Required low-level waste radioactivity by nuclide (curies).<sup>a</sup>

Nuclide <sup>b</sup>	Radioactivity
Hydrogen-3	$5.9 \times 10^6$
Carbon-14	$8.3 \times 10^{2}$
Cobalt-60	$1.1 \times 10^{6}$
Nickel-59	$9.0 \times 10^{1}$
Nickel-63	$1.3 \times 10^4$
Strontium-90	$7.4 \times 10^{3}$
Niobium-94	$1.4 \times 10^{2}$
Technetium-99	3.3
Cesium-137	$3.1 \times 10^{1}$
Radium-226	$3.0 \times 10^{1}$
Plutonium-239	$2.0 \times 10^{1}$
Americium-241	$2.4 \times 10^2$
a	(4000 11)

- a. Source: DIRS 104411-Picha (1998, all).
- b. Half-lives are listed in Table A-11.

**Table A-62.** Oak Ridge National Laboratory Special-Performance-Assessment-Required low-level waste radioactivity by nuclide (curies).<sup>a</sup>

Nuclide <sup>b</sup>	Radioactivity
Hydrogen-3	$1.9 \times 10^{6}$
Carbon-14	$1.0 \times 10^{1}$
Cobalt-60	$1.9 \times 10^{6}$
Nickel-59	$7.6 \times 10^3$
Nickel-63	$7.5 \times 10^5$
Strontium-90	$8.3 \times 10^7$
Niobium-94	$1.0 \times 10^4$
Technetium-99	$8.0 \times 10^{-1}$
Iodine-129	$7.5 \times 10^{-5}$
Cesium-137	1.7×10 <sup>-4</sup>

a. Source: DIRS 104411-Picha (1998, all).

**Table A-63.** Radioactivity of naval Special-Performance-Assessment-Required waste (curies per package).<sup>a</sup>

Radionuclide <sup>b</sup>	Short canister	Long canister	Radionuclide	Short canister	Long canister
Americium-241	$5.4 \times 10^{-2}$	$6.0 \times 10^{-2}$	Nickel-59	$2.2 \times 10^{2}$	$2.5 \times 10^{2}$
Americium-242m	$5.8 \times 10^{-4}$	$6.5 \times 10^{-4}$	Nickel-63	$2.7 \times 10^4$	$3.0 \times 10^4$
Americium-243	$5.8 \times 10^{-4}$	$6.5 \times 10^{-4}$	Plutonium-239	$2.1 \times 10^{-2}$	$2.4 \times 10^{-2}$
Carbon-14	3.2	3.6	Plutonium-240	$5.4 \times 10^{-3}$	$6.0 \times 10^{-3}$
Chlorine-36	$5.3 \times 10^{-2}$	$6.0 \times 10^{-2}$	Plutonium-241	4.1	4.6
Curium-242	$1.4 \times 10^{-3}$	$1.5 \times 10^{-3}$	Plutonium-242	$4.5 \times 10^{-5}$	$5.1 \times 10^{-5}$
Curium-243	$6.6 \times 10^{-4}$	$7.4 \times 10^{-4}$	Ruthenium-106	$2.1 \times 10^{-1}$	$2.3 \times 10^{-1}$
Curium-244	$7.0 \times 10^{-2}$	$7.9 \times 10^{-2}$	Selenium-79	$1.2 \times 10^{-5}$	$1.3 \times 10^{-5}$
Curium-245	$1.3 \times 10^{-5}$	$1.5 \times 10^{-5}$	Samarium-151	$1.7 \times 10^{-2}$	$1.9 \times 10^{-2}$
Cesium-134	1.6	1.8	Tin-126	$1.2 \times 10^{-5}$	$1.3 \times 10^{-5}$
Cesium-135	$1.1 \times 10^{-5}$	$1.2 \times 10^{-5}$	Strontium-90	$4.2 \times 10^{-1}$	$4.7 \times 10^{-1}$
Cesium-137	1.1	1.3	Technetium-99	$5.3 \times 10^{-4}$	$6.0 \times 10^{-4}$
Hydrogen-3	1.5	1.7	Uranium-232	$1.2 \times 10^{-4}$	$1.4 \times 10^{-4}$
Krypton-85	$4.9 \times 10^{-2}$	$5.6 \times 10^{-2}$	Uranium-233	$7.8 \times 10^{-5}$	$8.8 \times 10^{-5}$
Niobium-93m	$3.6 \times 10^{-1}$	$4.1 \times 10^{-1}$	Zirconium-93	$3.8 \times 10^{-1}$	$4.3 \times 10^{-1}$
Niobium-94	$5.9 \times 10^{-1}$	$6.7 \times 10^{-1}$			

a. Source: DIRS 124679-Beckett (1998, Attachment 1).

**Table A-64.** Typical chemical composition of Special-Performance-Assessment-Required low-level waste.<sup>a</sup>

Source <sup>b</sup>	Composition
Hanford	Vitrified fission products in glass waste form; hot cell waste
INEEL	Activated metal
ORNL	Activated metal; isotope production waste; hot cell waste
WVDP	Activated metal; vitrified transuranic waste
Naval Reactors Facility at INEEL	Activated metal (zirconium alloy, Inconel, stainless steel)
Other generators	Stainless-steel sealed sources

a. Source: DIRS 104411-Picha (1998, all).

b. Half-lives are listed in Table A-11.

b. Half-lives are listed in Table A-11.

b. INEEL = Idaho National Engineering and Environmental Laboratory; ORNL = Oak Ridge National Laboratory; WVDP = West Valley Demonstration Project.

#### A.2.6.4 Final Waste Form

The final disposal method for DOE Special-Performance-Assessment-Required waste is not known. If the Department disposed of such waste in a repository, it is assumed that the material would be placed in a disposable package before shipment to the repository. The EIS assumes the use of a disposable canister similar to those used for naval fuels for all rail shipments and packages similar to a high-level radioactive waste canister for all truck shipments.

#### A.2.6.5 Waste Characteristics

The low-level waste from West Valley consists of material in the Head End Cells (5 cubic meters [177 cubic feet]) and remote-handled and contact-handled transuranic waste (545 cubic meters [19,000 cubic feet]). The estimated radioactivity of the material in the Head End Cells is 6,750 curies, while the activity of the remote-handled and contact-handled transuranic waste is not available at present (DIRS 104411-Picha 1998, all). The naval Special-Performance-Assessment-Required waste consists primarily of zirconium alloys, Inconel, and stainless steel (DIRS 124679-Beckett 1998, all); Table A-63 lists the specific radioactivity of the projected material 5 years after discharge.

The specific activity associated with the radium sources at Argonne National Laboratory-East has not been determined. However, in comparison to the other Special-Performance-Assessment-Required waste included in this section, its impact would be small.

## REFERENCES

Note: In an effort to ensure consistency among Yucca Mountain Project documents, DOE has altered the format of the references and some of the citations in the text in this Final EIS from those in the Draft EIS. The following list contains notes where applicable for references cited differently in the Draft EIS.

156369	Arenaz 2001	Arenaz, M.R. 2001. "DOE SNF Data Used in the Final Environmental Impact Statement (EIS) (EM-NP-01-004) Reference: Letter from S.J. Brocoum to M.R. Arenaz Dated December 13, 2000." Memorandum from M.R. Arenaz (DOE) to S.J. Brocoum (DOE), January 24, 2001. ACC: MOL.20010808.0476.
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